

# ORGANOPHOSPHORUS-HYDRAZIDES AS POTENTIAL REACTIVE FLAME RETARDANTS FOR EPOXY

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# Outline

- Review of Current Solutions for Fire Protection of Epoxy Composites
  - Engineering Solutions (Insulation, Metal Shields)
  - Intumescent Barriers
  - Flame Retardant Additives
- Organophosphorus-Hydrazides
  - Synthesis Details
  - Heat Release Reduction Results
- Conclusions & Acknowledgements

# Current Flame Retardant / Fire Protection Solutions for Epoxy Composites

# Epoxy Composite Use in Transportation

- Epoxy + Carbon or Glass Fiber Composite Benefits:
  - Light Weight
  - Resistance to Corrosion / Rust
  - Unique shapes and forms due to manufacturing process.
- Drawbacks:
  - Non-electrical conductivity
  - Thermal properties
    - Failure of structural composite well before ignition
  - Flammability
    - Inherently higher heat release when compared to metal
    - Requires different fire fighting measures

# Epoxy Composite Use Examples



# Extinguishing Carbon Fiber Composite Fires

- Due to differences in fire behavior, fire fighting is different for composites vs. metal/fuel pool fires.
  - Fire fighting foam has little to no effect on composite fires, as it does not cool the burning composite – re-ignition common.
  - Water / CO<sub>2</sub> required to put out composite fires.
    - Foam still required to address the fuel pool, as well as tamp down flying ashes from the fire.
  - Cutting into structure to put out fire or release occupants requires diamond saws to cut through carbon fiber.
  - Guidance from Boeing on 787 fire fighting:
    - [http://www.boeing.com/assets/pdf/commercial/airports/faqs/787\\_composite\\_arff\\_data.pdf](http://www.boeing.com/assets/pdf/commercial/airports/faqs/787_composite_arff_data.pdf)

# Other Fire Hazards from Carbon Fiber Composites

- Recent studies have shown that carbon particulates are released when carbon fiber composites burn.
  - More than just soot – parts of the carbon fibers themselves are released.
  - “Dangers relating to fires in carbon-fibre based composite material” Hertzberg, T. *Fire and Materials* **2005**, 29, 231-248.
  - “Potential for the formation of respirable fibers in carbon fiber reinforced plastic materials after combustion” Eibl, S. *Fire Mater.* **2017**, 41, 808-816.
- All fires hazardous, but additional care / SCBA equipment and post-fire cleanup may be required with carbon fiber composite fires.

# Engineering Solutions

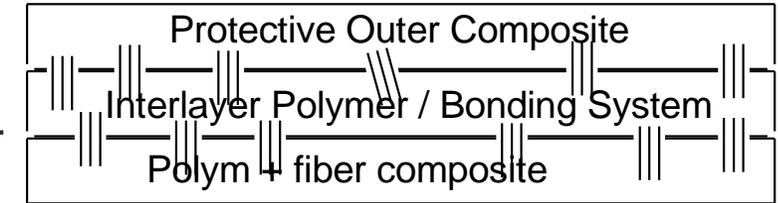
- Engineering approaches focus not on changing polymer chemistry, but on protection of flammable materials from ignition source/heat.
  - Fire protection barriers
    - Mineral Wool Batting / Insulation Packages
    - Metal decorative / structural panels



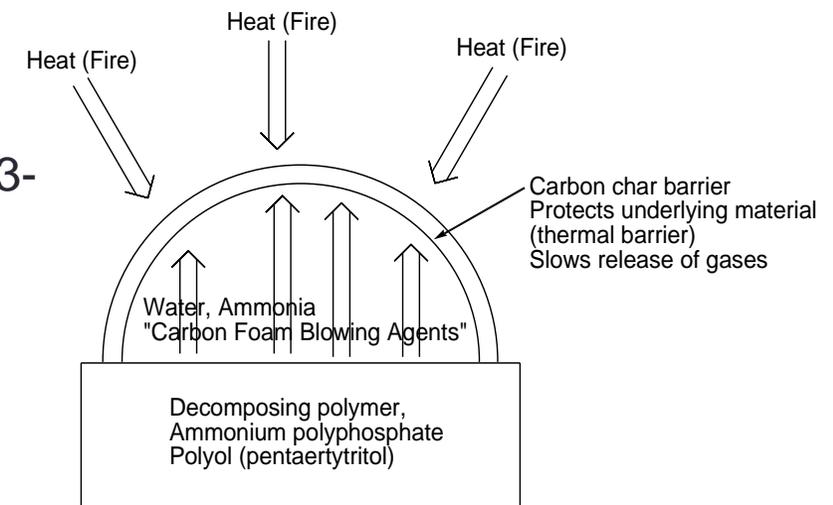
- Easy to implement, but, they are also easy to defeat.
  - Cut/break in barrier is weak point for fire damage/flame propagation to occur.
  - Barrier falling off/debonding can lead to rapid fire growth or mechanical failure

# Protective Integrated Coatings/Barriers for Composites

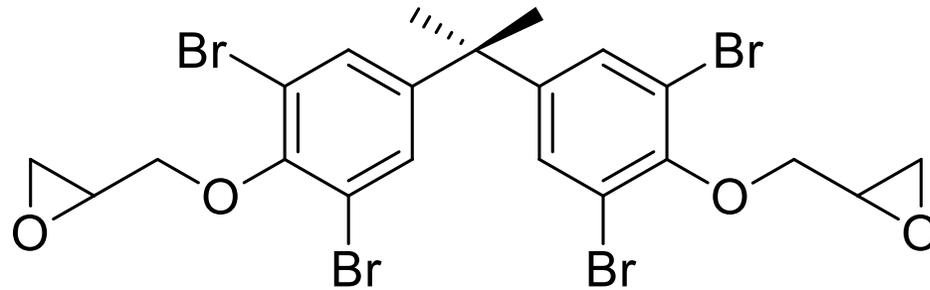
- Coating/Barrier can be misnomer.
  - The material is not for a paint, but an integrated outer layer to protect the composite part
- If successful, the “coating” will not crack/pop-off due to thermal stresses or damage since the coating is actually part of the entire component, not a bolt-on or glued on substrate.
- Some Co-cured Intumescent examples:
  - "Enhancement of Passive Fire Protection Ability of Inorganic Fire Retardants in Vinyl Ester Resin Using Glass Frit Synergists" Kandola, B. K.; Pornwannachai, W. *J. Fire Sci.* **2010**, *28*, 357-381.
  - “The effect of fire-retardant additives and a surface insulative fabric on fire performance and mechanical property retention of polyester composites” Kandare, E.; Chukwunonso, A. K.; Kandola, B. K. *Fire Mater.* **2011**, *35*, 143-155.
  - The use of fire-retardant intumescent mats for fire and heat protection of glass fibre-reinforced polyester composites: Thermal barrier properties" Kandare, E.; Chukwudole, C.; Kandola, B. K. *Fire and Materials* **2010**, *34*, 21-38.



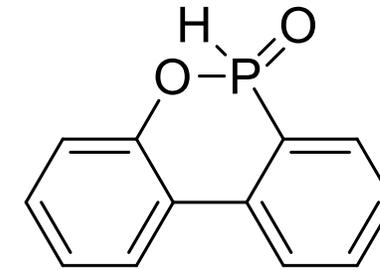
|| = polymer diffusion / chemical bonding between interlayers and outer composite



# Flame Retardant Additives



Tetrabromobisphenol A  
Bis Epoxide



9,10-dihydro-9-oxa-10-  
phosphaphenanthrene-  
10-oxide (DOPO)

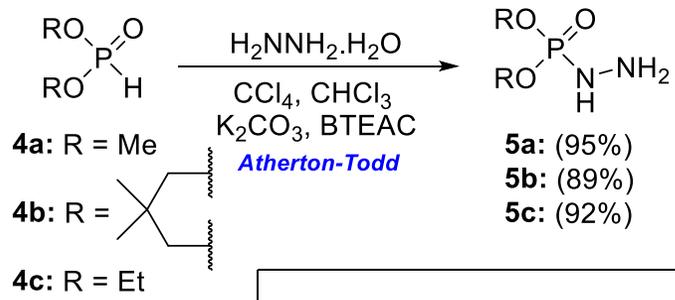
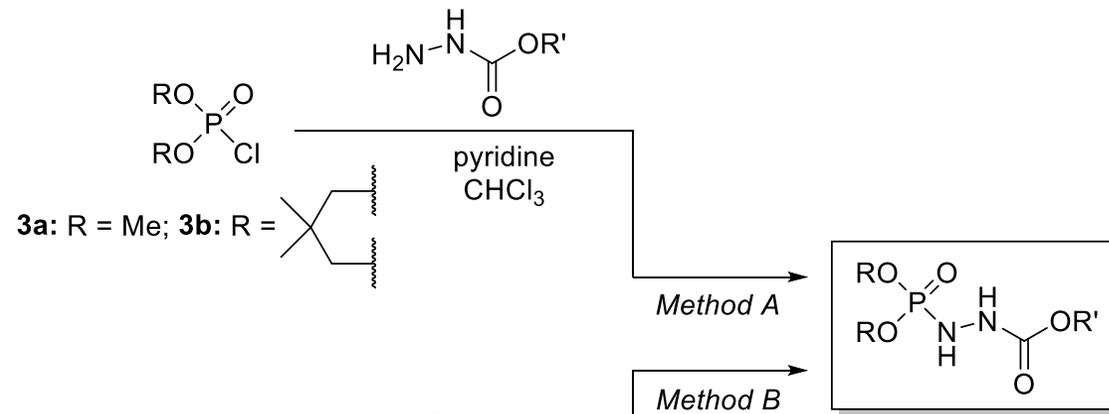
- Common flame retardant additives (FRs) for epoxy include both brominated and phosphorus based additives.
- Reactives are preferred from a durability / prevention of migration perspective.
  - Reactives are FRs which can react into / with the epoxy during composite manufacture, epoxy + curing agent mixing.

# Need for New Epoxy Reactive FRs

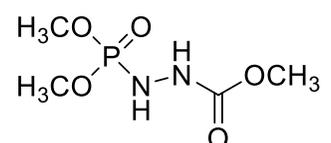
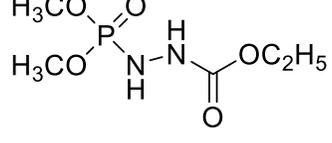
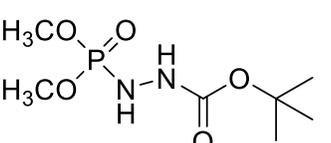
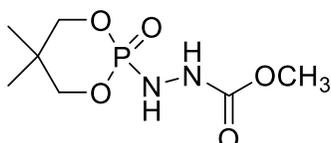
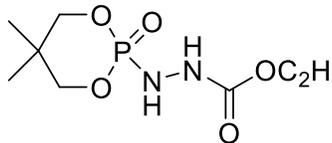
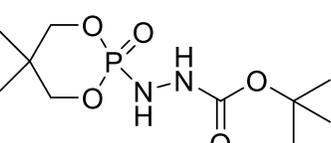
- TBBPA based epoxy proven and works, but is incompatible with non-Pb based solder.
  - TBBPA thermally decomposes when exposed to high melt temp solder.
- DOPO based epoxy proven and works for epoxy circuit boards, and likely will continue to gain in use.
  - Currently has no major issues, but may not be appropriate for large scale structural composites
- Very few reactive FRs available for epoxy materials.
  - Mixed vapor phase and condensed phase FR to allow for self extinguishment and reduction in heat release is desirable.

# Organophosphorus-Hydrazides

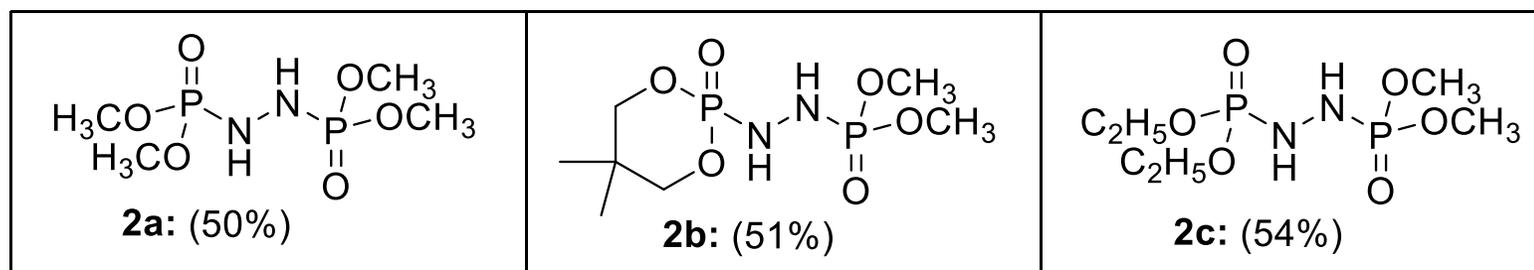
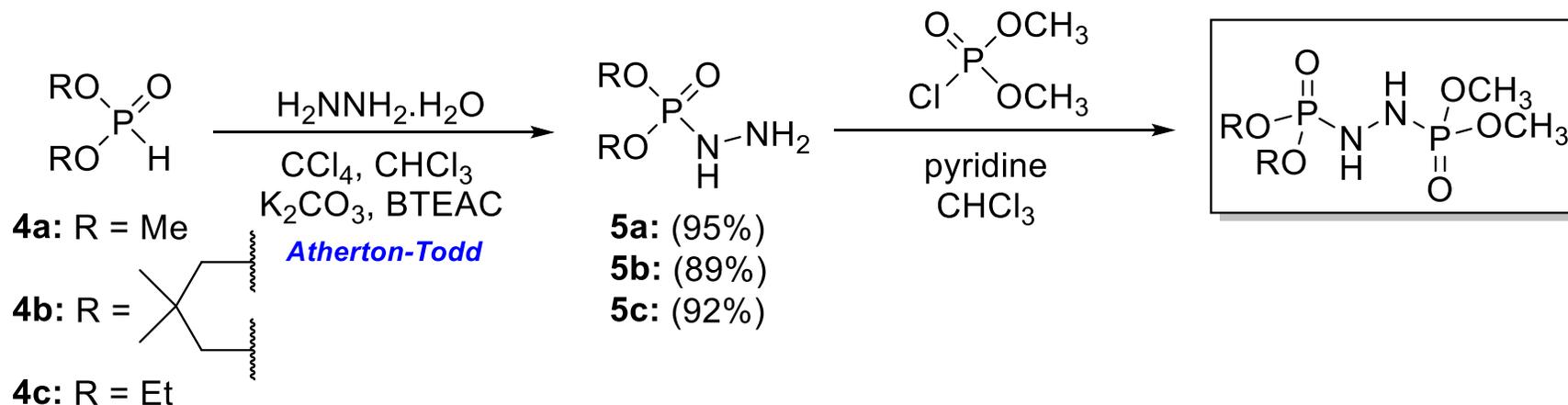
# Preparation of the Target Compounds: Hydrazine Monophosphonates



5a = P\_Me\_H  
 5b = P\_Cyc\_H  
 5c = P\_Et\_H  
 1a-I = 1P\_Me\_Me  
 1b-I = 1P\_Cyc\_Me

 <b>1a-I: Method B (73%)</b>	 <b>1a-II: Method A (76%), Method B (89%)</b>	 <b>1a-III: Method A (57%)</b>
 <b>1b-I: Method B (86%)</b>	 <b>1b-II: Method A (87%), Method B (95%)</b>	 <b>1b-III: Method A (81%)</b>

# Preparation of the Target Compounds: Hydrazine Diphosphonates



Not Yet Tested  
[2P\_me\_Me]

2P\_cyc\_Me

2P\_Et\_Me

# Heat Release Reduction Results

- Epoxy functionalized flame retardants added to Bisphenol F (Epon 862) epoxy
  - Epoxy cured with aliphatic amine
    - Epikure 3274: 50-70% polyoxypropylene diamine, balance 4-nonylphenol
  - Small 2.5 to 3 gram batches for MCC testing
  - Larger batches for cone calorimeter testing.
  - All formulations targeted to have 2.5wt% Phosphorus
- Materials tested by micro combustion calorimeter (ASTM D7309-13) and cone calorimeter (ASTM E1354-18).

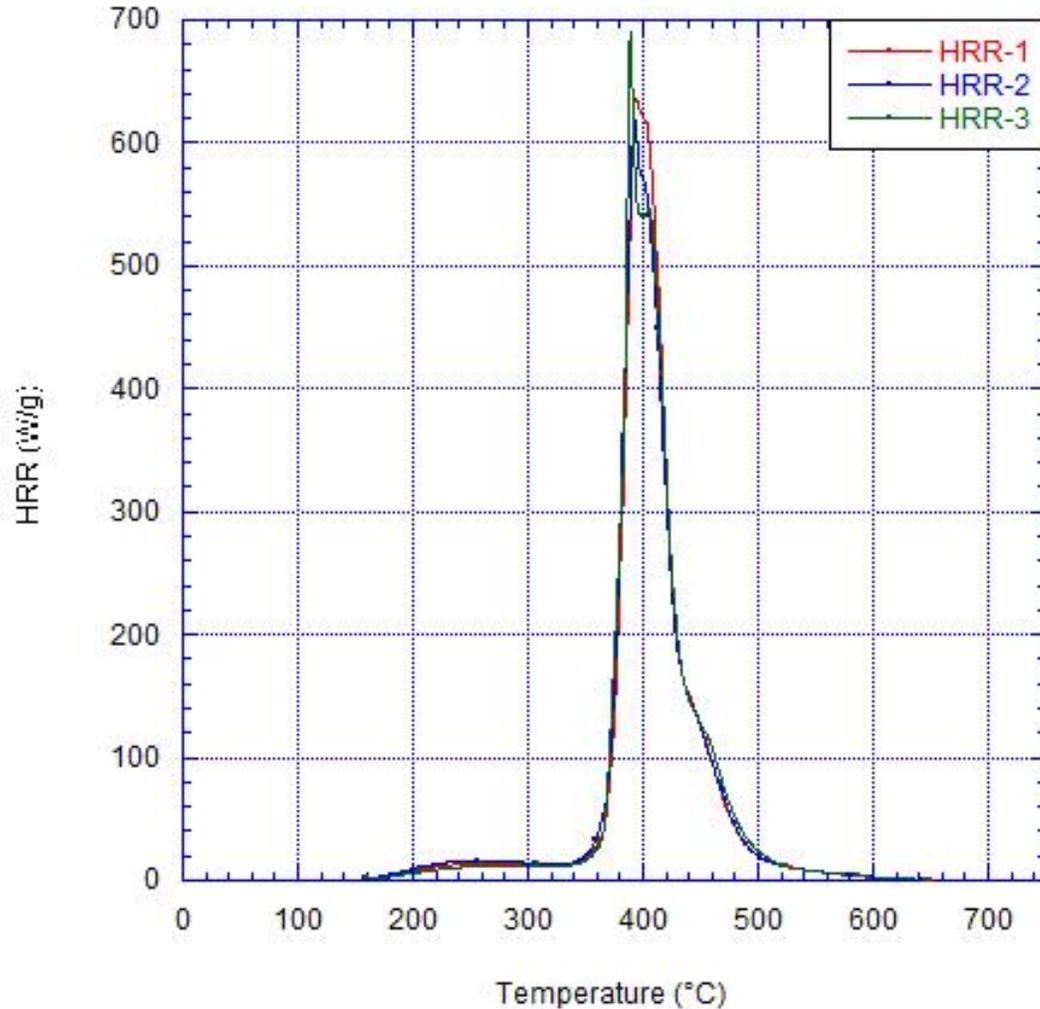
# Heat Release Data

- All samples show increased char yield, decreased total HR, decreased peak HRR values.
- Additional peaks of HRR found indicating a change in thermal decomposition chemistry for the epoxies.
- Additional studies needed to determine exact chemistry, but, mechanism of flame retardancy is likely condensed phase char formation

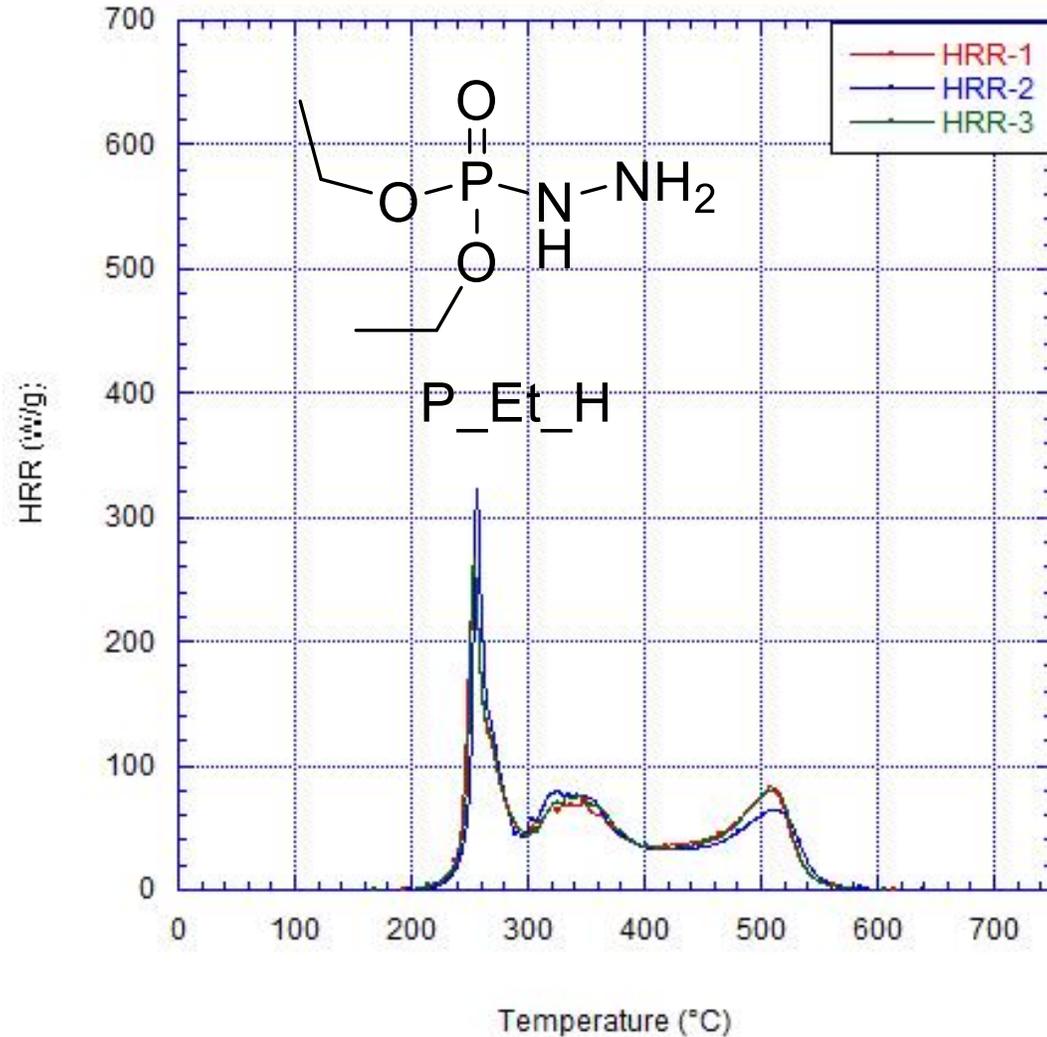
Sample	Char Yield (%)	HRR Peak(s) Value (W/g)	HRR Peak(s) Temp(s) (°C)	Total HR (kJ/g)
Epon 862/Epikure Control	9.21	633, 132	394, 512	23.7
RT + 140C/1 hr	9.36	621, 132	392, 516	23.5
	9.01	684, 124	389, 516	23.8
Epon 862/Epikure/P-Et-H	27.77	205, 207, 31	286, 337, 530	16.6
RT + 140C/1 hr	28.04	270, 193, 34	285, 339, 534	16.9
	27.85	290, 184, 33	284, 334, 528	16.7
Epon 862/Epikure/1P-Me-Me	18.52	292, 110, 22	331, 377, 514	20.3
RT + 140C/1 hr	19.43	294, 106, 19	334, 374, 508	20.3
	17.77	291, 110, 24	334, 390, 520	20.6
Epon 862/Epikure/2P-Et-Me	22.46	260, 196, 14	336, 354, 526	20.0
RT + 140C/1 hr	23.18	265, 185, 16	336, 356, 526	19.7
	22.57	280, 197, 15	337, 355, 539	19.9
Epon 862/Epikure/2P-cyc-Me	19.44	311, 190, 15	336, 353, 523	21.0
RT + 140C/1 hr	21.82	293, 192, 14	334, 354, 530	20.5
	20.71	298, 164, 16	336, 361, 526	20.6
Epon 862/Epikure/P-Me-H	27.13	25, 183, 19	254, 359, 530	17.6
RT + 140C/1 hr	27.59	27, 192, 19	254, 355, 540	17.4
	26.54	31, 171, 14	253, 359, 539	16.9
Epon 862/Epikure/1P-cyc-Me	25.59	38, 166, 22	287, 338, 529	17.5
RT + 140C/1 hr	24.92	36, 168, 22	282, 338, 499	17.9
	25.69	35, 164, 17	280, 348, 534	17.0
Epon 862/Epikure/P-cyc-H	15.21	35, 296	284, 351	22.1
RT + 140C/1 hr	15.55	37, 251	280, 357	21.9
	15.51	34, 275	277, 360	21.7
PS std 6 2018	0.03	1246	447	36.2
	0.05	1268	444	36.3
	0.00	1295	446	36.4

# Heat Release Curves – MCC Testing

**Epon 862 + Epikure**  
cure RT + postcure 140C/1h

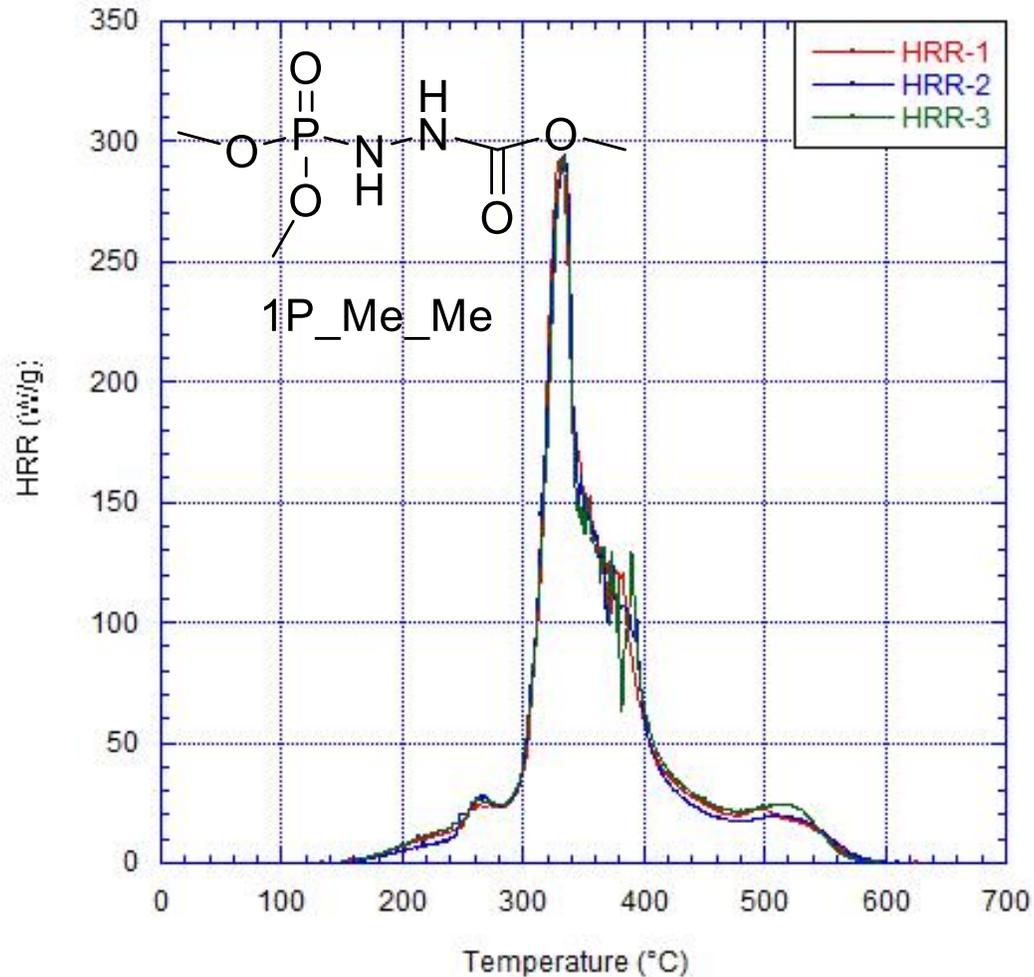


**Epon 862 + FR (P\_Et\_H)**  
cure 120C/2h + postcure 140C/1h

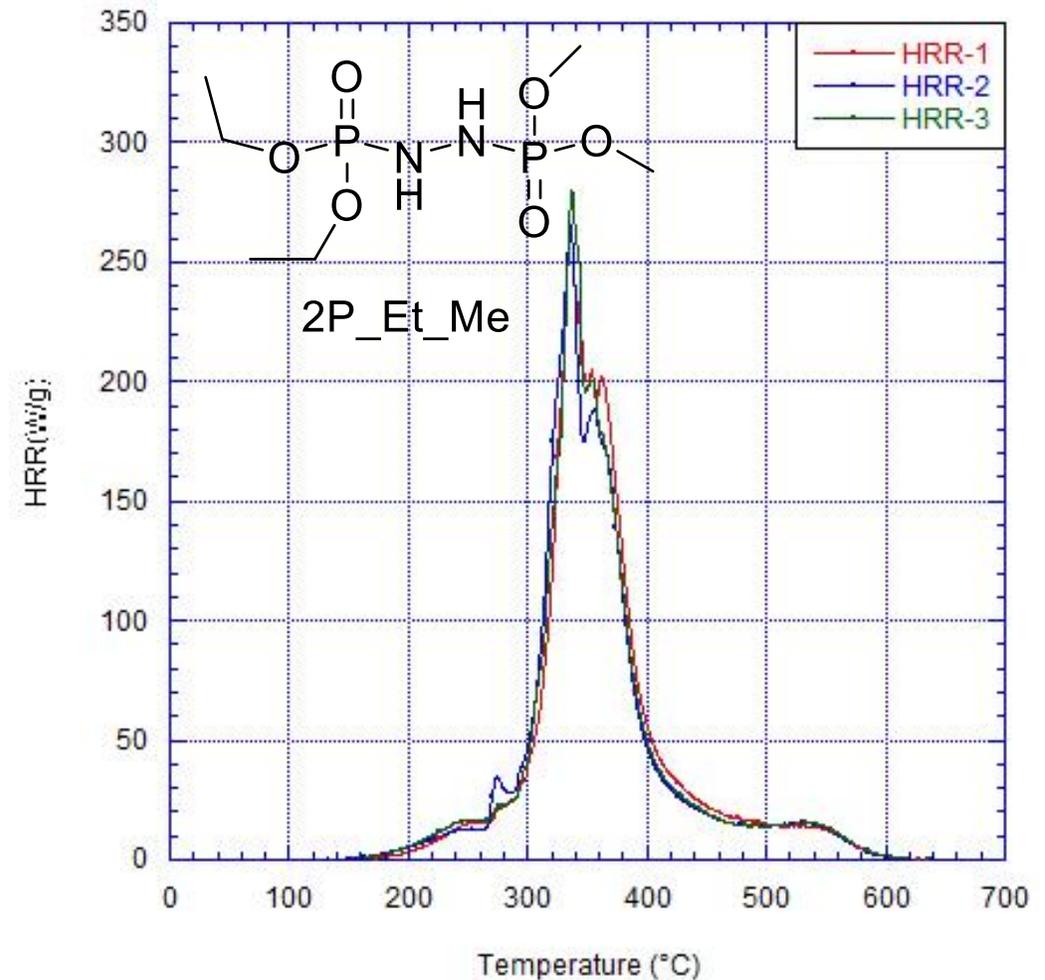


# Heat Release Curves – MCC Testing

**Epon/Epikure + 1-P-Me-Me**  
(cure RT + 140C/1 hr)

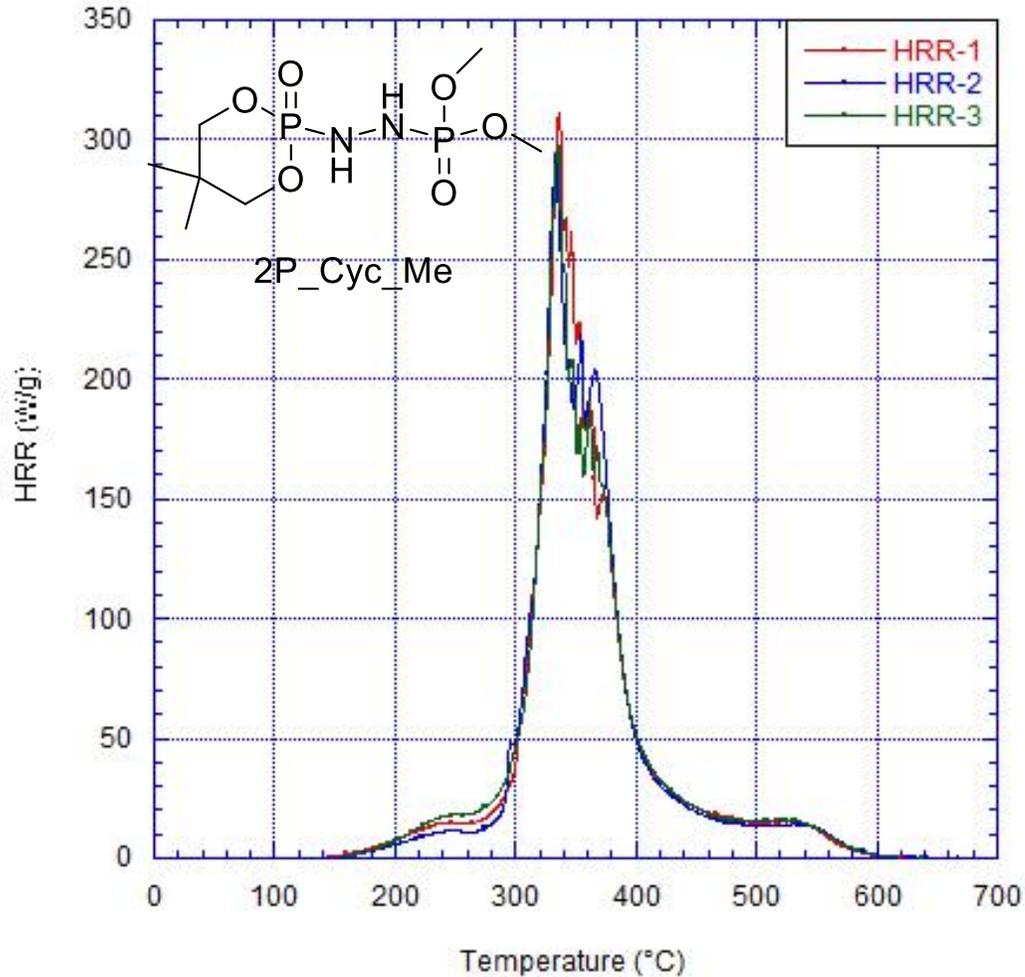


**Epon/Epikure + 2P-Et-Me**  
(cure RT + 140C/1hr)

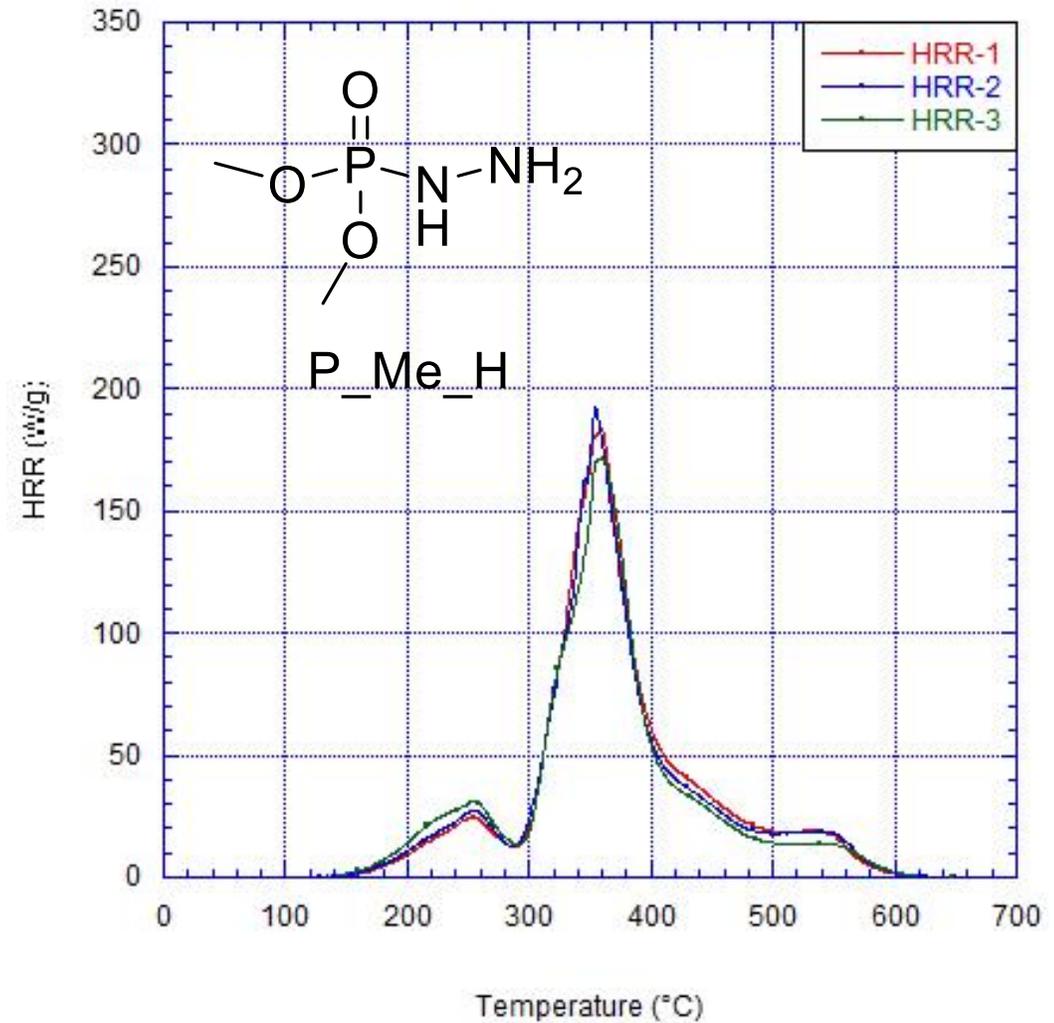


# Heat Release Curves – MCC Testing

**Epon/Epikure + 2P-cyc-Me**  
(cure RT + 140C/1hr)

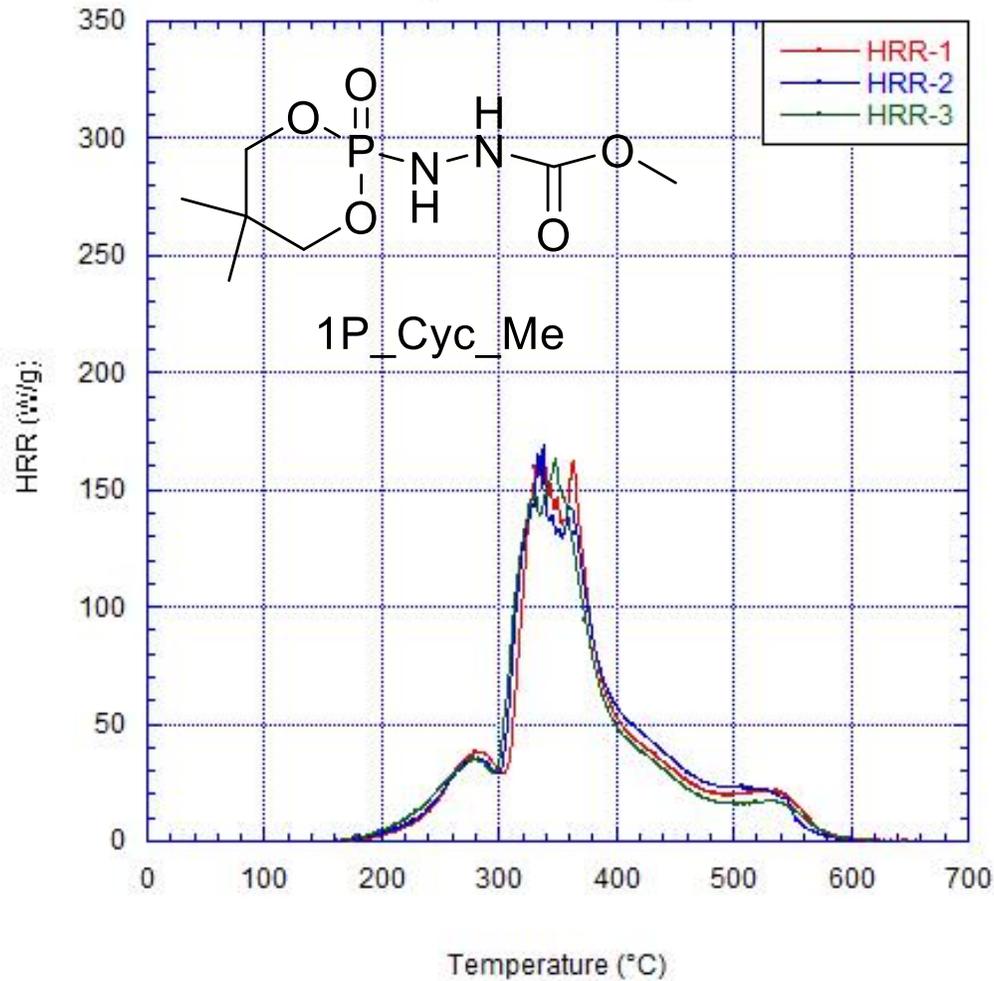


**Epon/Epikure + P-Me-H**  
(cure RT + 140C/1hr)

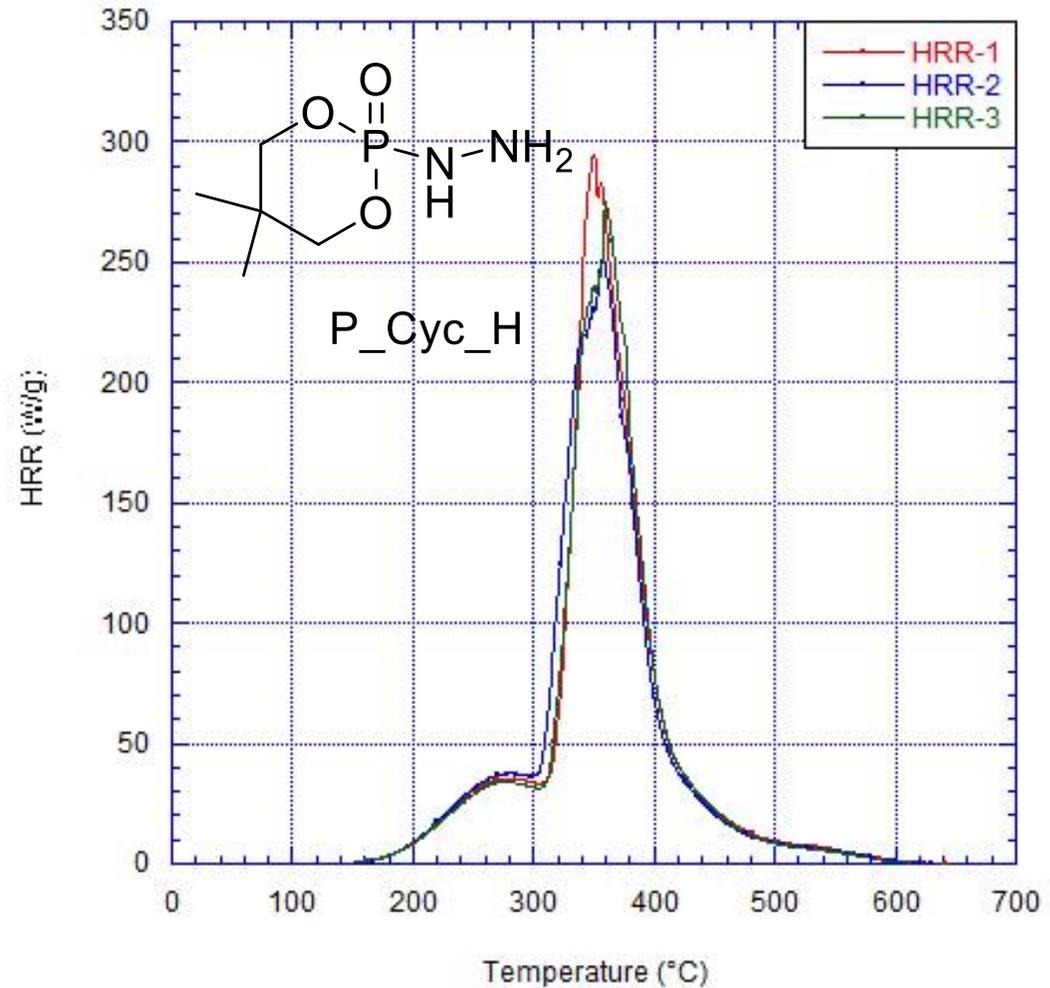


# Heat Release Curves – MCC Testing

**Epon/Epikure + 1P-cyc-Me**  
(cure RT + 140C/1hr)

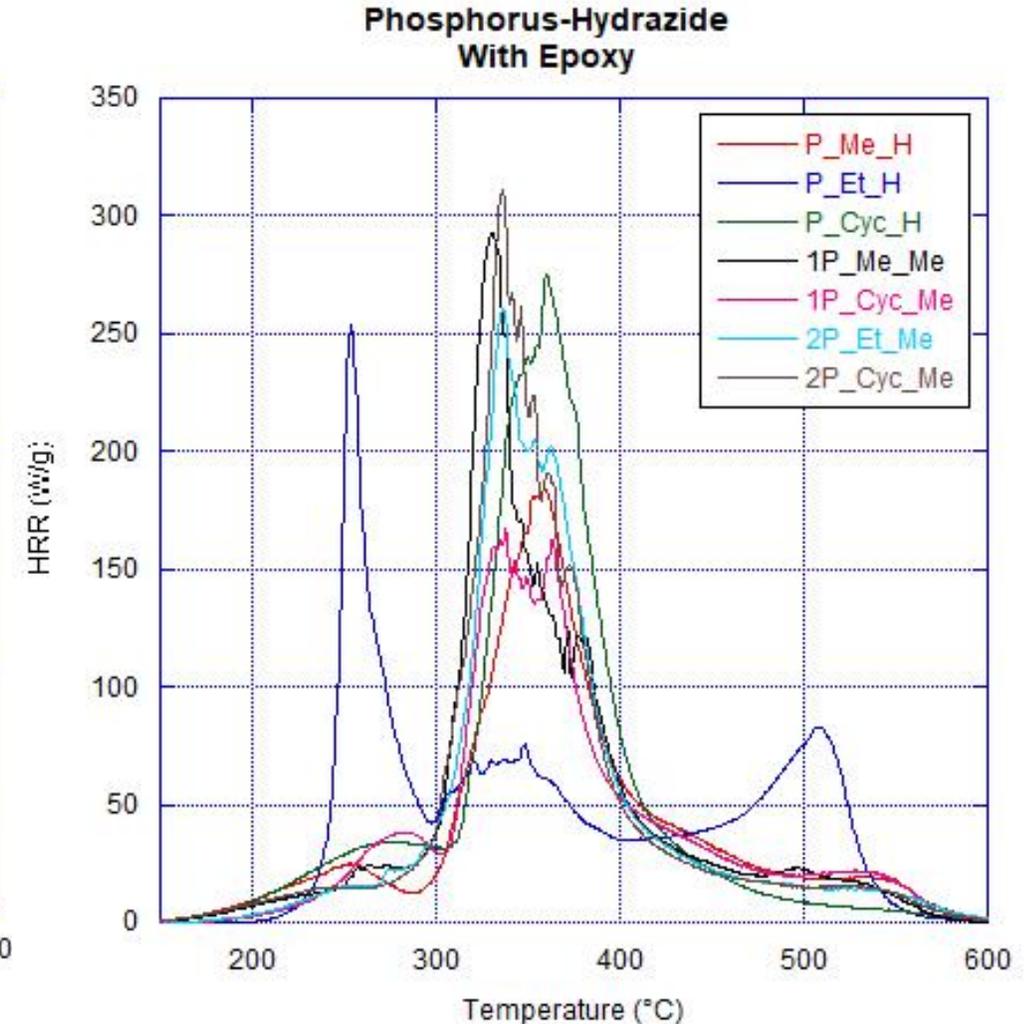
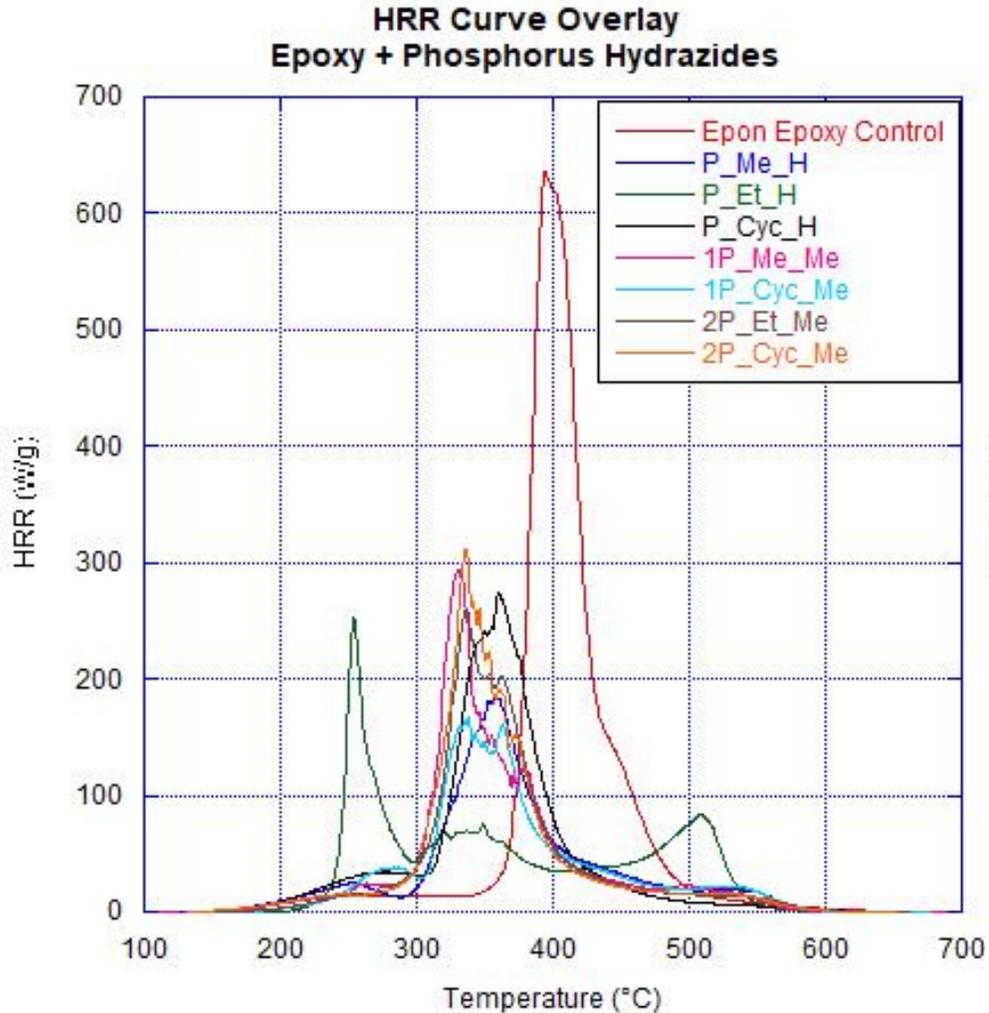


**Epon/Epikure + P-cyc-H**  
(cure RT + 140C/1hr)



# MCC Summary of Results

- Phosphorus-Hydrazides show reductions in HRR
- Greatest reductions in HRR selected for scale-up for cone calorimeter testing



# Cone Calorimeter Testing

- Best candidates from MCC testing scaled up and tested via cone calorimeter.
  - P-Et-H, P-Me-H easily scaled up, but P-Me-H interfered with epoxy curing – too reactive.
  - 1P-Cyc-Me could not be scaled up
  - 2P-Et-Me could be scaled up and was tested, even though MCC did not show superior performance.
- P-Et-H, 2P-Et-Me blended into Epon 862 with Epikure 3274
- Cone calorimeter testing at 35 kW/m<sup>2</sup> heat flux, 3mm thick, with and without frame and grid.
  - Samples showed some deformation during burning – used frame and grid to force them to lay flat during testing.

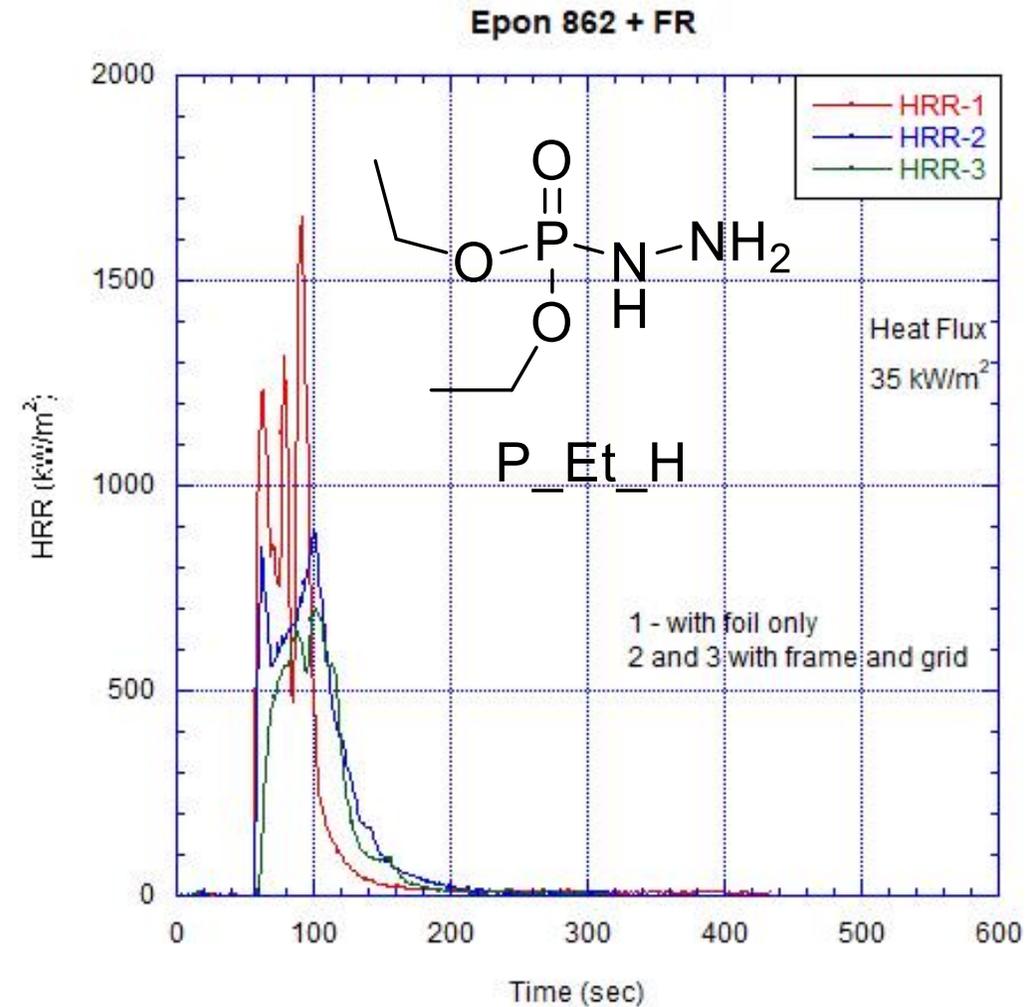
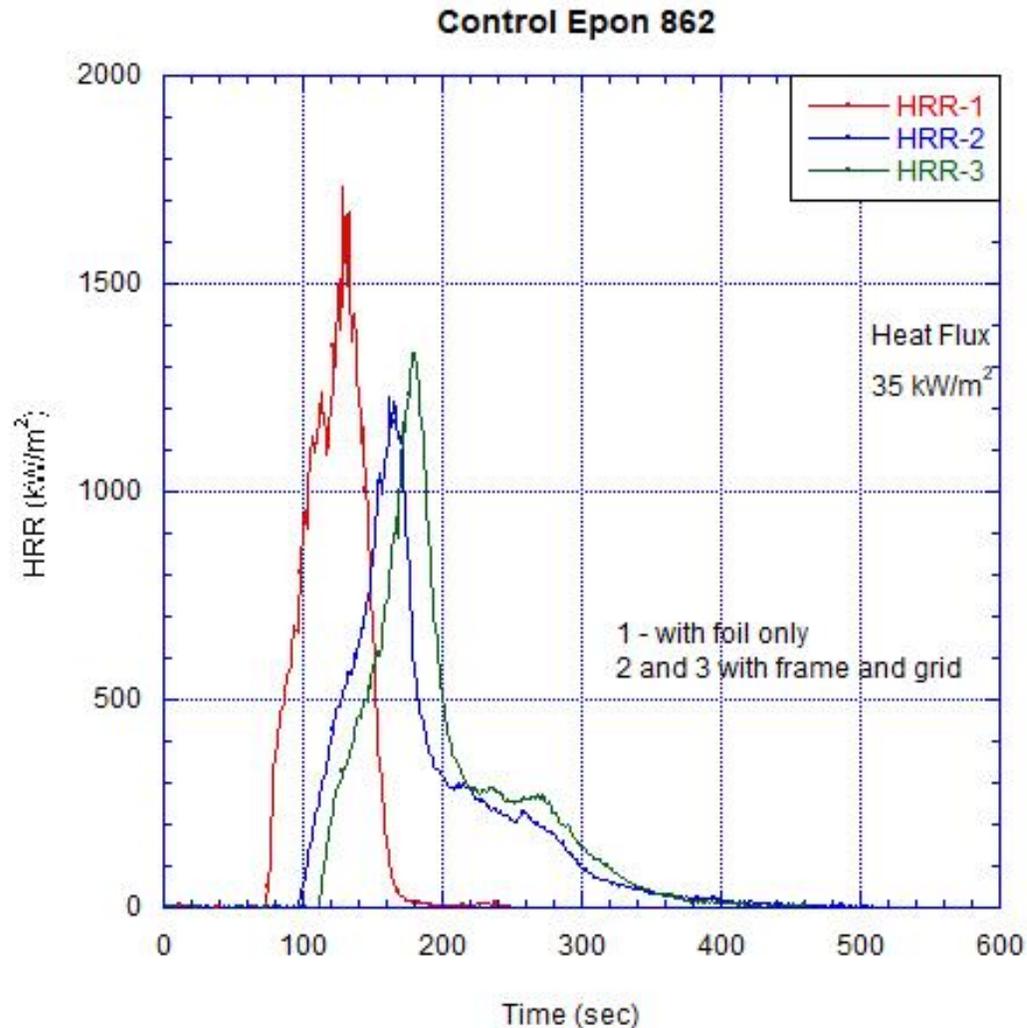
# Cone Calorimeter Testing

Sample Description	Time to ignition (s)	Peak HRR (kW/m <sup>2</sup> )	Time to Peak HRR (s)	Average HRR (kW/m <sup>2</sup> )	Weight % Lost (%)	Total Heat Release (MJ/m <sup>2</sup> )	Total smoke Release (m <sup>2</sup> /m <sup>2</sup> )	Avg. Effective Heat of Comb. (MJ/kg)	MARHE (kW/m <sup>2</sup> )
Epoxy Control	69	1735	129	718	93.6	78.3	2054	24.63	494
	89	1227	162	250	92.2	87.6	2690	24.33	304
	105	1334	178	297	91.1	91.0	2681	24.98	300
<b>Average Data</b>	<b>88</b>	<b>1432</b>	<b>156</b>	<b>422</b>	<b>92.3</b>	<b>85.6</b>	<b>2475</b>	<b>24.65</b>	<b>366</b>
Epoxy + FR (P-Et-H)	50	1654	91	152	79.9	51.3	1623	17.99	432
	49	889	100	260	68.8	50.1	1713	19.26	340
	53	700	101	220	55.0	38.6	1330	17.68	268
<b>Average Data</b>	<b>51</b>	<b>1081</b>	<b>97</b>	<b>211</b>	<b>67.9</b>	<b>46.7</b>	<b>1555</b>	<b>18.31</b>	<b>347</b>
Epoxy + FR (2P-Et-Me)	57	987	120	183	73.0	71.7	2169	20.48	355
	54	772	131	251	68.9	63.5	2060	20.45	315
	55	826	111	209	72.0	63.1	2149	21.03	330
<b>Average Data</b>	<b>55</b>	<b>862</b>	<b>121</b>	<b>214</b>	<b>71.3</b>	<b>66.1</b>	<b>2126</b>	<b>20.65</b>	<b>334</b>

- No strong reduction in peak HRR, MARHE
- Notable reductions in total HR, total smoke, average effective heat of combustion

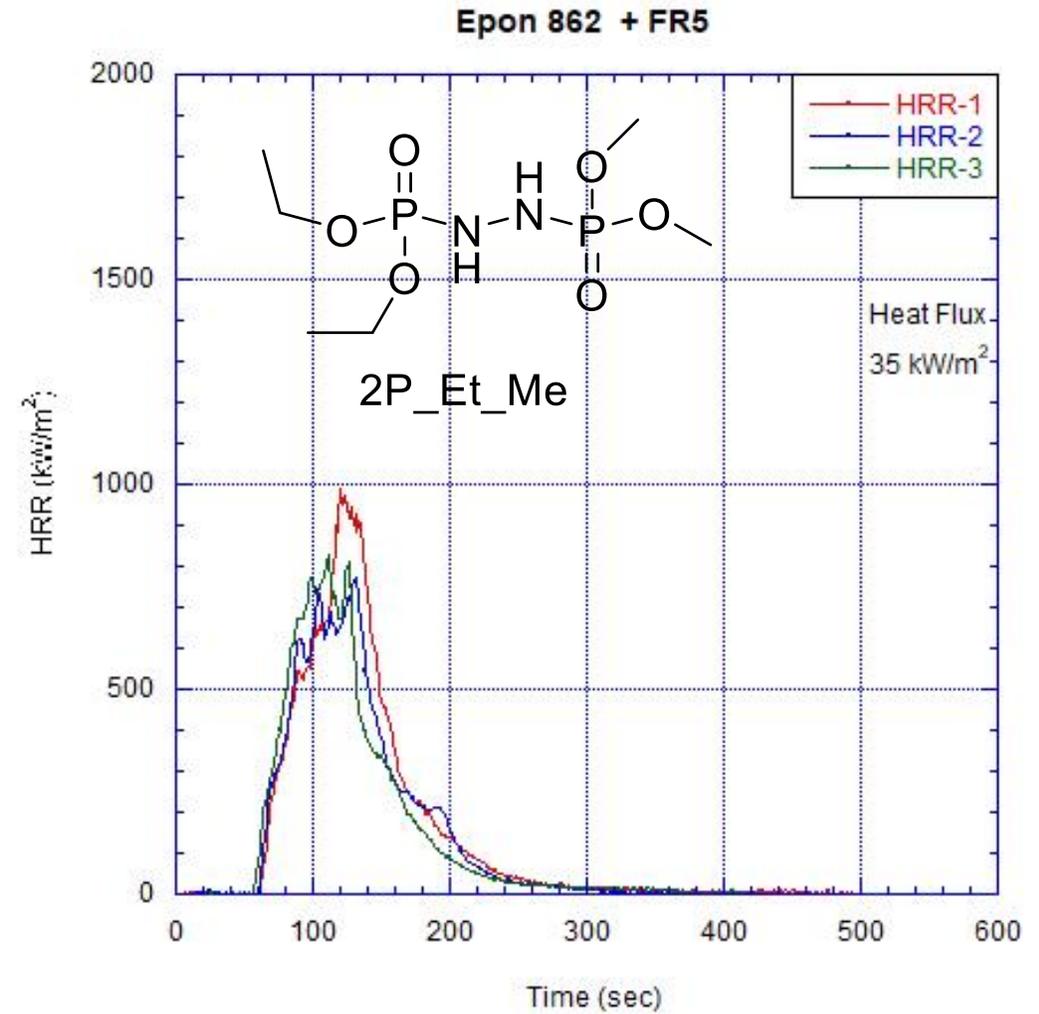
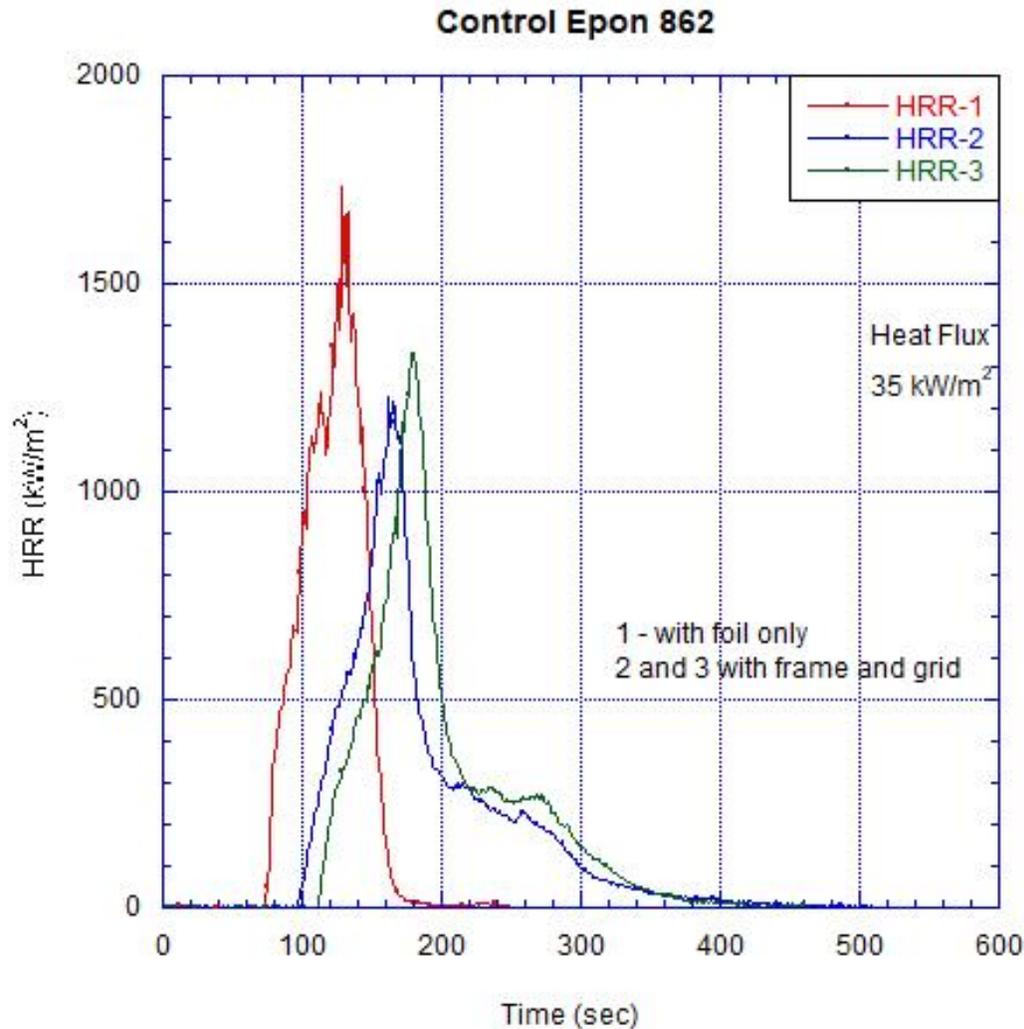
# Cone Calorimeter Results

- Erratic results due to strong physical effects of burning



# Cone Calorimeter Results

- Erratic results due to strong physical effects of burning



# Cone Calorimeter Chars

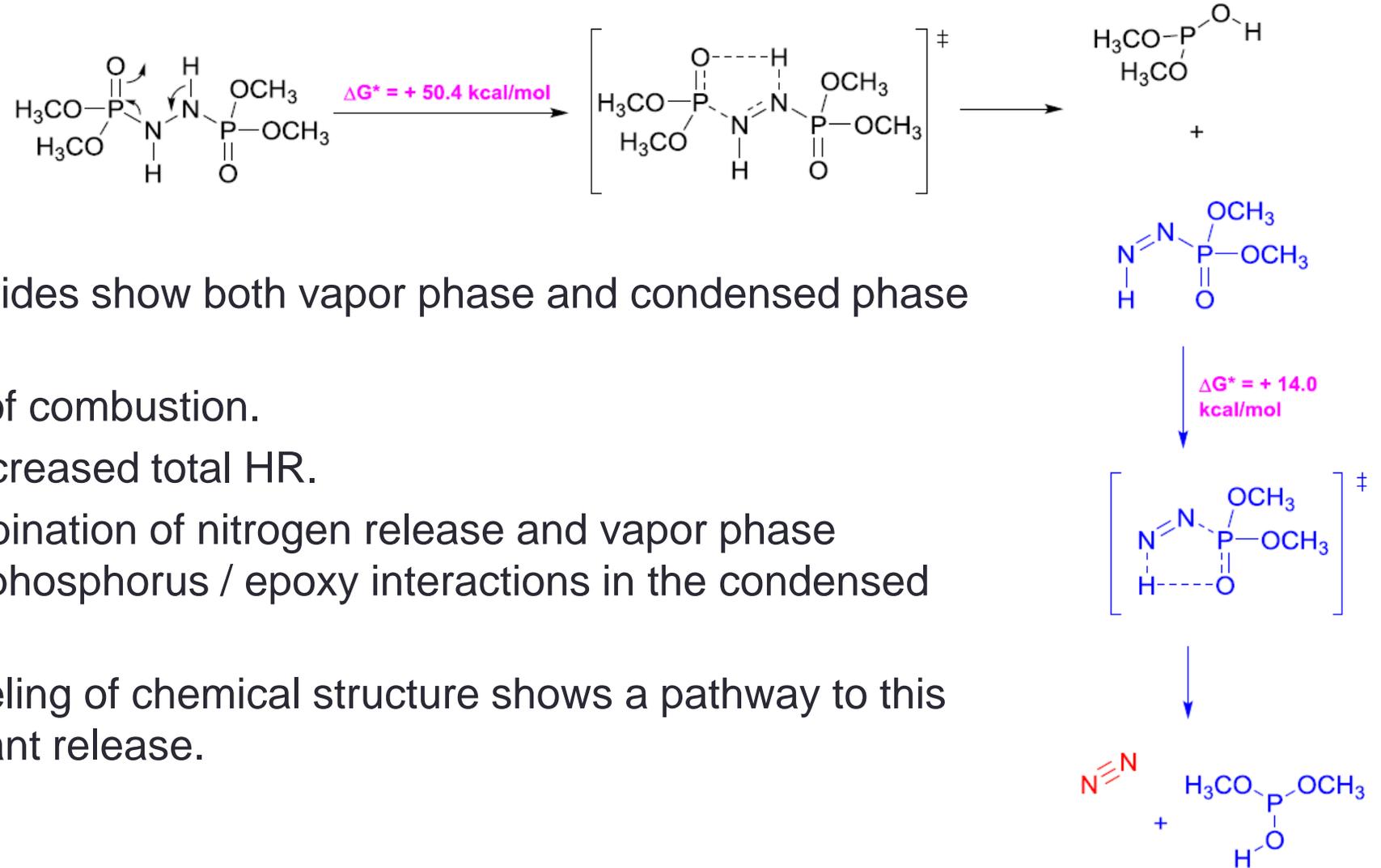


Control Sample

Epoxy + P-Et-H

Epoxy + 2P-Et-Me

# Conclusions



- Organophosphorus Hydrazides show both vapor phase and condensed phase flame retardancy:
  - Reduced effective heat of combustion.
  - Increased char yield, decreased total HR.
  - Mechanism likely a combination of nitrogen release and vapor phase phosphorus, along with phosphorus / epoxy interactions in the condensed phase.
    - Thermodynamic modeling of chemical structure shows a pathway to this potential flame retardant release.

# Conclusions

- Peak HRR, MARHE not reduced as much as desired.
  - Unexpected benefit: reduction in smoke release
- Chemistry may be useful to combine with other flame retardants.
- Further study needed to verify utility, effect on epoxy T<sub>g</sub>, and reactivity into epoxy matrix.
  - Epoxy reactivity assumed based upon known chemical interactions between epoxy, aliphatic amines, and phosphorus esters at elevated temperatures.

# Acknowledgements

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  - Abdhulhamid bin Sulayman (graduated Master's Student)

# Questions?

