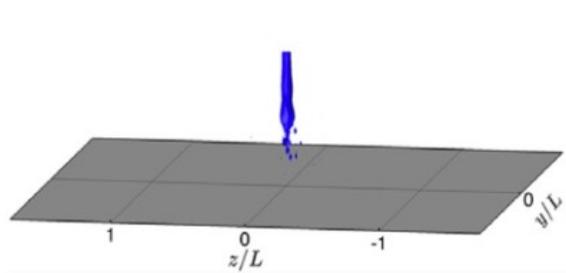


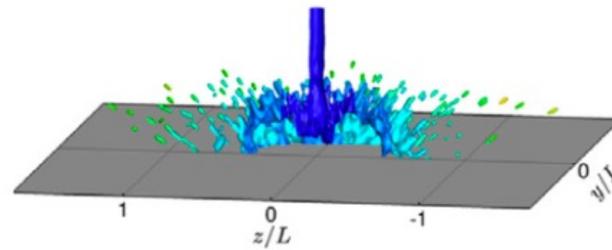
Hot-Surface Ignition of Fuel Sprays in Aircraft Compartment Fires

A COMPUTATIONAL & EXPERIMENTAL ANALYSIS

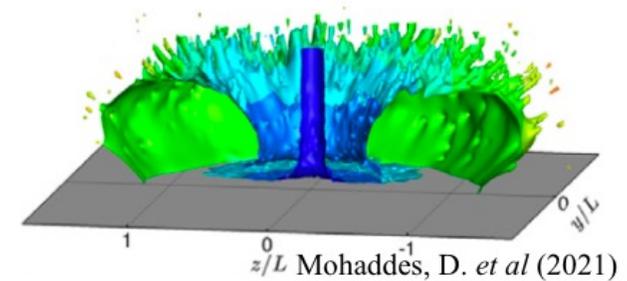
Lauren Simitz, Danyal Mohaddes, Jesus Cervantes,
Guillaume Vignat, Matthias Ihme



$t/\tau = 1.1$



$t/\tau = 3.3$



$t/\tau = 16.7$

Acknowledgements: Philipp Boettcher, Brad Moravec, Jason Damazo

Outline

■ Introduction

- Motivation
- Physics Overview
- Research Challenges, Objectives

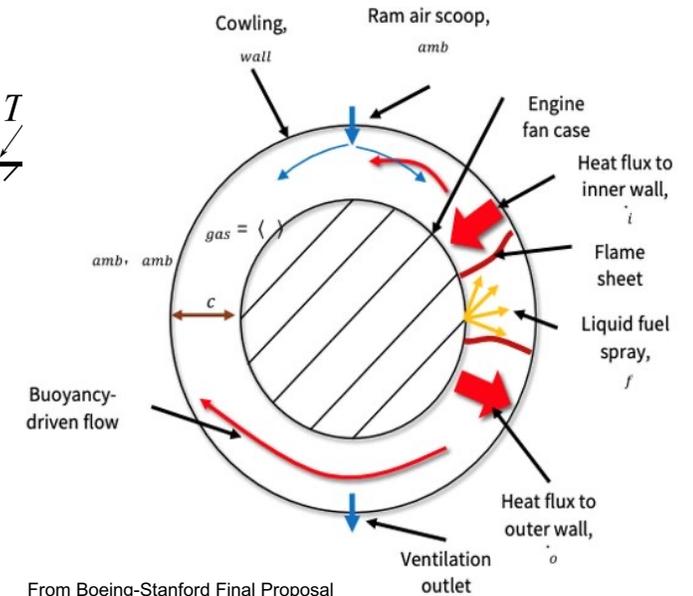
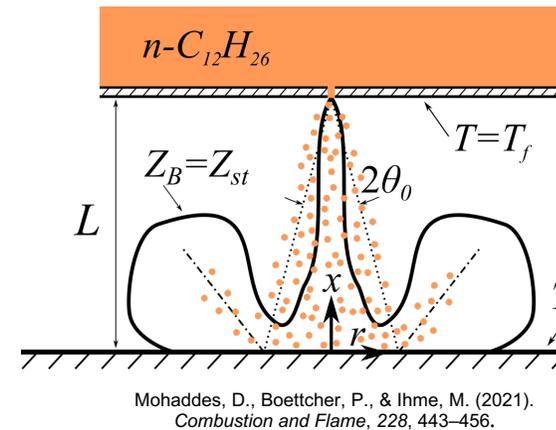
■ Geometric & Time Scale Analysis

■ Simulations

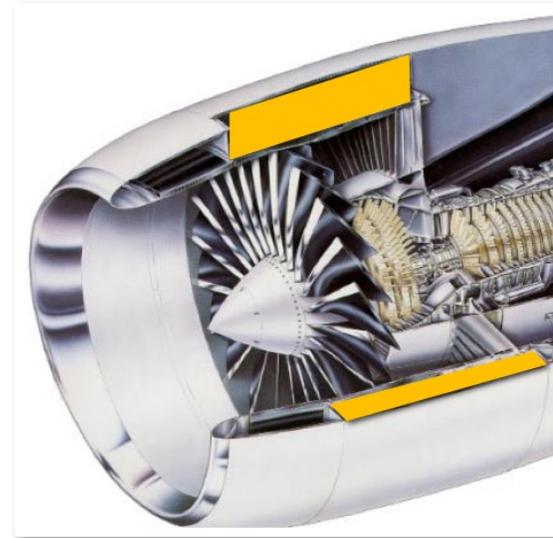
- Detailed (3D) Modeling
- Reduced-Dimension (1D) Modeling

■ Conclusion

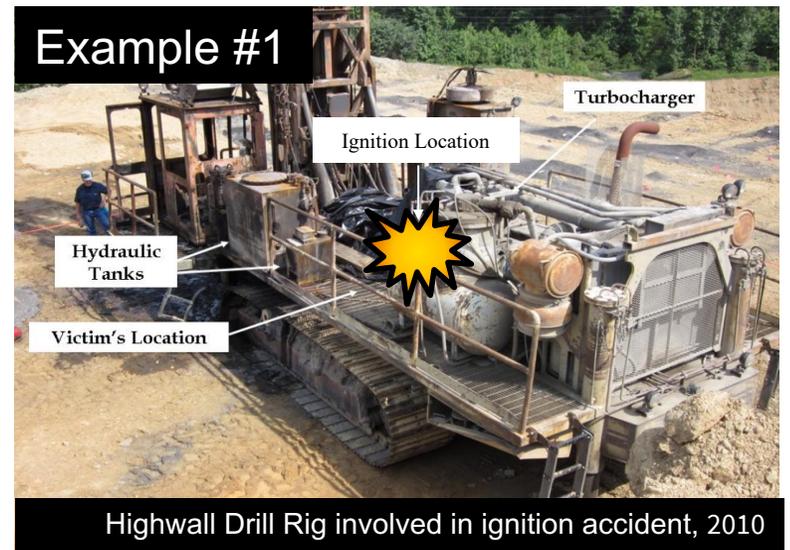
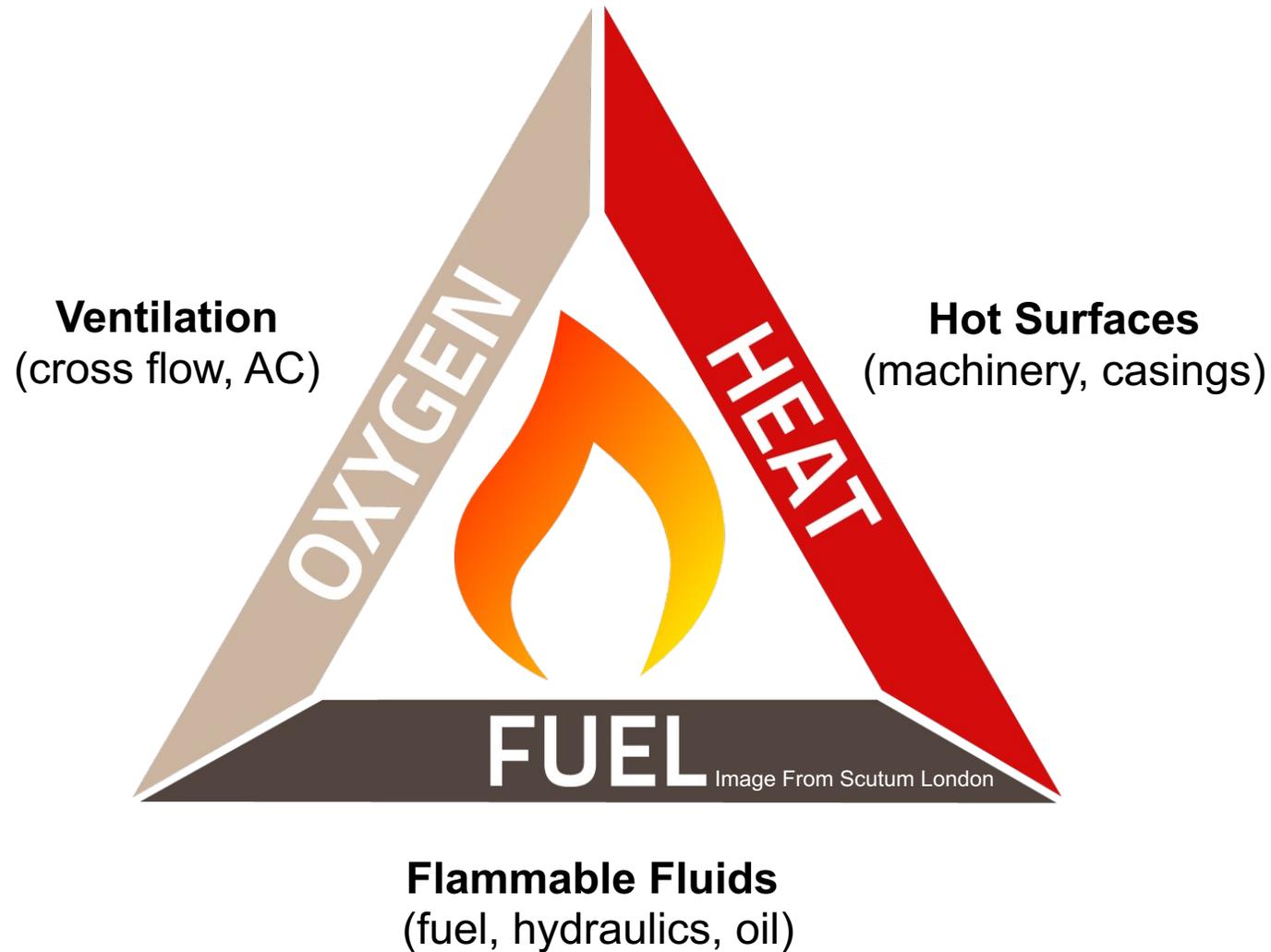
- Key Findings
- Future Work



Introduction



Motivation | Accidental Fires



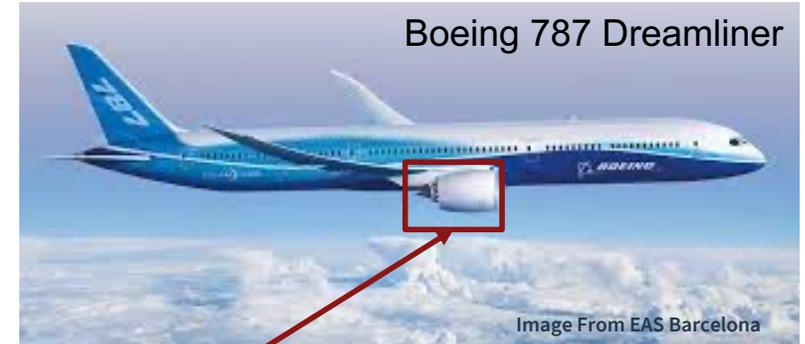
Fatal Ignition Investigation Report, Mining Safety & Health Agency



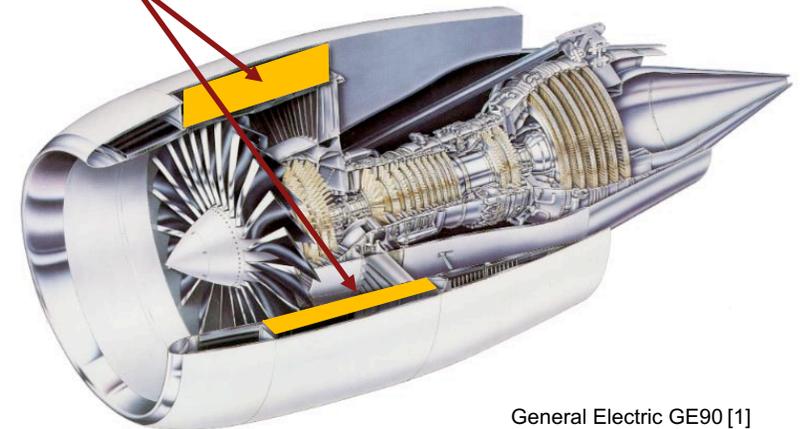
Final Report: Airbus A330 Engine Fire Event, AAIB Singapore

Motivation | Fire Safety Analysis in Engine Fan Case Compartment

- Small annular space between fan case and cowling
- Must be considered for aircraft fire safety analysis in event of a fuel line leak/breakage
- Fuel source:
 - Liquid Jet-A, $p_f \leq 8.3 \text{ Mpa}$
 - Hydraulic fluid, lube oil (future work)
- Air sources: leakage through cowling from aerodynamic pressure, natural ventilation
 - $30 \text{ kPa} \leq p \leq 101 \text{ kPa}$
 - $230 \text{ K} \leq T \leq 330 \text{ K}$
- Ignition sources: hot surfaces
 - Starter motor
 - Pumps, gearbox



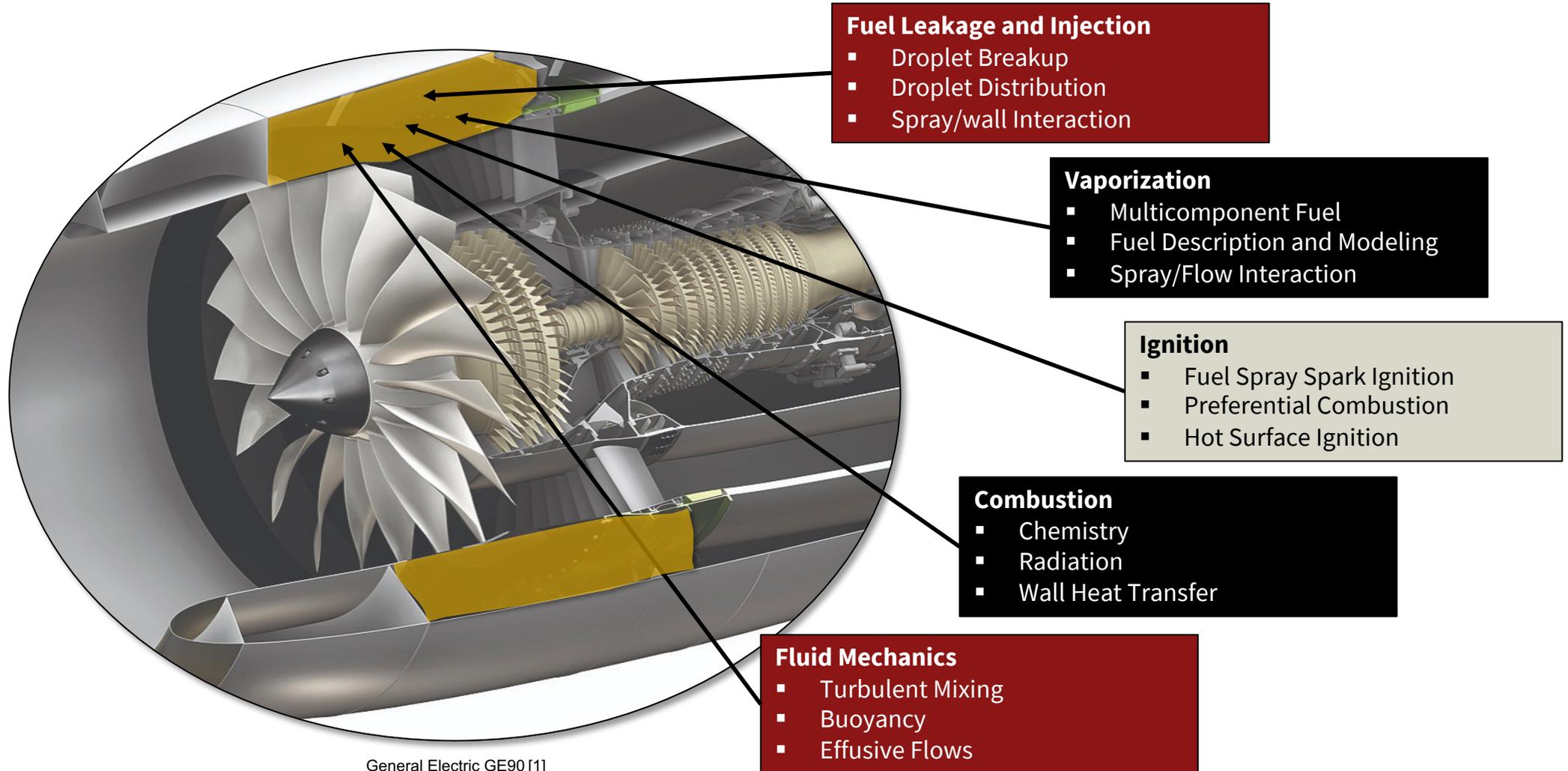
“Fan case compartment”



[1] Adapted from *Towards an Automated Full-Turbofan Engine Numerical Simulation*, NASA Report, 2003.

[2] *Final Report: Airbus A330 Engine Fire Event*, AAIB Singapore

Physics Overview



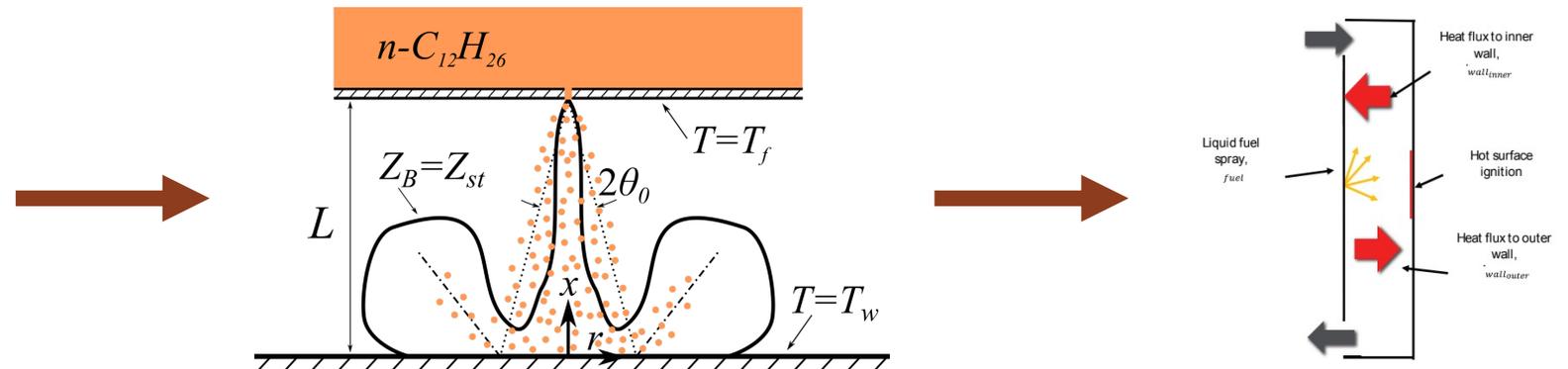
Research Challenges & Objectives

Research Challenges

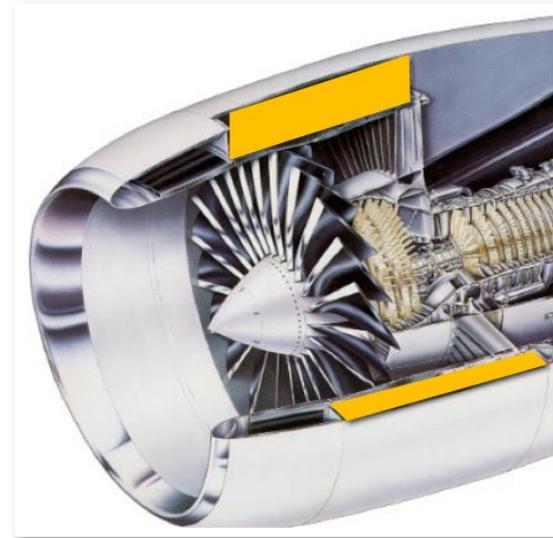
- Lack of fundamental understanding of multi-physics interaction in hot surface ignition (HSI)
- Lack of predictive models for HSI
- Need for simulation-informed fire safety certification for aircraft

Objectives

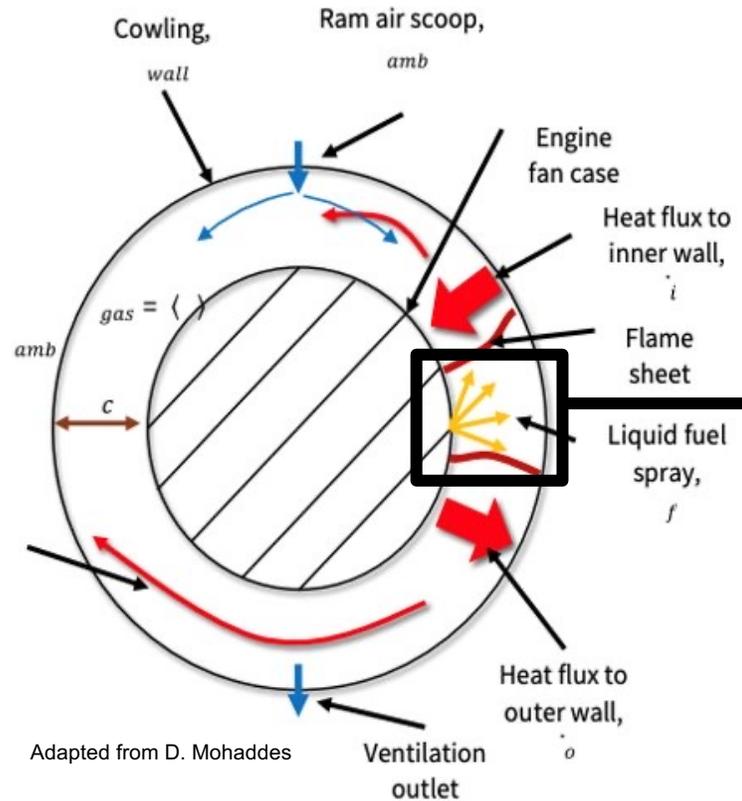
- Generate physics-based modeling tools to analyze and predict compartment fires
- Validate modeling tools with high realism, in-house experimental data



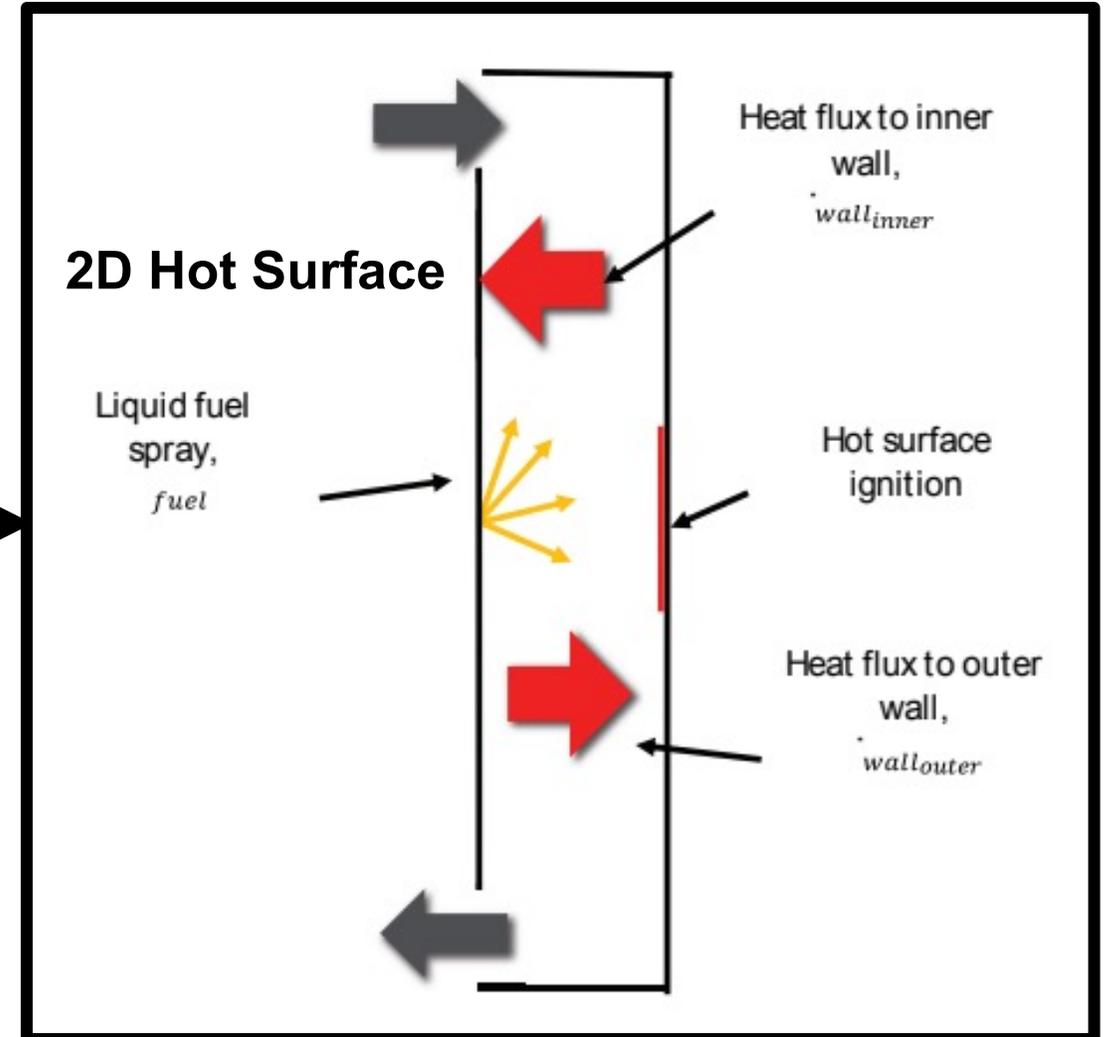
Geometric & Time Scale Analysis



Geometric Representation

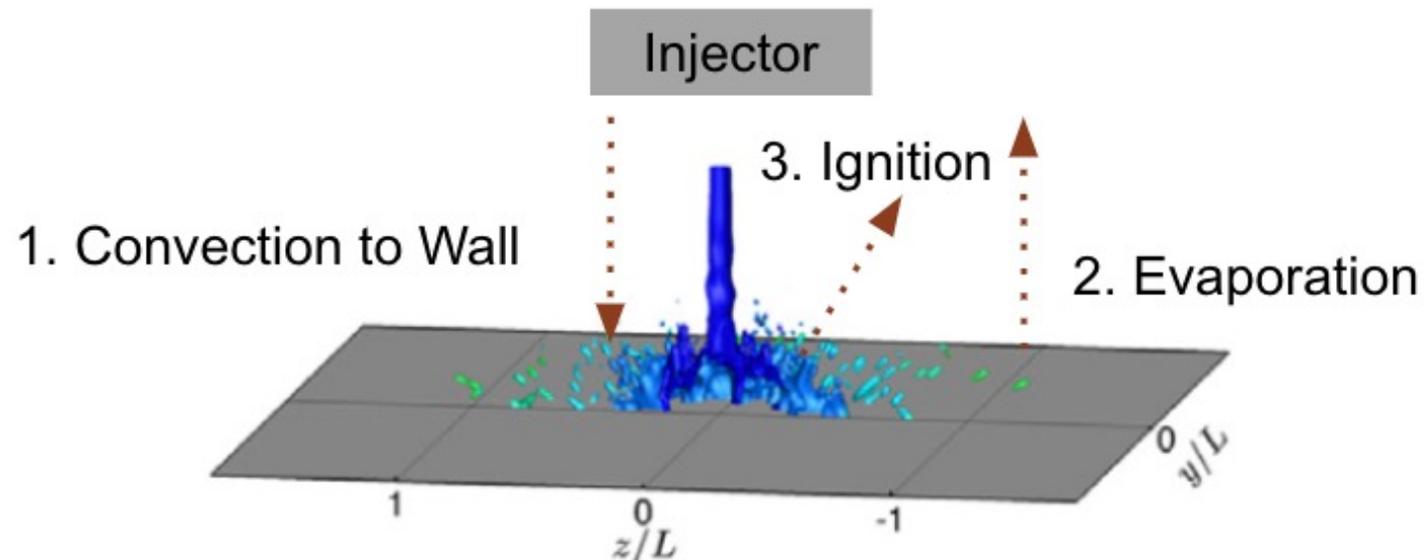
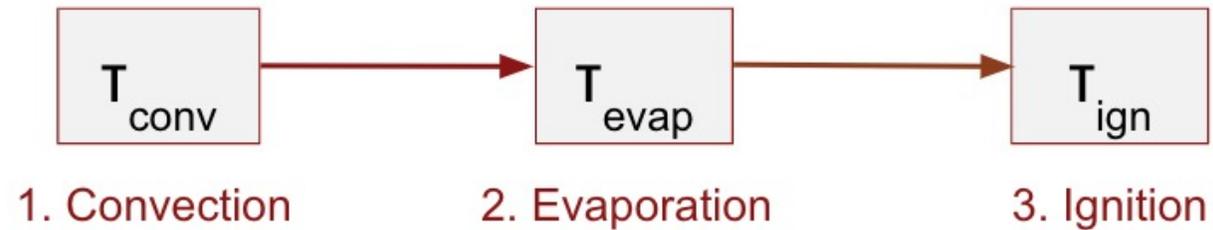


High-Realism Cylindrical Cowl



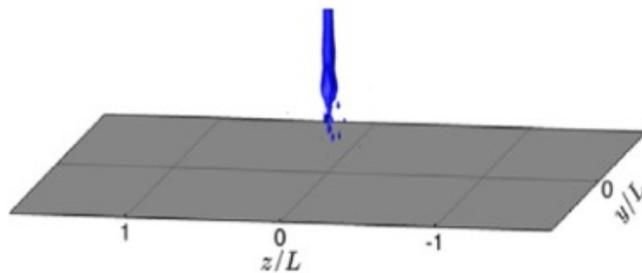
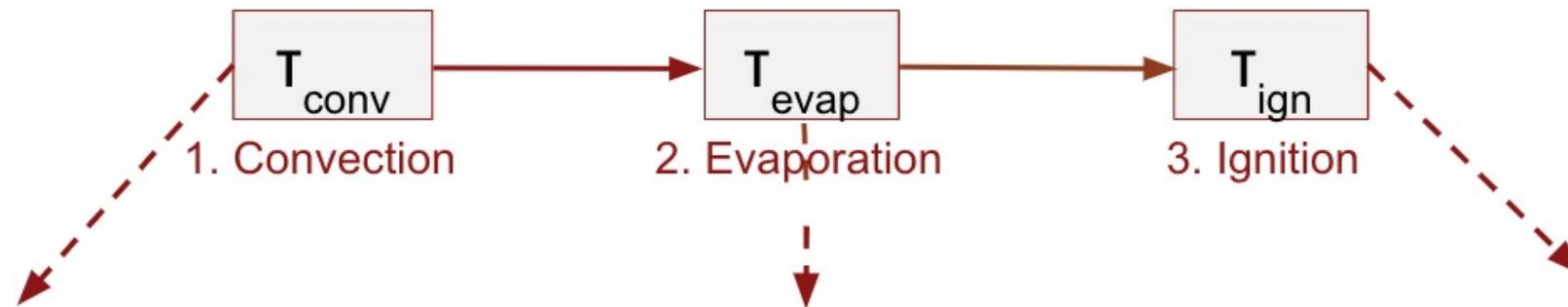
Time Scale Analysis

- We can consider the hot surface ignition of fuel spray impinging on a wall as an approximate three-step sequence:



Time Scale Analysis | Methodology

- We can consider the hot surface ignition of fuel spray impinging on a wall as an approximate three-step sequence:



Mohaddes, D., Boettcher, P., & Ihme, M. (2021). *Combustion and Flame*, 228, 443–456.

Convective Rate

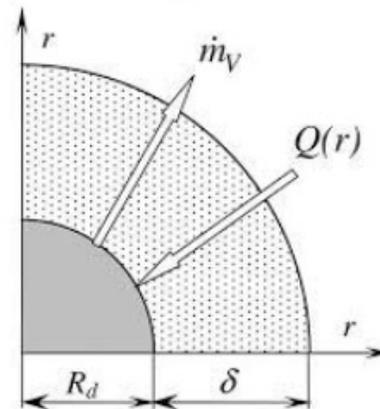
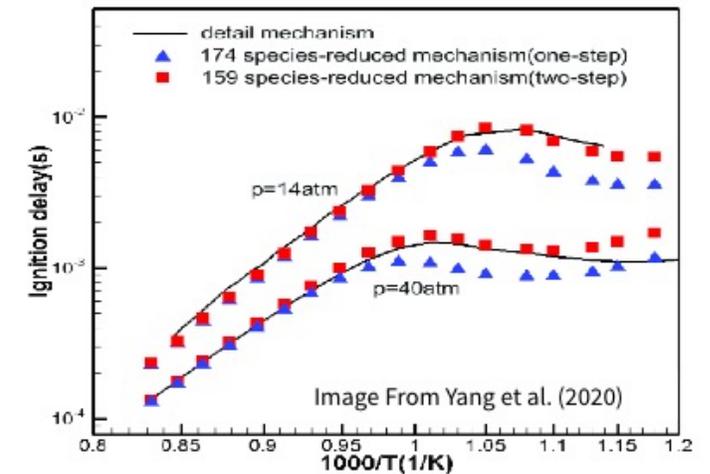


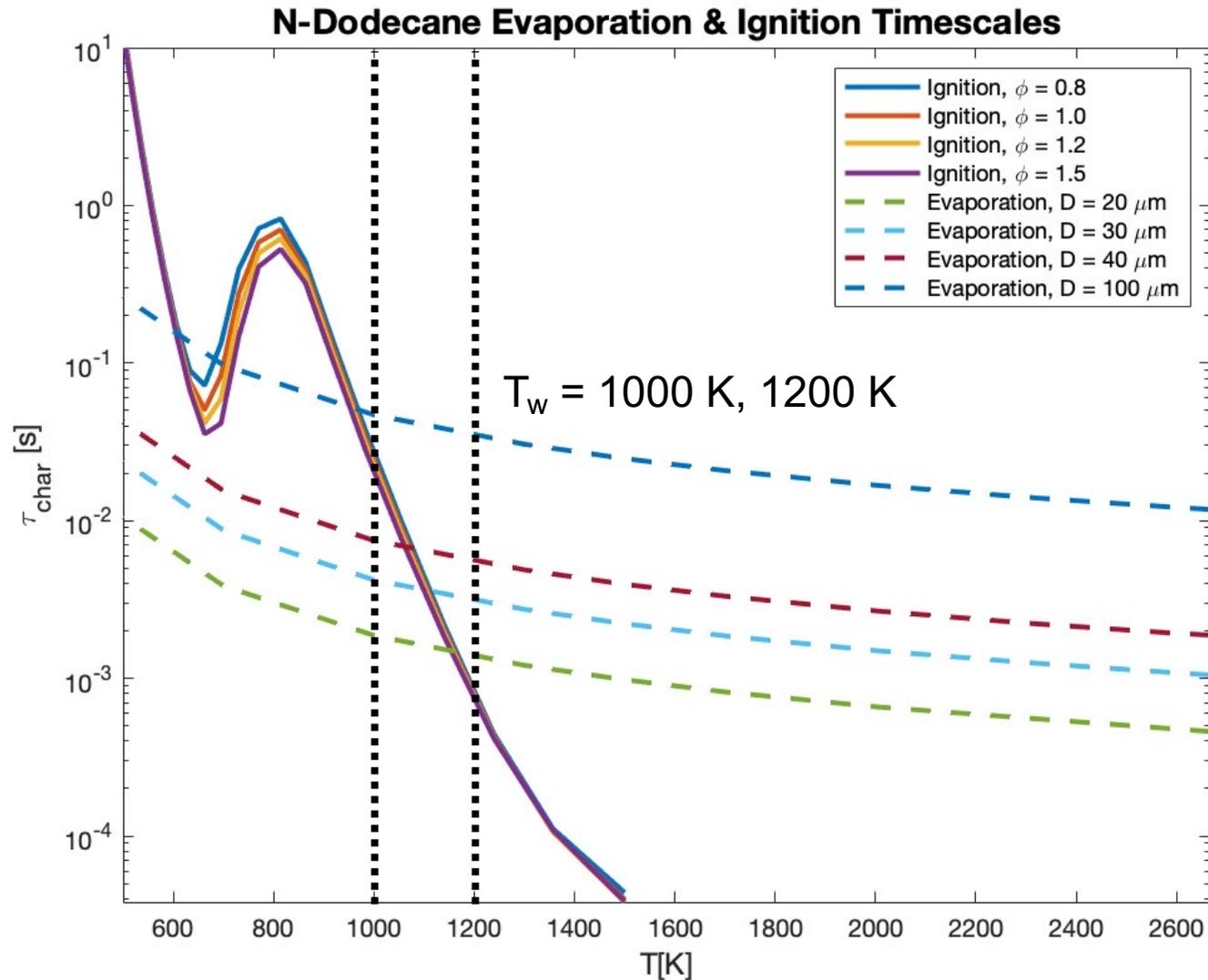
Image From Cambridge University Press

0D Droplet Evaporation Model



Ignition Delay Time

Time Scale Analysis | Results



- **Ignition Time Scale**

- 3×10^{-2} s (1000 K)
- 8×10^{-4} s (1200 K)

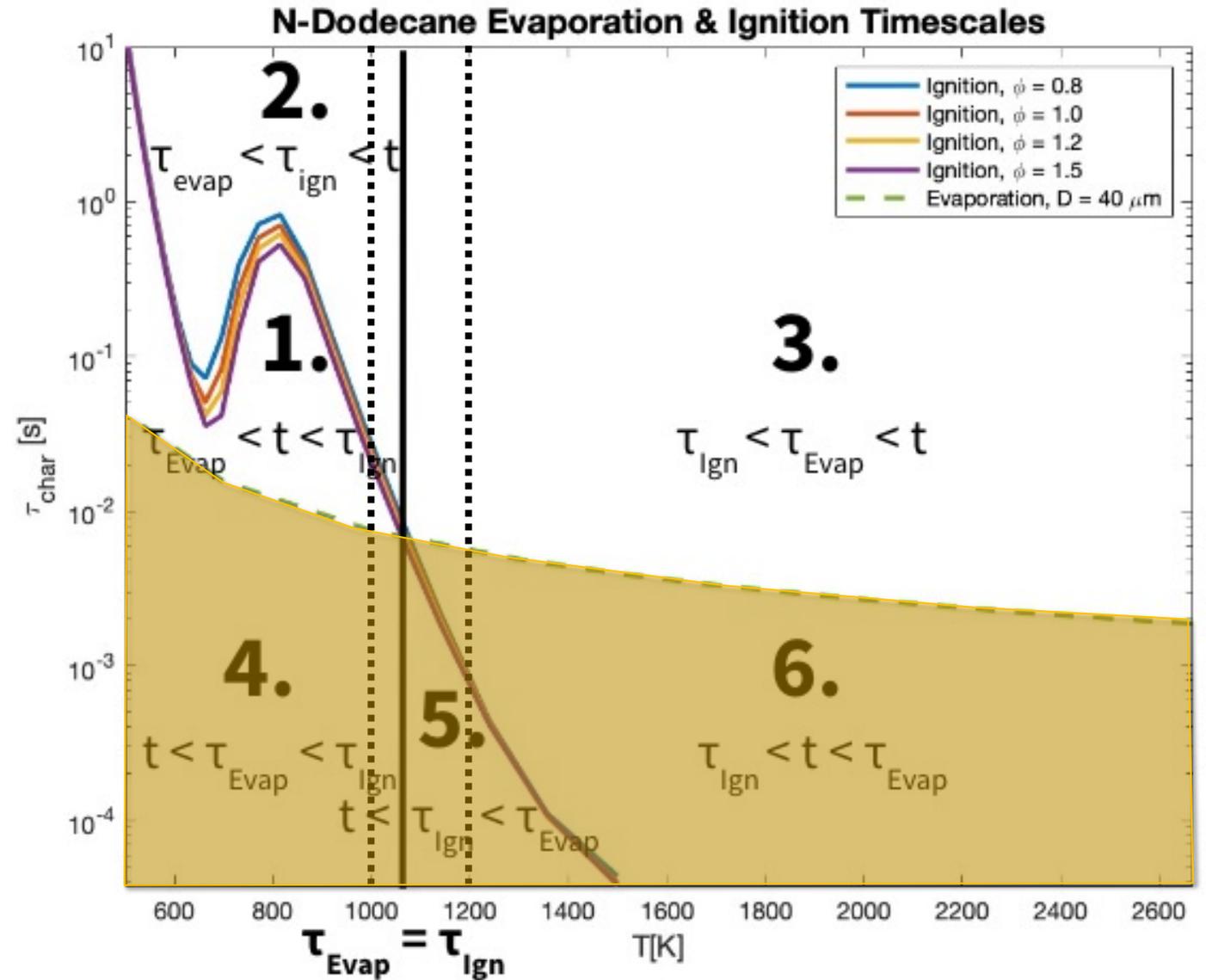
- **Evaporation Time Scale**

- 8×10^{-3} s (1000 K)
- 6×10^{-3} s (1200 K)

- For **experimental design**, balance timescales for ignition and evaporation
 - Fast enough so not affected by environment
 - Slow enough for diagnostics

Time Scale Analysis | Results

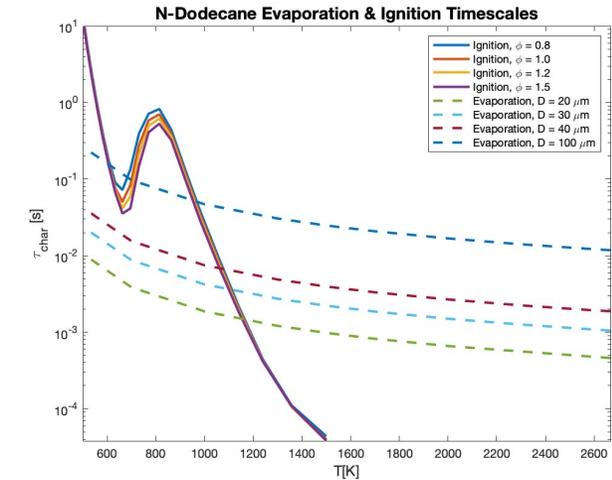
- Want droplets to convect and **hit the wall as a liquid** (t must $< t_{evap}$) → **Regions 4, 5, 6**
- Based on ignition timescale, can further separate into **ignition** and **flame-stabilized** regimes.



Time Scale Analysis | Bridge to Simulation Work

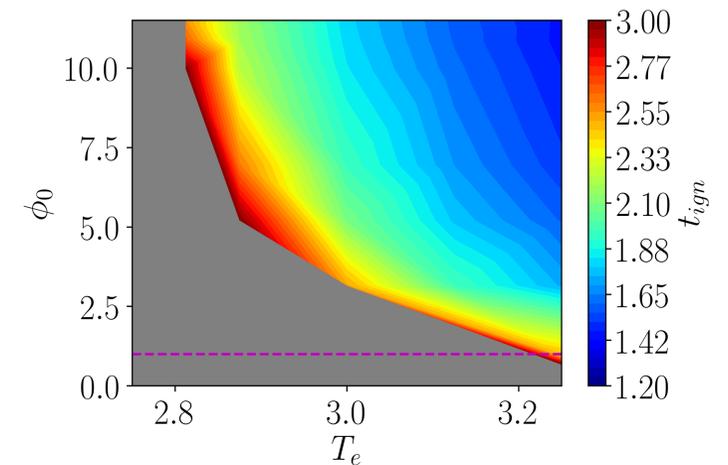
Outcomes

- Time scale analysis anticipates two regimes:
 - Ignition on short times
 - Stable, steady state combustion on long times
- Chronology of relevant processes important design consideration

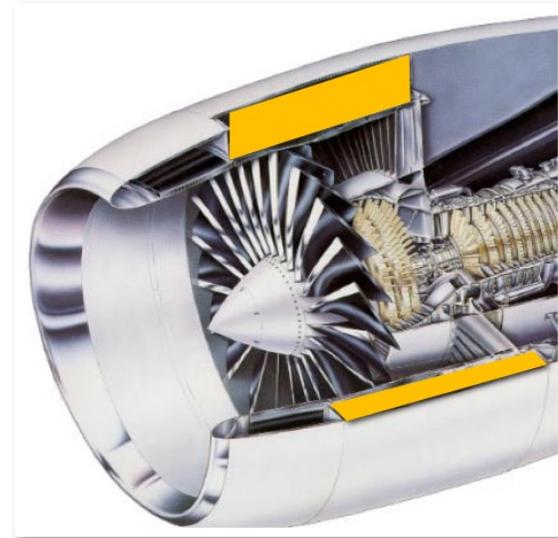


Research Objectives

- Perform **detailed simulations** to examine unsteady ignition process at high spatial and temporal resolution
- Perform **low order modeling** to explore comprehensive parameter space and identify safety-critical ignition scenarios (ex. ignition limits)

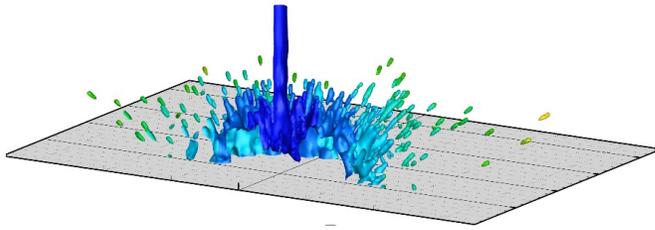


Simulations



Approach | 3D vs. 1D Simulations

3D Large-Eddy Simulations



Mohaddes, D., Boettcher, P., & Ihme, M. (2021). *Combustion and Flame*, 228, 443–456.

- Objectives
 - Resolve full flow field
 - Detailed combustion evolution
 - Estimates for surface temperature and heat flux
- Key Differences
 - Fully resolved flow-field simulation
 - Lagrangian spray model

1D Eulerian-Eulerian Simulations

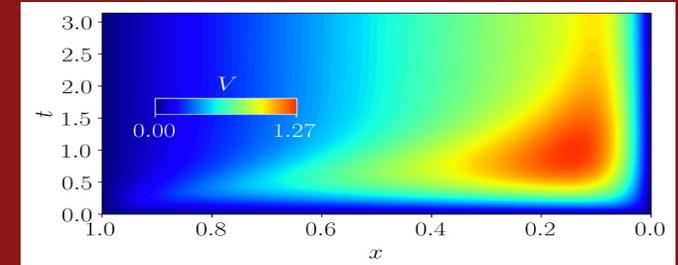
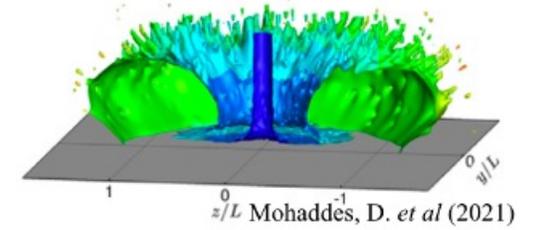


Image By Danyal Mohaddes

- Objectives
 - Describe phenomenology of spray-HSI
 - Analyze spray-HSI parametrically using non-dimensional variables
- Key Differences
 - Reduced dimensionality (1D)
 - Eulerian spray model (continuum)

3D Simulations | Methodology



Governing Equations

- Favre-filtered Navier-Stokes for large-eddy simulation of chemically reacting flows
- Lagrangian representation of spray

Chemistry Modeling

- 55-species chemical mechanism for *n*-dodecane/air combustion with low-temperature chemistry [1]
- Finite-rate chemistry with dynamic flame thickening

Numerical Method

- In-house finite-volume solver, nominally 4th order
- Splitting scheme for explicit/semi-implicit treatment of transport/reaction terms, 2nd order [2]

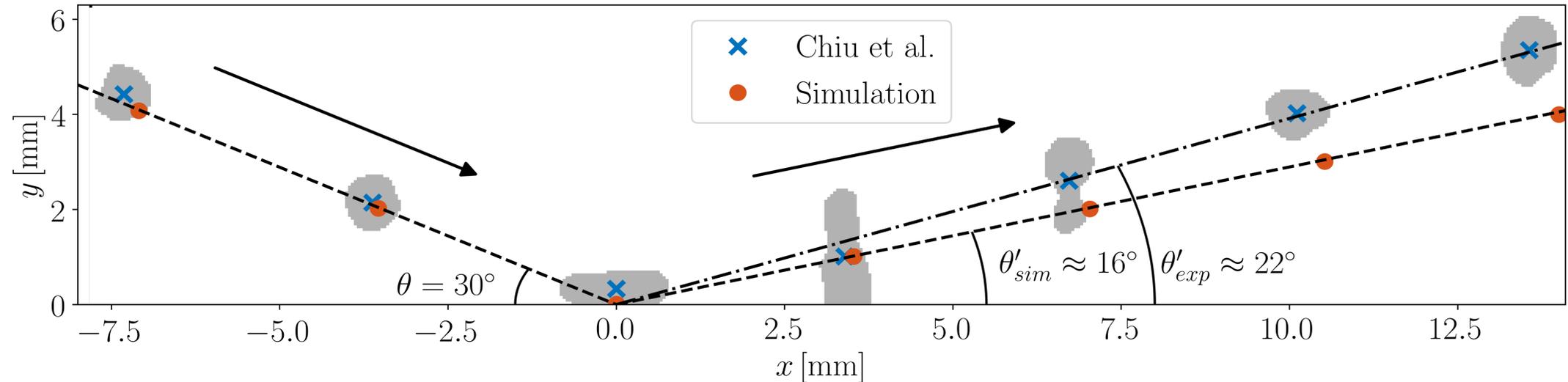
Mesh

- Hexahedral; uniform isotropic with $\Delta = 0.2\text{mm}$ in region of interest
- 1.7 million elements per axial quarter sector (6.8 million for full mesh)

[1] P. Ma, H. Wu, T. Jaravel, M. Ihme, "Large-eddy simulations of transcritical injection and auto-ignition using diffuse-interface method and finite-rate chemistry," PCI 2019

[2] H. Wu, P. Ma, M. Ihme, "Efficient time-stepping techniques for simulating turbulent reactive flows with stiff chemistry," Comput. Phys. Comm. 2019

3D Simulation | Evaluation of Spray-Wall Interaction

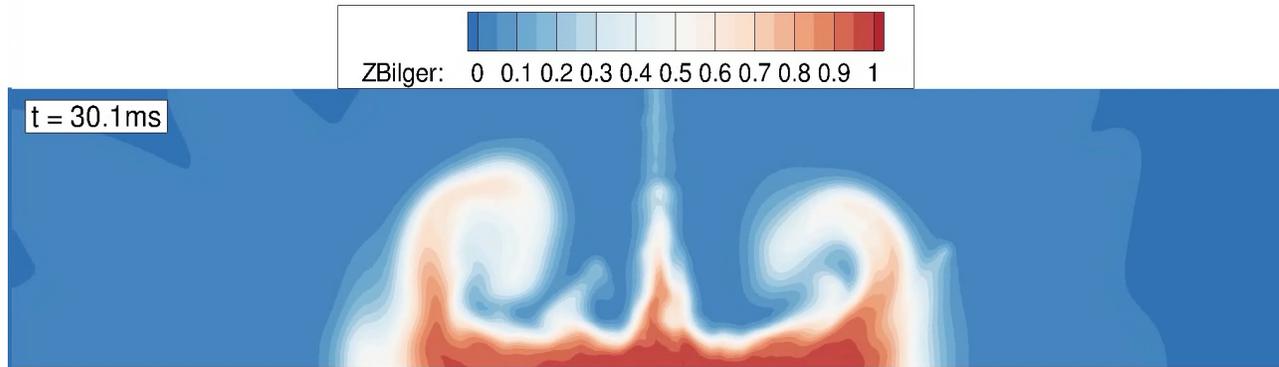


- For spray-wall interactions with $T_w > T_L$, droplets rebound inelastically
- Characterize coefficient of restitution L_n using impact Weber number We_n
 - $L_n = 0.263We_n^{0.257}$ [1], where $We_n = \frac{\rho_l u_{d,n}^2 D_d}{\sigma}$
- Simulated diesel spray-wall interaction experiment by Chiu et al. [2] to evaluate model performance
 - Acceptable accuracy, on order of spread in experimental data

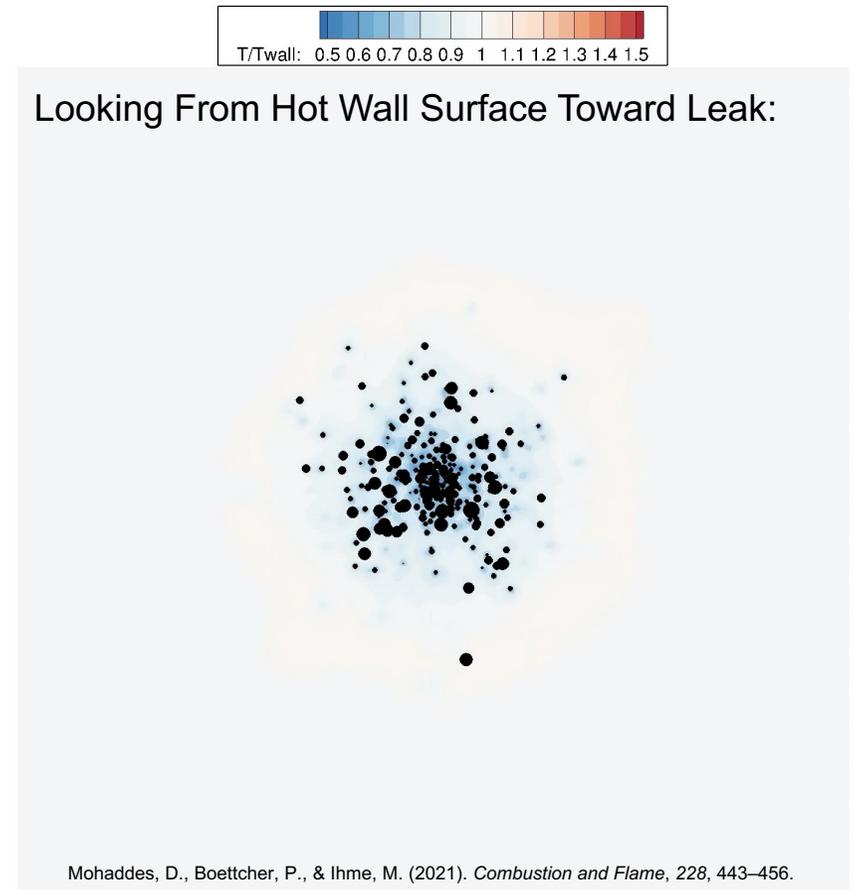
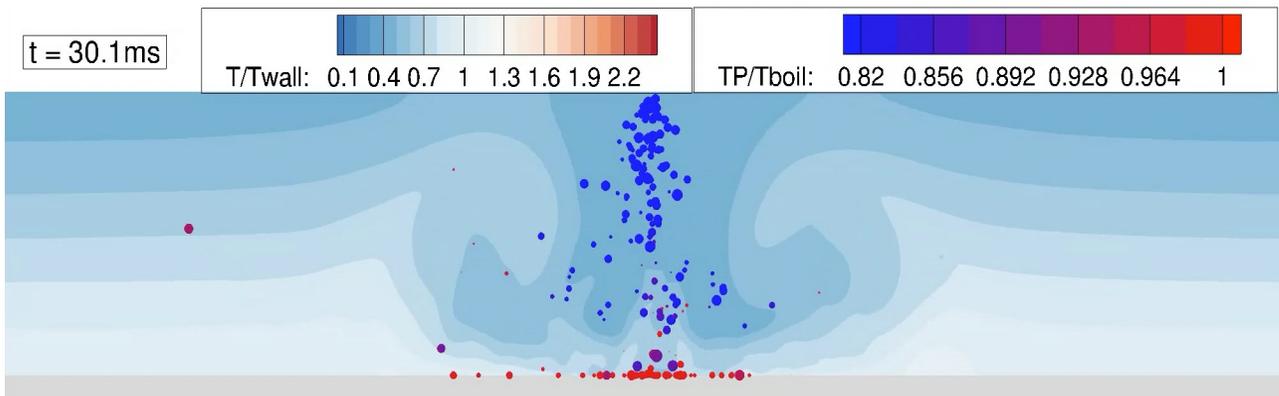
[1] Karl et al., "Experimental investigation of interaction processes between droplets and hot walls," Physics of Fluids, 2000.

[2] Chiu et al., "Experiment on the dynamics of a compound droplet impinging on a hot surface," Physics of Fluids, 2005.

3D Simulation | Evolution of Flow Structure



Mohaddes, D., Boettcher, P., & Ihme, M. (2021). *Combustion and Flame*, 228, 443–456.



Mohaddes, D., Boettcher, P., & Ihme, M. (2021). *Combustion and Flame*, 228, 443–456.

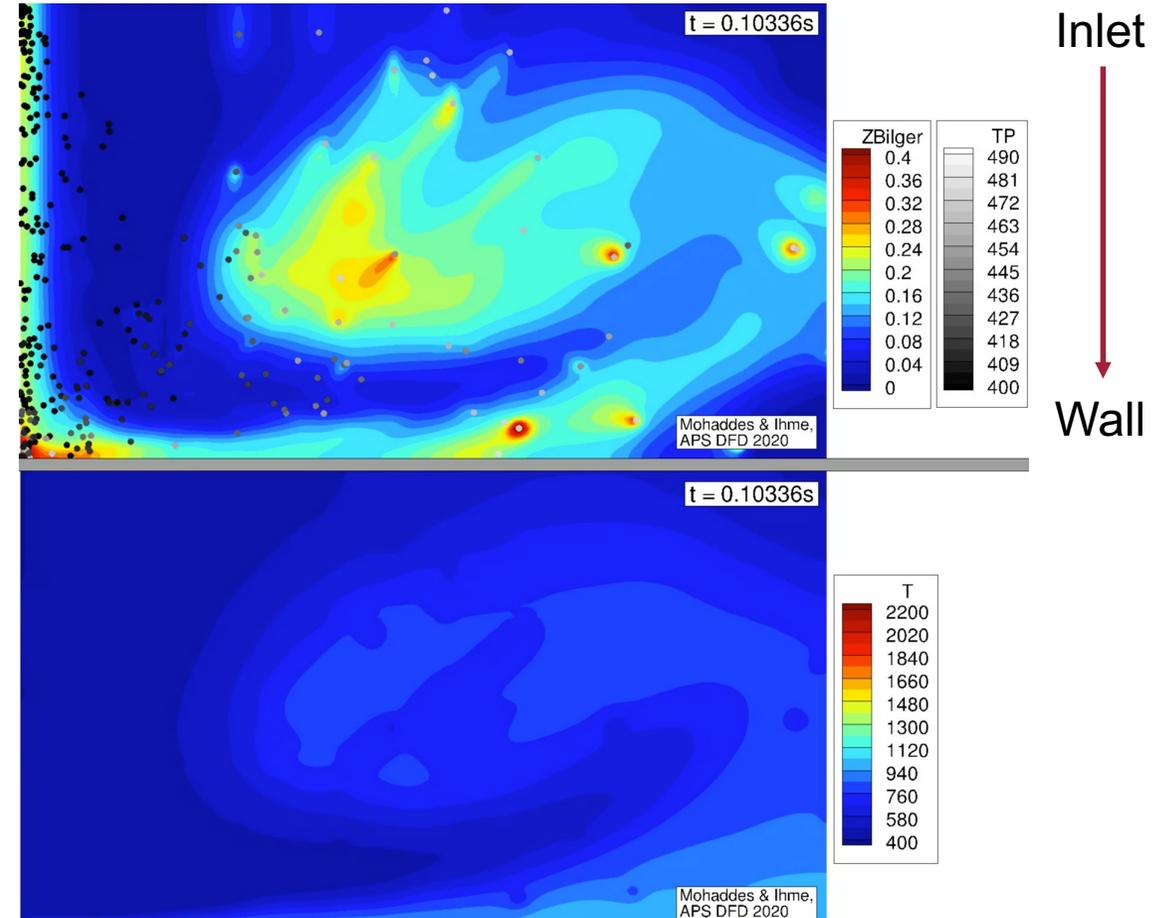
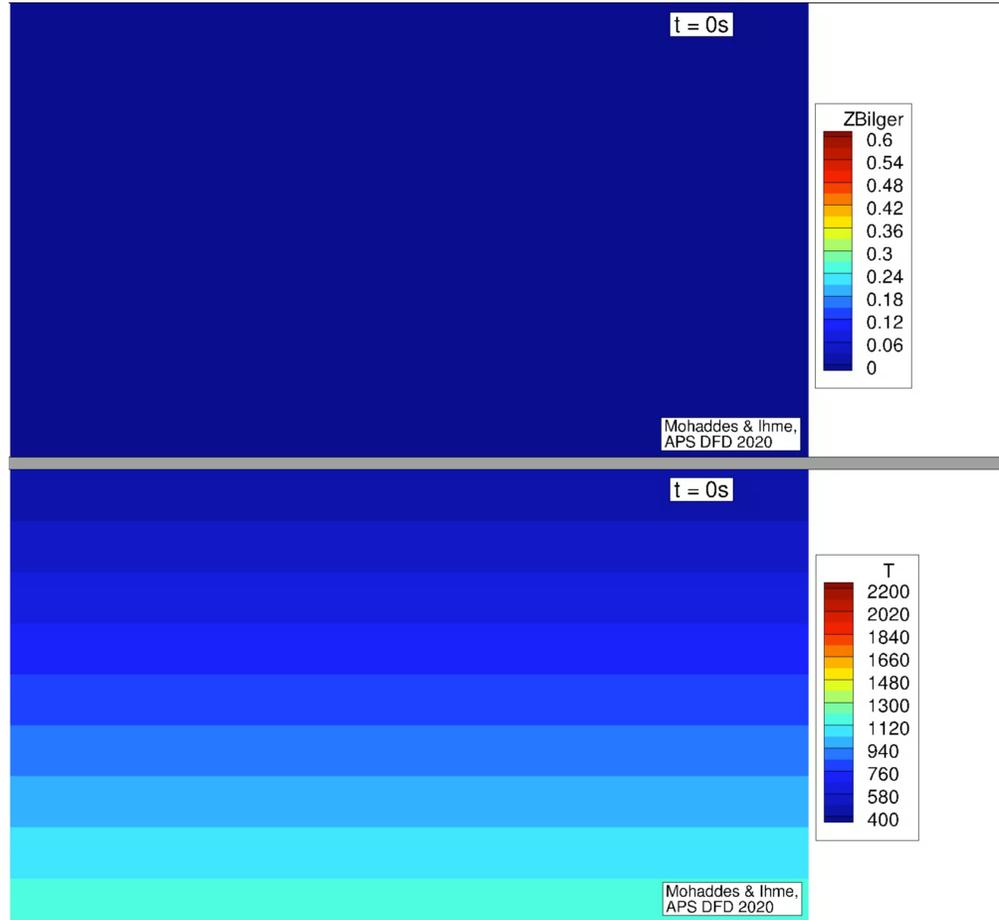
- Fuel film forms and evaporates rapidly
- Interaction of injected droplets with fuel vapor forms impinging jet-like core with rolling vortices
- Ignition occurs at edge of fuel vapor core at 34 ms

3D Simulations | Flow-field Temperature

$$T_w = 1200\text{K}$$

$$T_w = 1000\text{K}$$

Inlet
↓
Wall



Inlet
↓
Wall

- Higher wall temperature results in shorter ignition delay
- Less fuel vapor available for combustion at high wall temperature → rapid transition to compact flame

1D Simulation | Set-up, Methodology

Objective

- Develop low-order model to enable comprehensive parametric investigation of hot-surface ignition

Method

- Dimensionality reduction (1D model)
- Eulerian spray representation
- Coupled phases

Non-Dimensional Parameters

Parameter	Symbol
Normalized Solid External Temperature	T_e
Fuel/Air Equivalence Ratio	ϕ_0
Stokes Number (Droplet response)	St
Damköhler Number (Flow vs. Chemical time scale)	Da

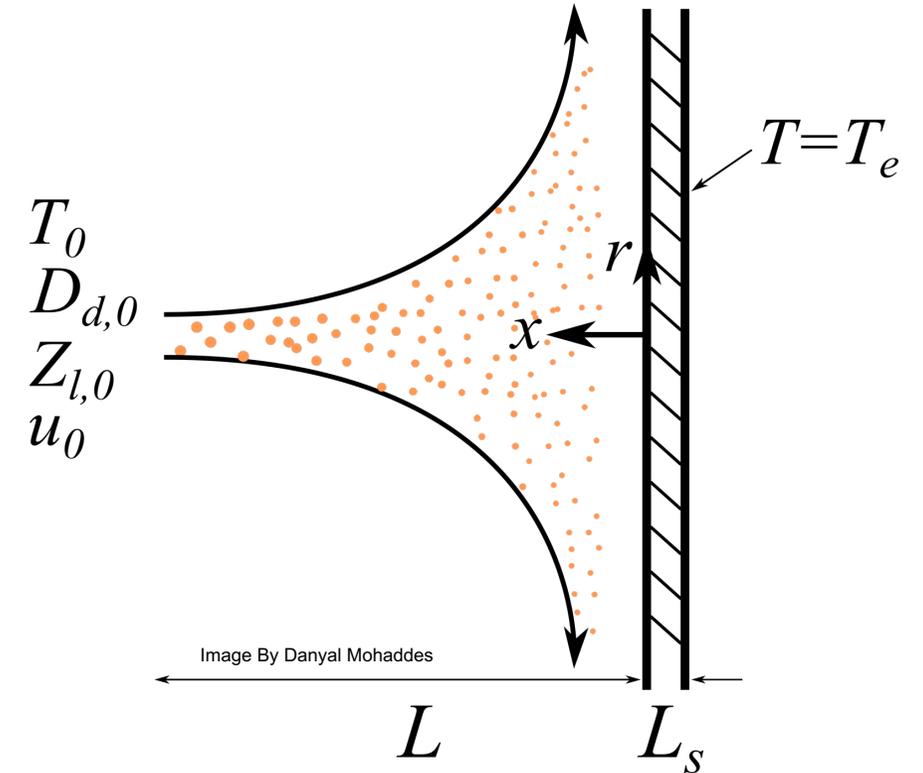
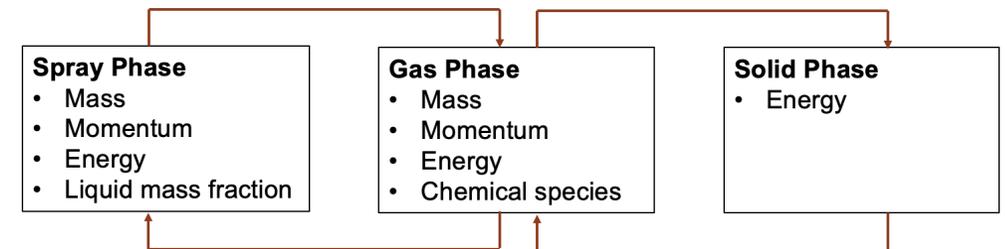
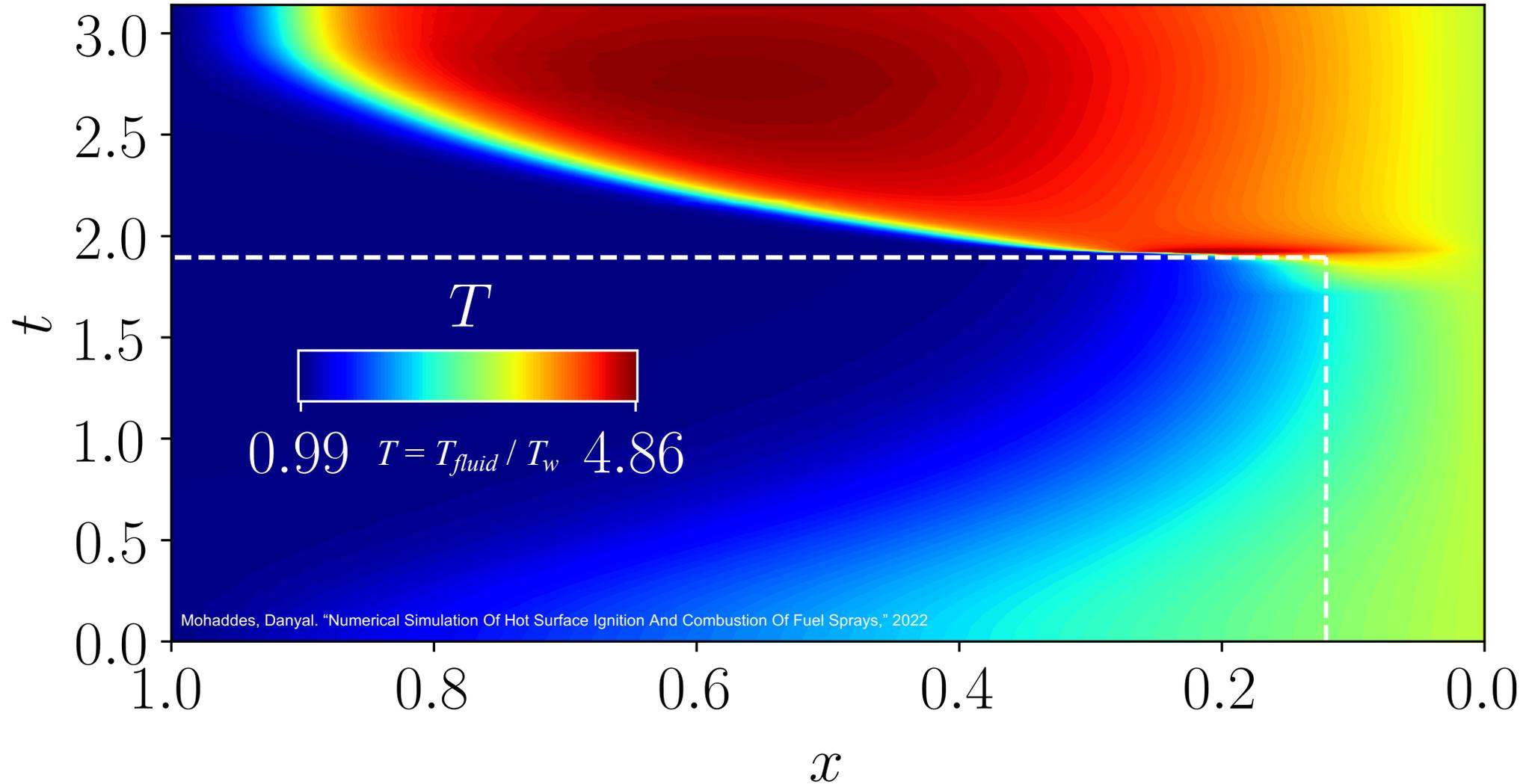
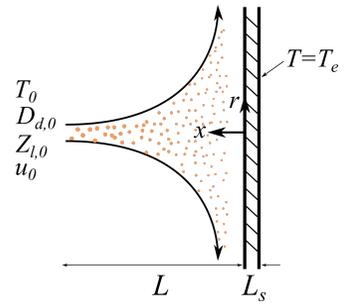


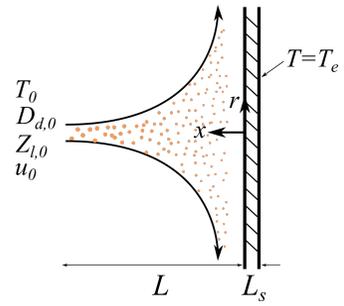
Image By Danyal Mohaddes



1D Simulation | Ignition Phenomenology

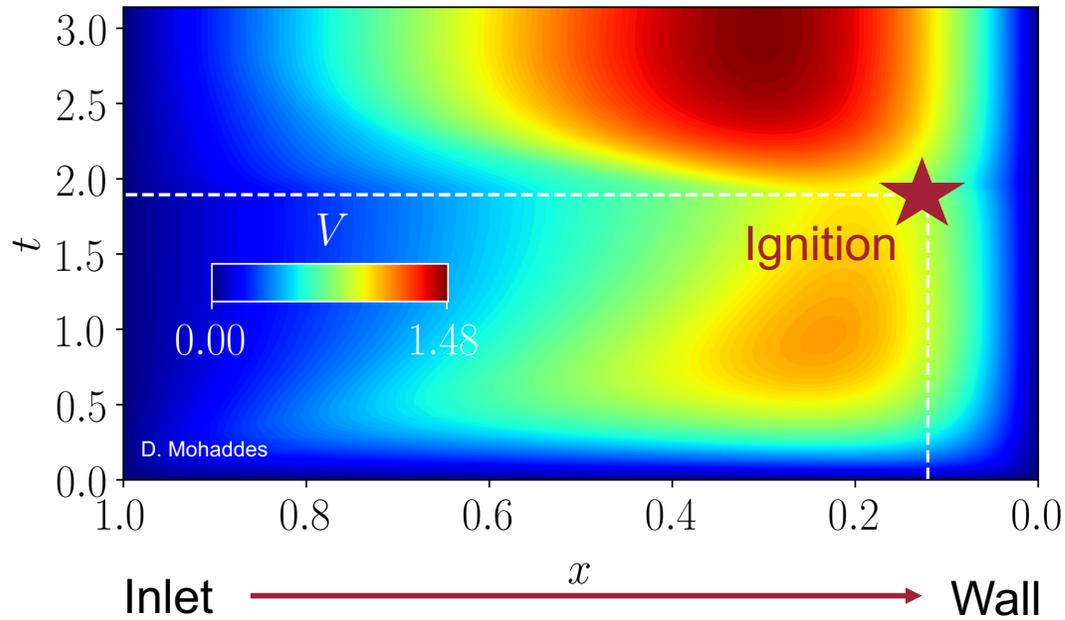


1D Simulation | Ignition Phenomenology



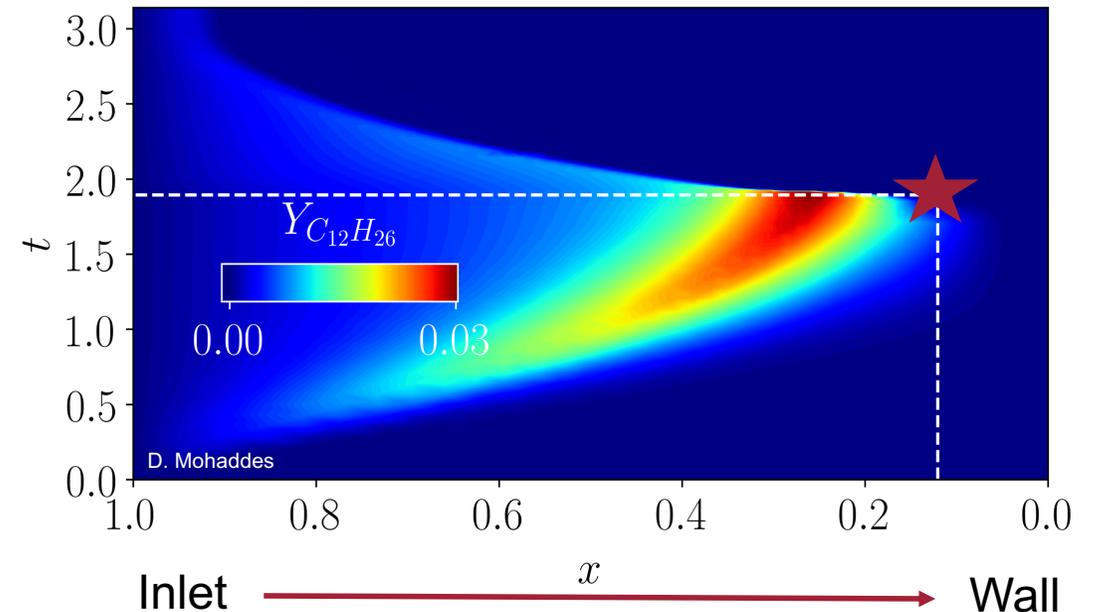
Fluid Dynamics

$$V \equiv \frac{v}{r}$$

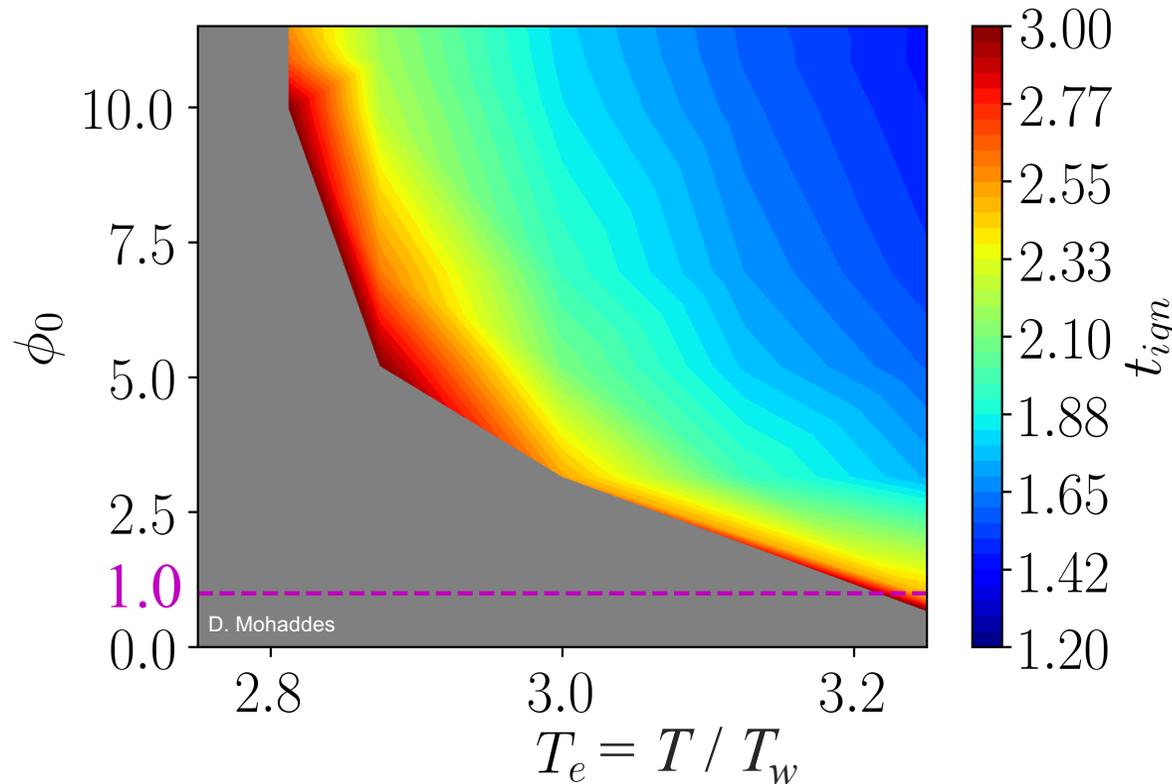


Thermochemistry

$$\phi_0 = 1.0$$



1D Simulation | Parametric Study



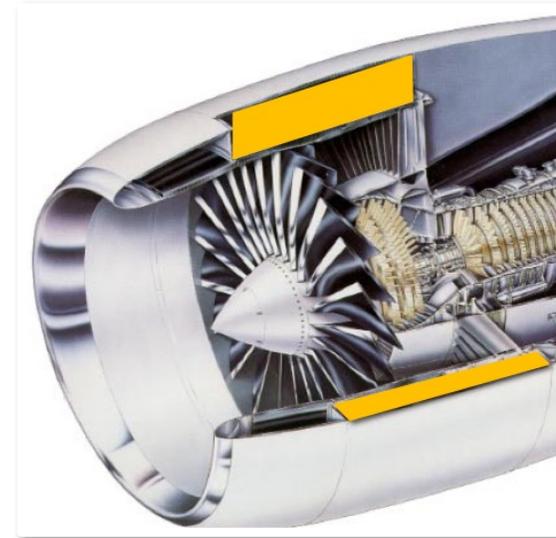
Ignition Limits:

- Da largely determines igniting region
 - Lean and near-stoich ϕ_0 non-igniting
- Large gradient in T_e

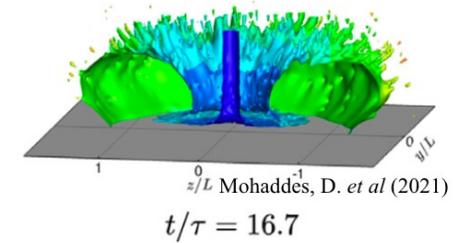
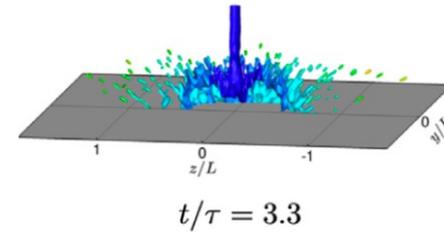
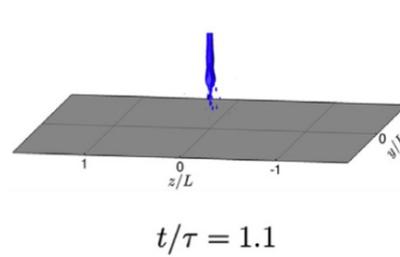
Ignition Delay Time:

- Reaches maximum at ignition limit
- Minimum at high T_e, ϕ_0
- Modest variation relative to purely chemical ignition delay

Conclusion



Key Findings



Developed **predictive modeling tools for exploration and analysis of hot surface ignition** scenarios at quantitative level.

3D Simulations

- Higher surface temperature reduces ignition delay, resulting in ignition occurring before gaseous flow field is fully developed
 - Higher wall temperature = lower maximum wall heat flux

1D Simulations

- Spray interaction with thermal boundary layer causes fuel deposition in vapor phase
- Ignition phenomenology
 - Flame stabilization near surface → enhanced wall-heat flux
 - Injector-stabilized flame → reduced heat-flux and early droplet combustion
- Parametric study
 - Ignition limit at richer ϕ_0 and higher T_e , depends on Da
 - Formation of ignition kernels in premixed region near the wall

Future Work | Research Issue

- **Research Issue:** Lack of high-quality experimental thermo-fluid data to validate model
- **Research Objective:** Develop and perform experiments to experimental study hot-surface ignition phenomena using advanced diagnostics and target quantities
 - High speed imagery
 - Droplet properties
 - Combustion properties
 - Heat transfer at wall

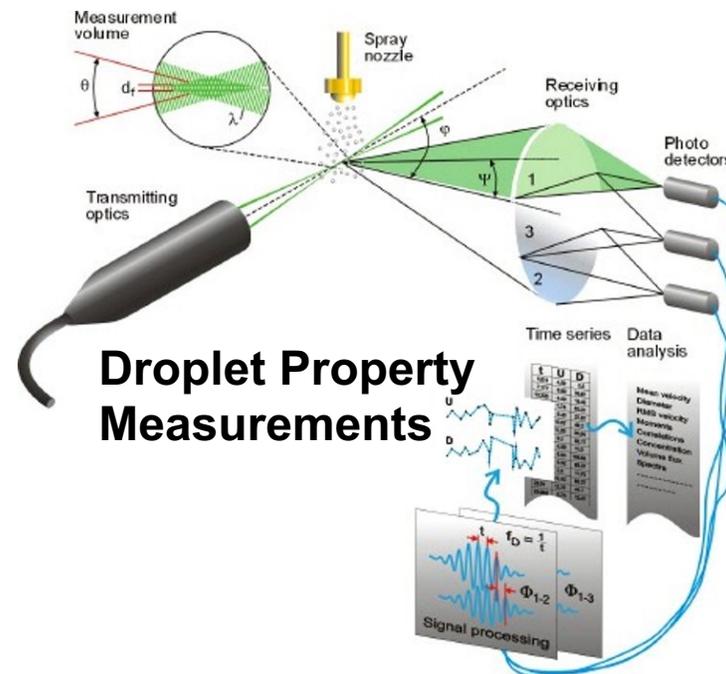
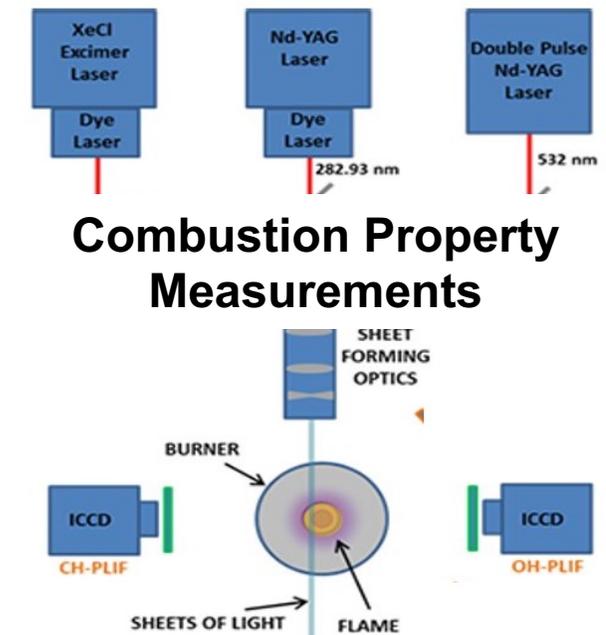


Image From Dantec Dynamics

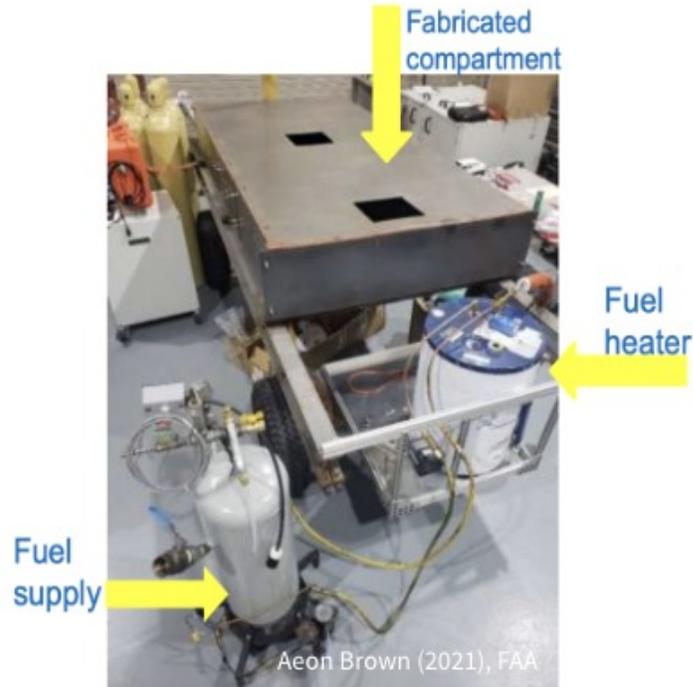
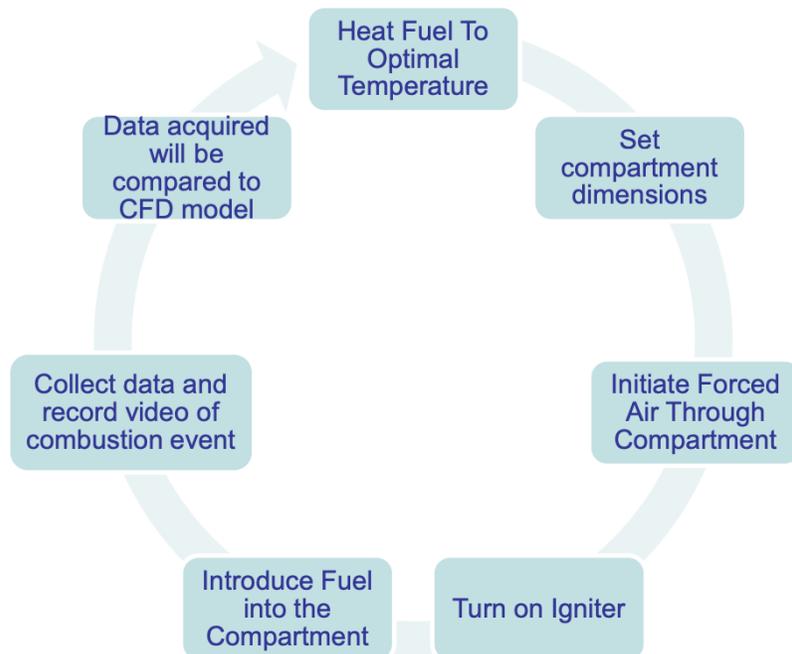


Seitzman, J.M. and Hanson R.K. (1993), ISBN 0-12-683920-4.

Future Work | FAA Inspiration (Aeon Brown)

Goals

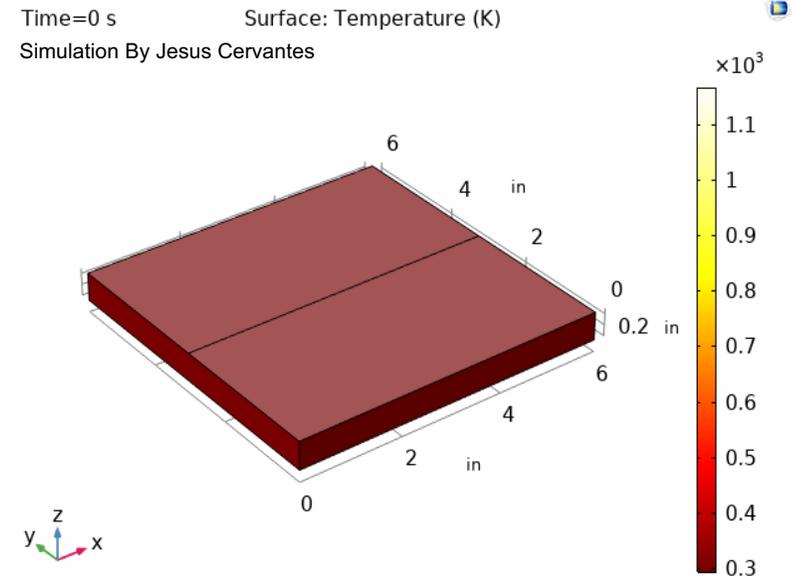
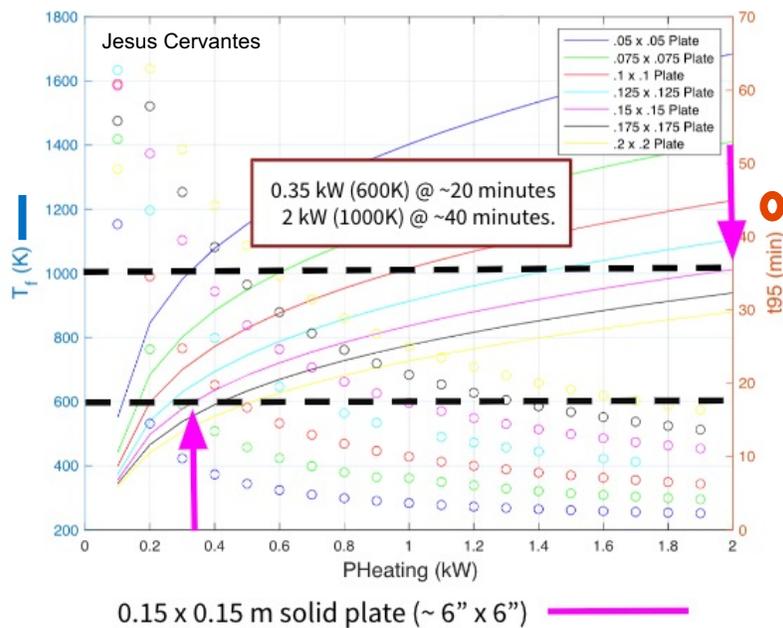
1. Study effect of jet fuel delivery, air flow delivery, and compartment dimensions on combustion
2. Use to validate fire CFD modeling



- **Aspects to Emulate:** Full-size compartment, modular dimensions, forced convection, use for CFD validation
- **Differences in Scope:** Fuel ignited via spark plug (cannot study hot surface ignition), minimal diagnostics

Future Work | Experimental Vision

- **Complement FAA set-up and work, with a focus on:**
 - Incorporate key elements in real aircraft, like cylindrical geometry and cross-flow.
 - Employ a suite of high fidelity diagnostics matching the physical and temporal resolution of the phenomena.
 - Establish a flexible and modular experimental platform that enables parametric dependency exploration
 - Different spray-angle, geometries, stream properties, etc.



Up First: Hot Surface Design!

Acknowledgements



Ihme Group @ Stanford University

Specifically Danyal Mohaddes, Jesus Cervantes,
Dr. Guillaume Vignat, and Prof. Matthias Ihme



FAA



Boeing Company
under Grant No. 134708
[IC2017-2182]



U.S. Department of Energy Office of Science

National Energy Research Scientific Computing Center
under Contract No. DE-AC02-05CH11231.

Thank You!

QUESTIONS?

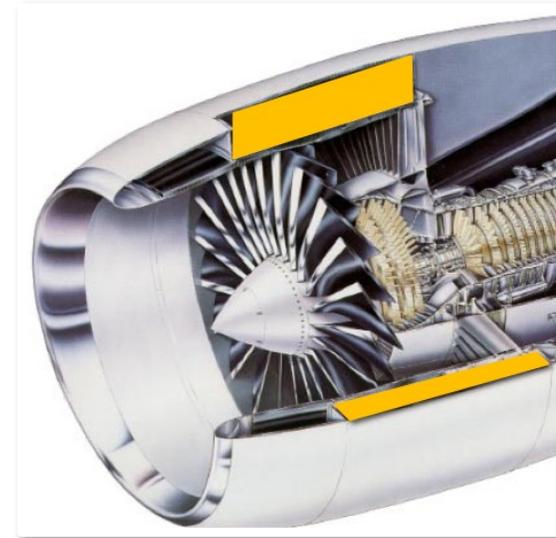
Matthias Ihme

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Back-up

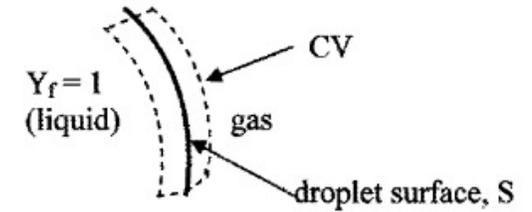


Time Scale Analysis | 0D Droplet Model

Inputs: T_{liq} , T_{gas} , Droplet Diameter, Liquid Properties CSV

Antoine Parameters (Vapor $P = f(T)$)

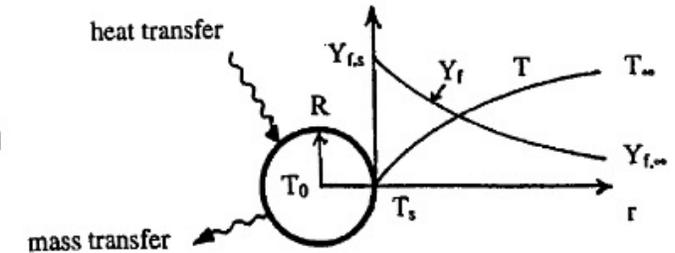
name	rho	cp	Lv	prs_A	prs_B	prs_C	prs_D	Tb	unit
C6H14	653.17	2280.3	3.65E+05	4.00266	1171.53	-48.784		342.1	bar
NC12H26	669.43	2593.9	2.56E+05	4.10549	1625.928	-92.839		489.3	bar
N-C12H26	669.43	2593.9	2.56E+05	4.10549	1625.928	-92.839		489.3	bar



Bowman, C.T. ME 372 Course Reader.

Operating Principles:

- 0D Droplet Evaporation Model
- Heat conduction from surroundings supplies energy for evaporation
 - Evaporation Rate = $f(\text{fuel properties, environment properties})$
- Vaporized mass diffuses away from surface



Bowman, C.T. ME 372 Course Reader.

From Mass Conservation

$$\dot{m}''(4\pi r^2) = \dot{m}_s'(4\pi R^2) = \text{constant}$$

mass flux/area @ r mass evaporation rate/droplet surface area (@ $r = R$)

$$B = B_T = -\eta_{T,s} = \frac{c_{p,g}(T_\infty - T_s)}{Q}$$

$$B = B_f = -\eta_{f,s} = \frac{(Y_{f,s} - Y_{f,\infty})}{(1 - Y_{f,s})}$$

Common to rewrite equations in terms of transport number B (driving force for heat/mass transfer)

From Energy Conservation

$$\dot{m}_s' [h_{fg} + c_{l,s}(T_s - T_0)] = \lambda_g \left(\frac{dT}{dr} \right)_{s,g} = \dot{q}_s'$$

h_{fg} = latent heat of vaporization of the liquid at pressure, P
 $c_{l,s}$ = specific heat of the liquid
 T_s = droplet surface temperature = constant
 T_0 = initial liquid temperature
 Q = total energy required to increase the droplet temperature from $T_0 \rightarrow T_s$ and to evaporate the liquid.

From Species Conservation

$$\dot{m}_s' = \dot{m}_s' Y_{f,s} - \rho_g D_g \left(\frac{dY_f}{dr} \right)_{s,g}$$

Sum of gas-phase convection (Stefan flow) and gas-phase Fickian diffusion
 Y = gas phase mass fraction
 D_g = diffusion coefficient through gas

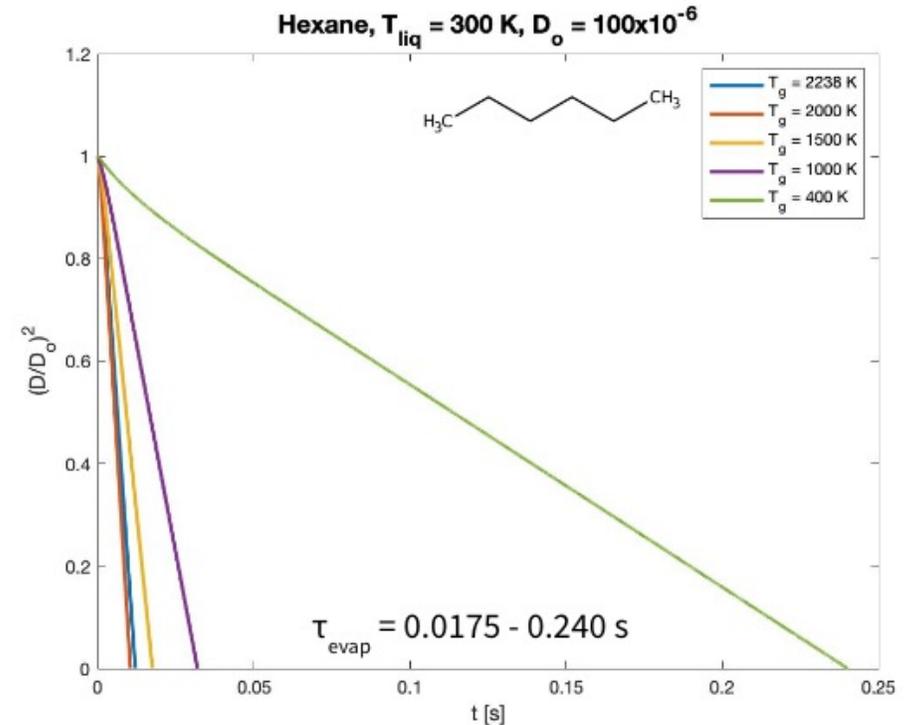
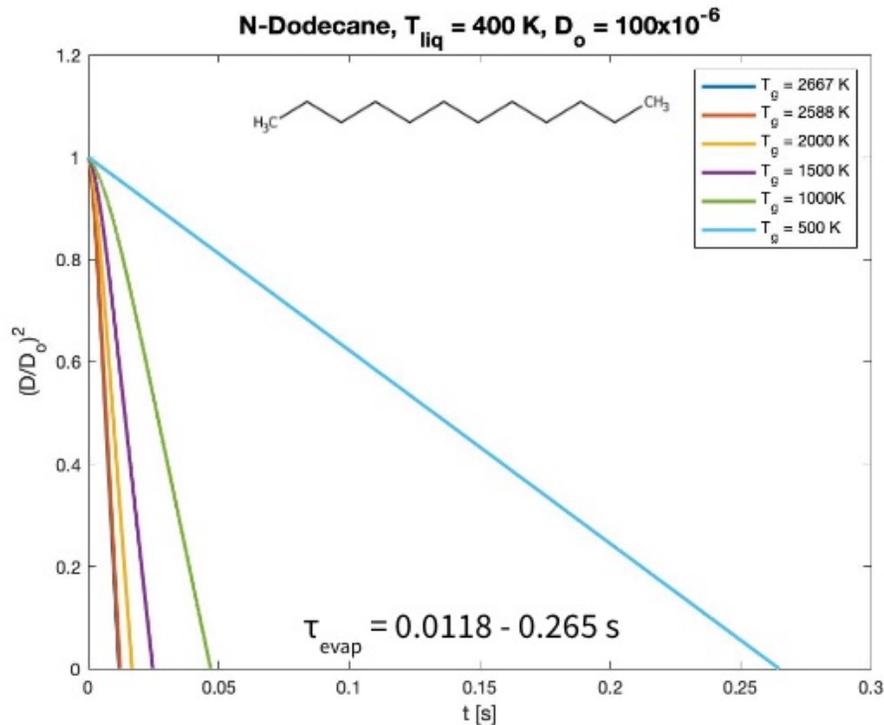
Primary Output: Droplet Diameter as $f(\text{Time})$

Time Scale Analysis | Evaporation as f(Temp)

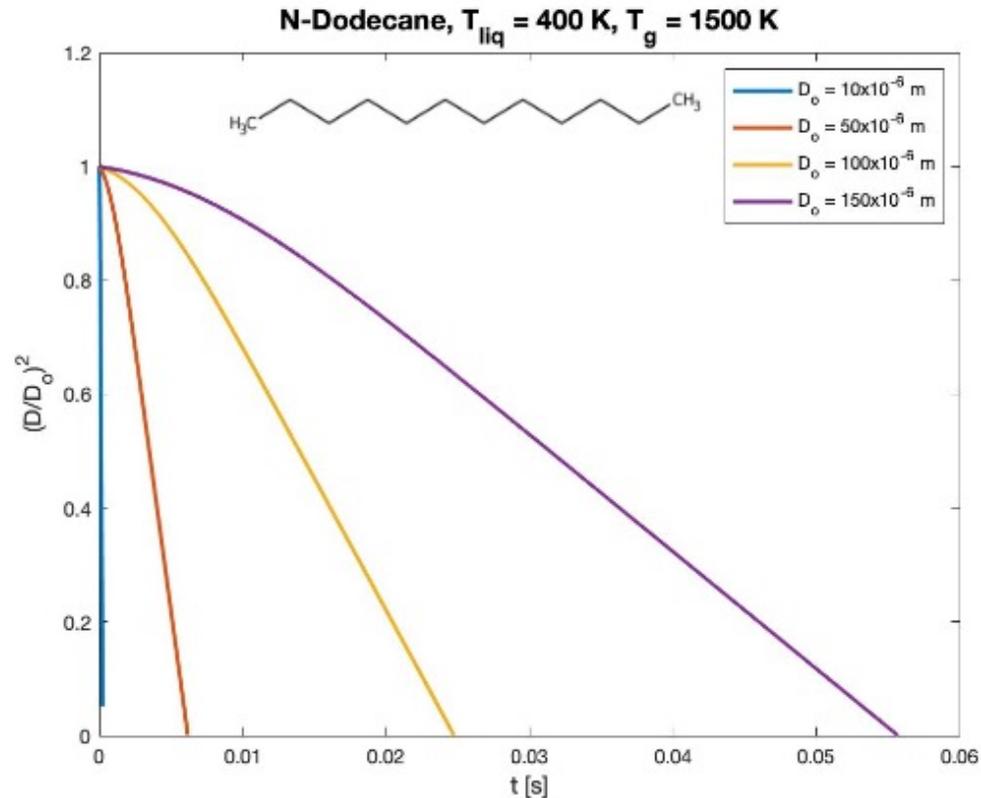
$$\frac{d(d^2)}{dt} = - \left(\frac{8\rho_g D_g}{\rho_L} \right) \ln(1+B) \equiv -k_{\text{evap}} \quad (\text{the evaporation constant})$$

Assuming $k_{\text{evap}} = \text{constant} \Rightarrow$

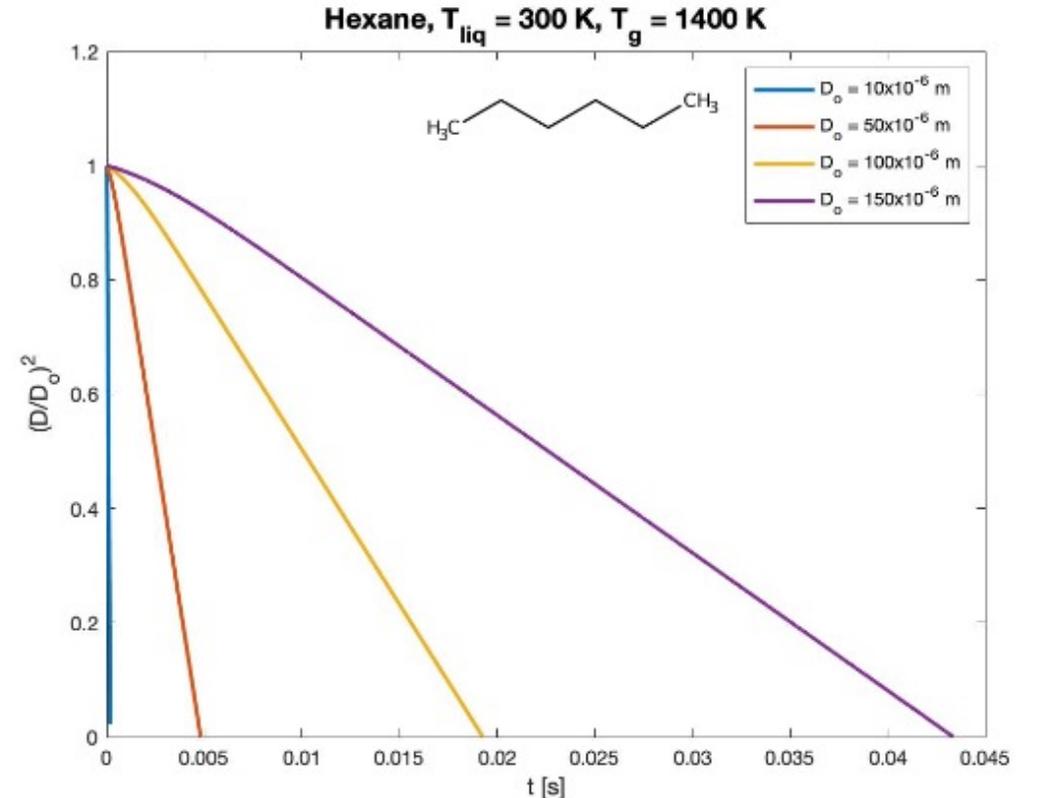
$$d^2 = d_0^2 - k_{\text{evap}} t \Rightarrow t_{\text{evap}} = d_0^2 / k_{\text{evap}}$$



Time Scale Analysis | Evaporation as $f(\text{Droplet } D)$

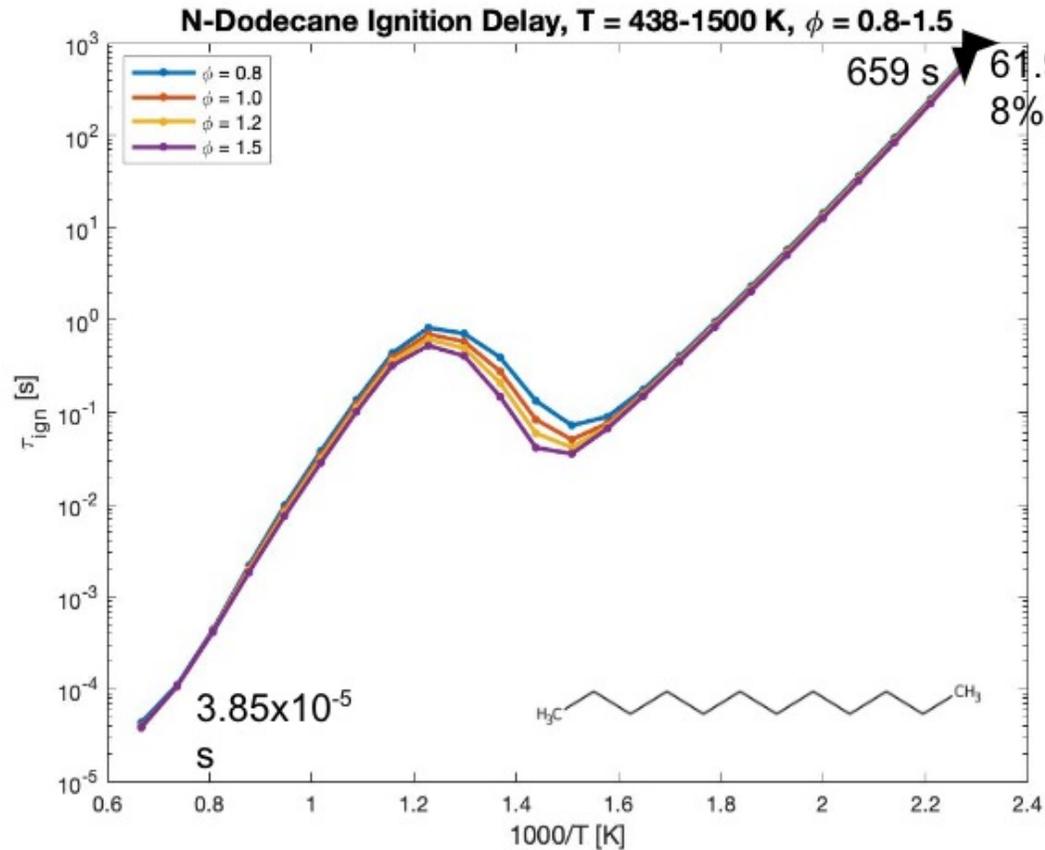


$$\tau_{\text{evap}} = 0.00024 - 0.056 \text{ s}$$

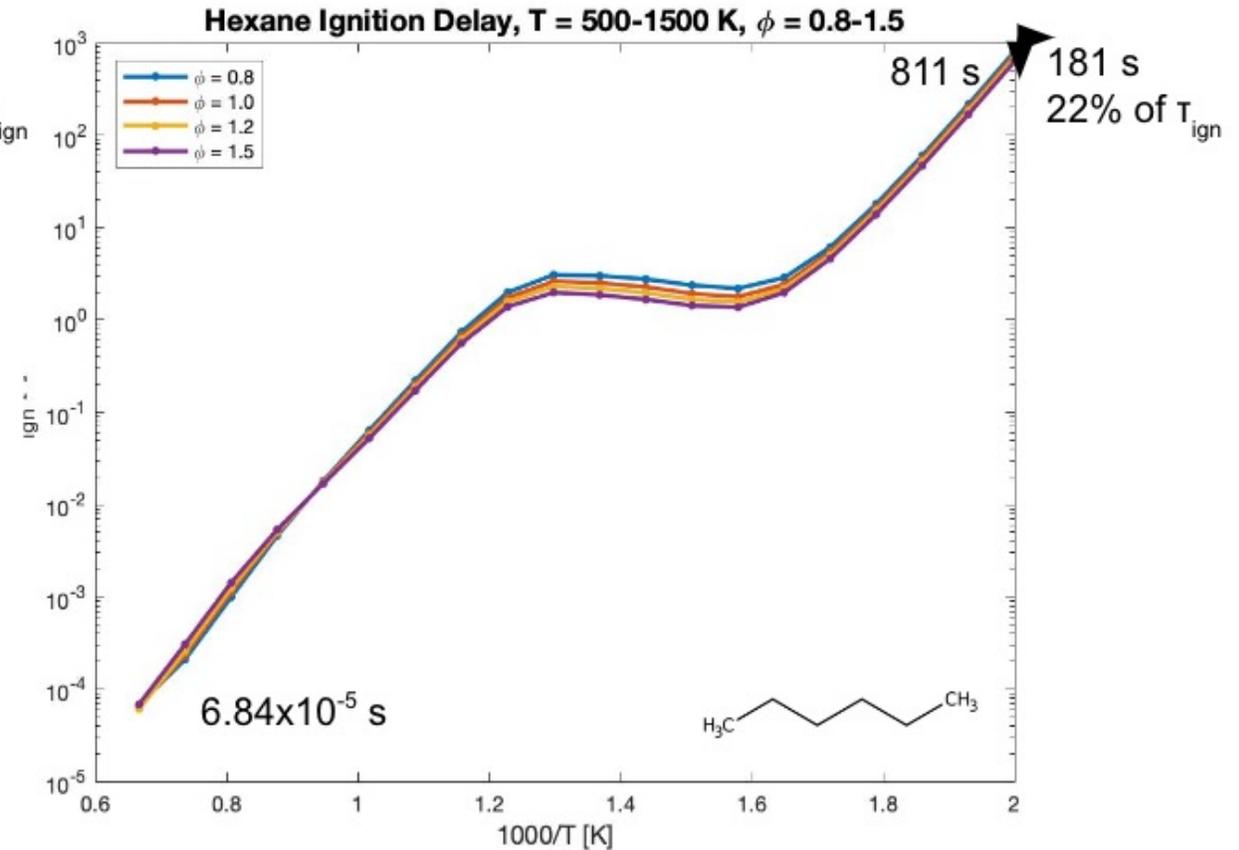


$$\tau_{\text{evap}} = 0.00015 - 0.043 \text{ s}$$

Time Scale Analysis | Fuel Ignition



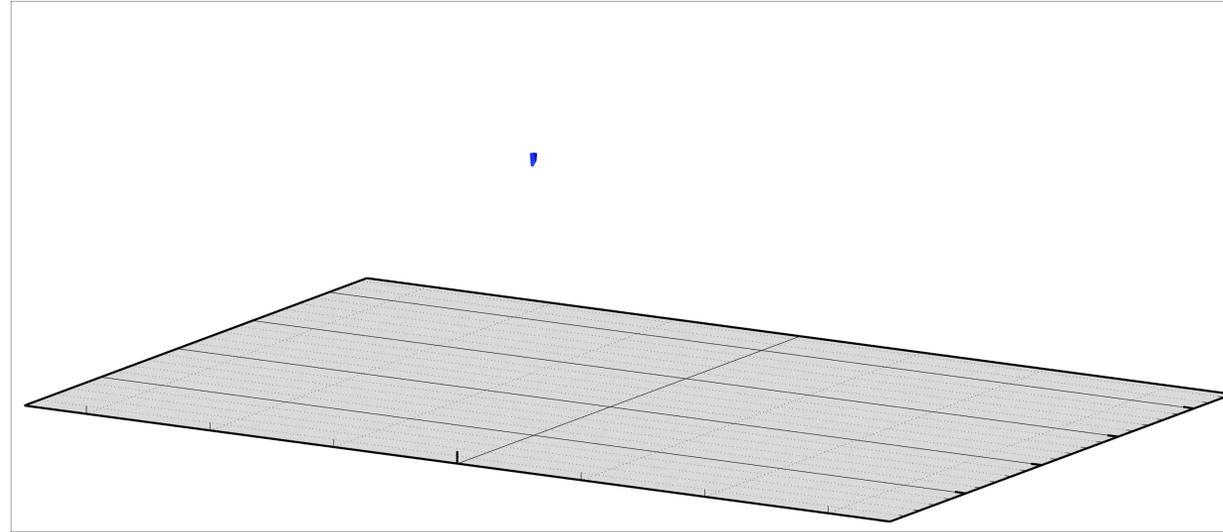
From Cantera, SK54 compact skeletal mechanism with optimized low-temperature chemistry (Yao *et al.*)



From Cantera, Hexane (C₆H₁₄)-air full mechanism optimized for thermal ignition experiments (Mével *et al.*)

Results: Flow structure

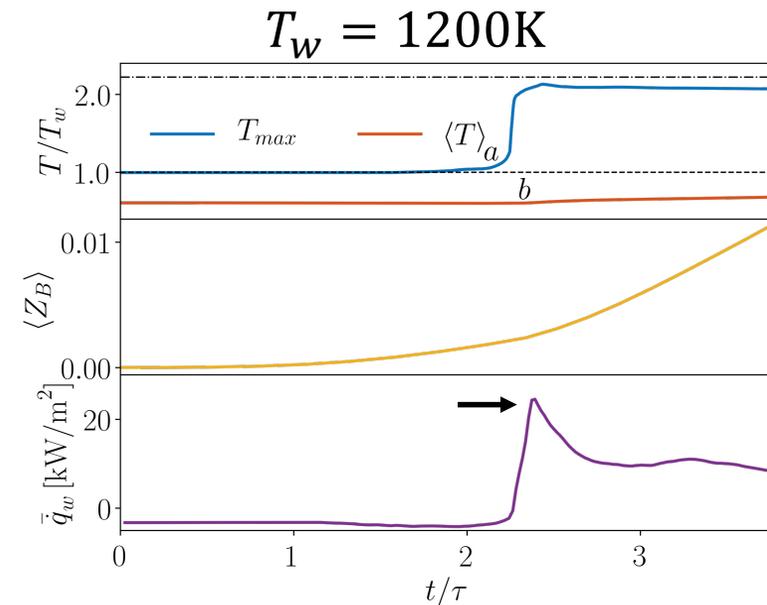
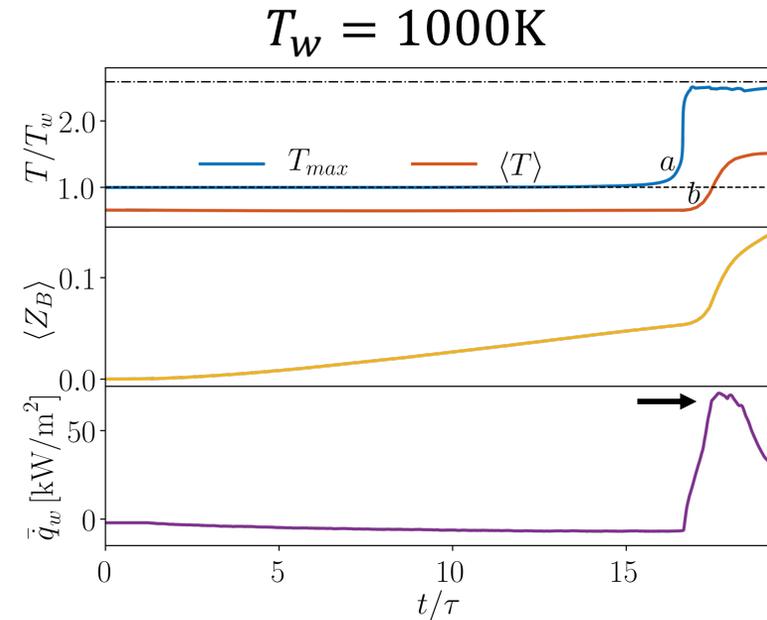
$T_w = 1000\text{K}$, Z_{st} iso-surface, half-plane



- Liquid injected stochastically in a conical spray, impinges on surface
 - Inelastic reflection due to Leidenfrost effect
- Spray drives gas-phase secondary flow due to momentum exchange
 - Droplet drag
- Spray evaporates due to interaction with hot air, mixes
 - Forms toroidal vortex, identifiable by stoichiometric (Z_{st}) iso-surface

Analysis: Volume-averaged

- Consider results averaged on entire simulated domain
 - Mass-weighted volume averaging
- In both cases, lag time between ignition (a) and increase in compartment mean temperature (b)
 - Kernel development and flame propagation
 - Mixture fraction increases more rapidly due to increased evaporation
- Increased wall temperature results in reduced ignition delay
 - Shorter evaporation time, so lower $\langle Z_B \rangle$
 - Less fuel in compartment, less mixing results in $\sim 3x$ lower wall heat flux



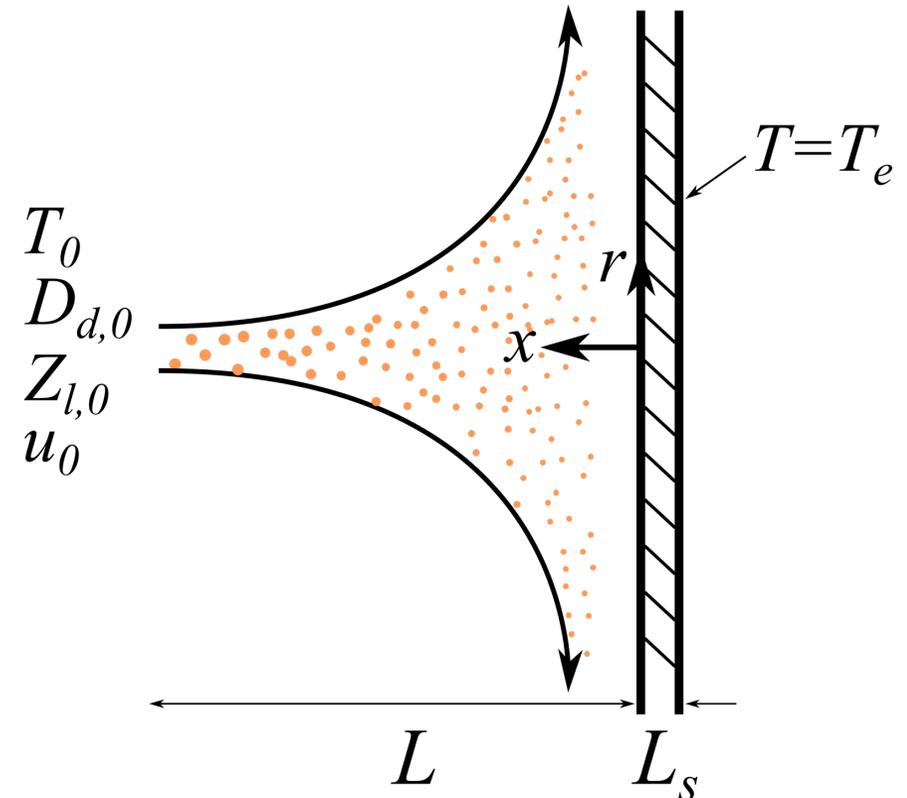
Problem definition and parametrization

Fixed parameters

- Fuel: liquid *n*-dodecane at $T_0^* = 400\text{K}$
- Oxidizer: air at $p^* = 1\text{atm}$, linear temperature between T_0^* and T_w^*
- Fluid gap size: $L = 2\text{cm}$
- Wall: solid steel, linear temperature between T_w^* and T_e^*
- Wall thickness: $L_s = 3\text{mm}$

Open parameters

- External wall temperature: $T_e^* = [1100\text{K}, 1300\text{K}]$
- Inlet liquid mass fraction: $Z_{l,0} = [0.03, 0.77]$
- Inlet droplet diameter: $D_{d,0}^* = [8\mu\text{m}, 346\mu\text{m}]$
- Global strain rate: $a^* = [1\text{s}^{-1}, 100\text{s}^{-1}]$



Governing equations

From 3D to 1D

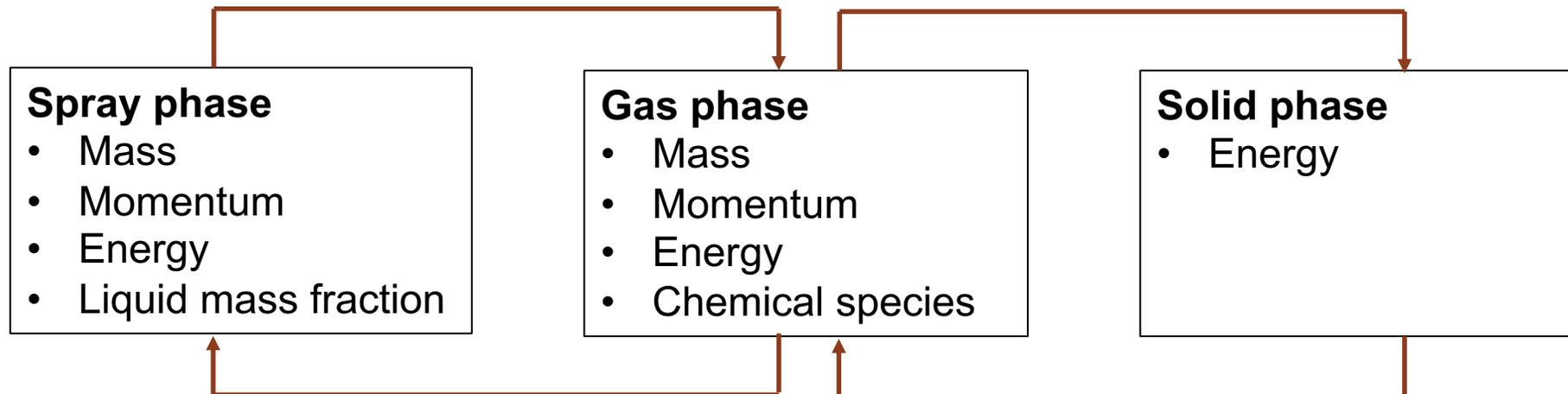
- Perpendicular impingement, thus assume axi-symmetry (3 → 2)
- Consider solution only on centerline $r = 0$ (2 → 1)

From Lagrangian to Eulerian

- Assume that on a length scale larger than the droplets, but smaller than the problem, droplets behave identically
- $[\mathbf{x}, \mathbf{u}, T_d, m_d]_i \forall i < N_d \rightarrow [Z_l, \mathbf{u}, T_d, m_d](\mathbf{x})$

Three coupled phases

- Spray-gas and gas-spray exchange conserved quantities through source terms
- Gas-solid and solid-gas exchange through boundary conditions



Non-dimensional parameters

From non-dimensionalization of governing equations and boundary conditions, we obtain:

Parameter	Symbol	Definition	Range Considered
Normalized solid ext. temp.	T_e	$T_e = \frac{T_e^*}{T_0^*}$	[2.75, 3.25]
Total equivalence ratio	ϕ_0	$\phi_0 = \frac{Z_{l,0}}{f_{st}}$	[0.5, 11.5]
Stokes number	St	$St = \frac{\rho_l^* D_{d,0}^{*2}}{18\mu_0^*} a^*$	[0.001, 0.3]
Damköhler number	Da	$Da = \frac{\dot{\omega}_c^*}{\rho_0^* a^*}$	$[10^1, 10^3]$

(*) indicates dimensional value

Results: Ignition phenomenology

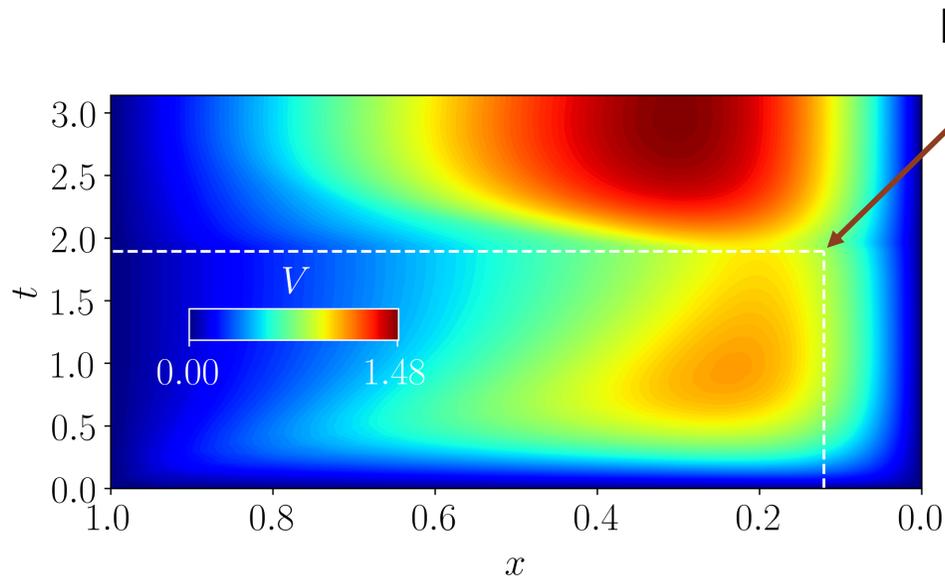
$$T_e = 3.25,$$

$$\phi_0 = 1.0,$$

$$St = 0.1$$

$$t = \frac{t^*}{t_0^*}, \quad t_0^* \equiv \frac{1}{a^*}$$

Fluid dynamics $V \equiv \frac{v}{r}$

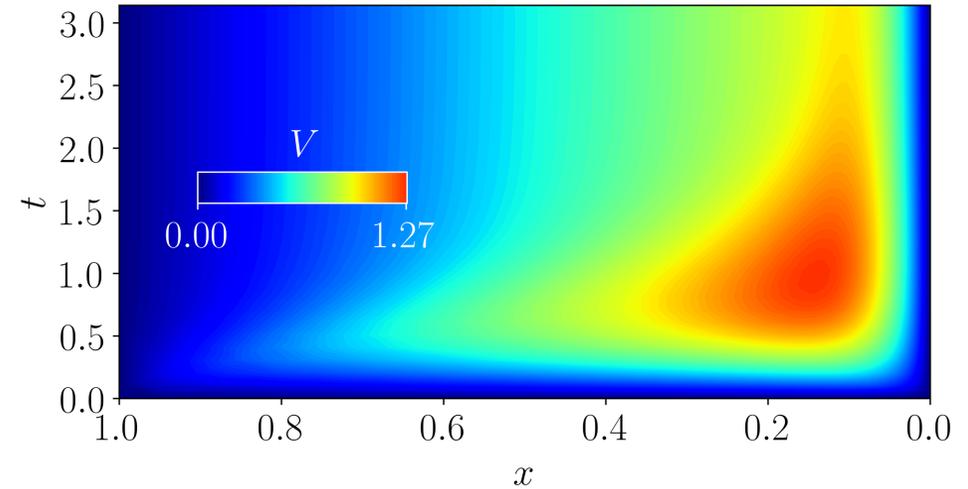


Inlet



Wall

Igniting (Da=120)



Non-igniting (Da = 30)

Results: Ignition phenomenology

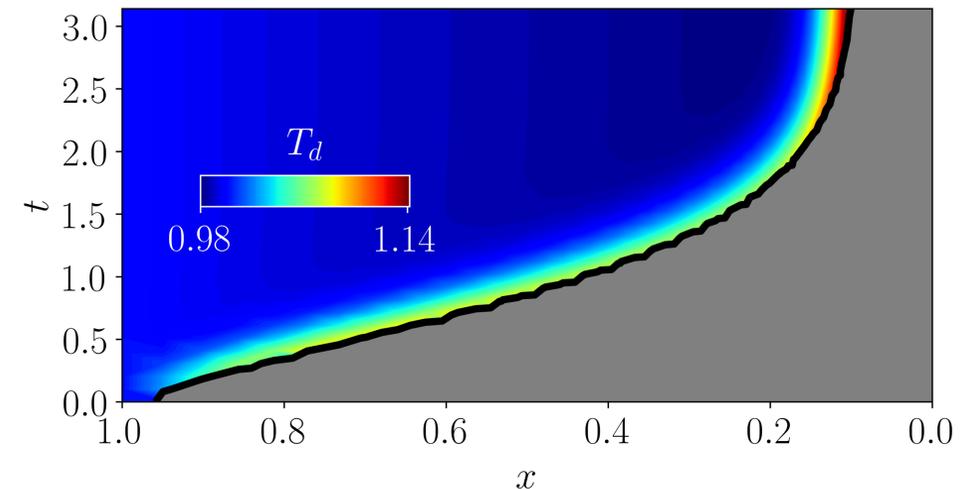
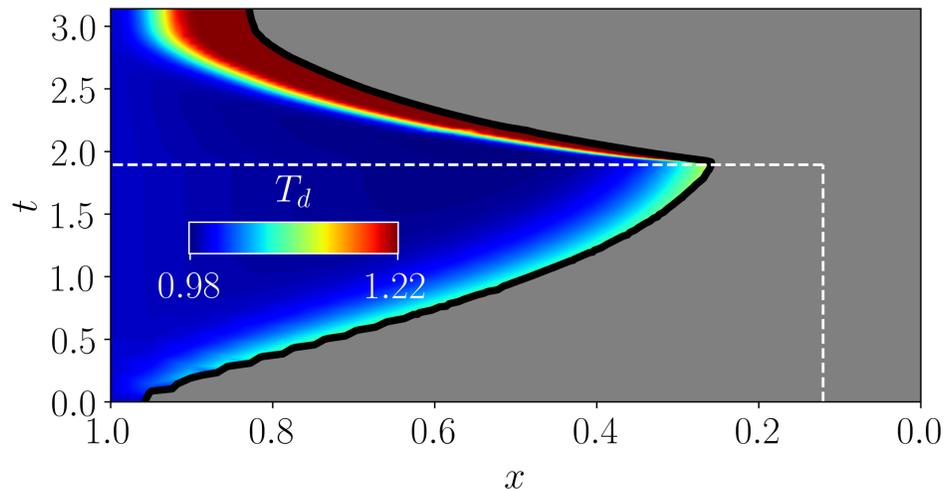
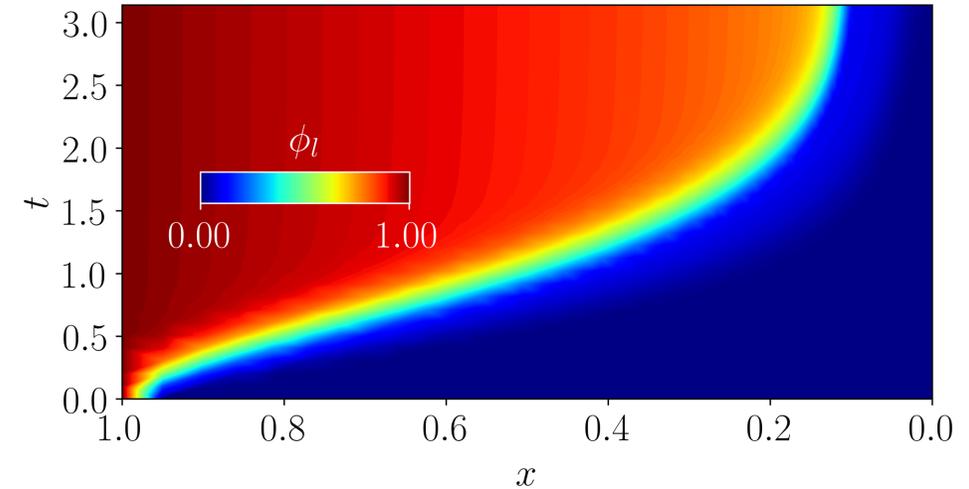
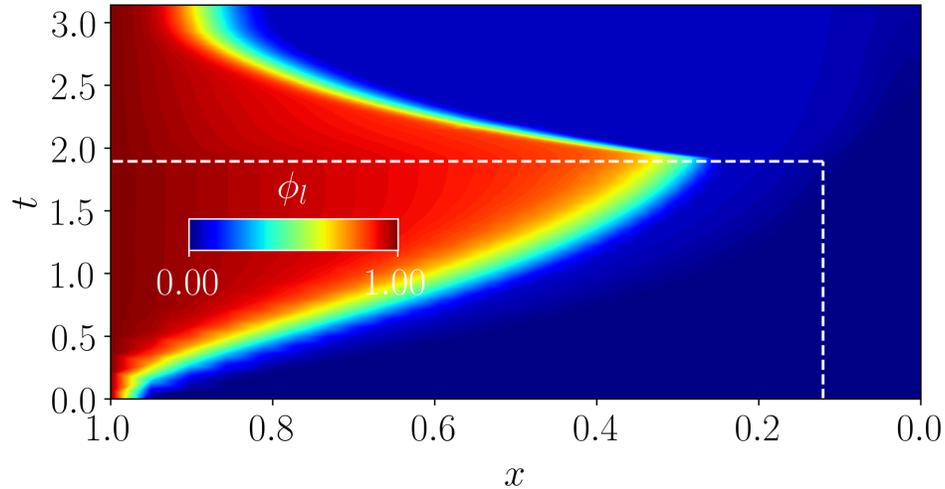
$$T_e = 3.25,$$

$$\phi_0 = 1.0,$$

$$St = 0.1$$

$$t = \frac{t^*}{t_0^*}, \quad t_0^* \equiv \frac{1}{a^*}$$

Spray dynamics $\phi_l \equiv Z_l/f_{st}$



Igniting (Da=120)

Non-igniting (Da = 30)

Results: Ignition phenomenology

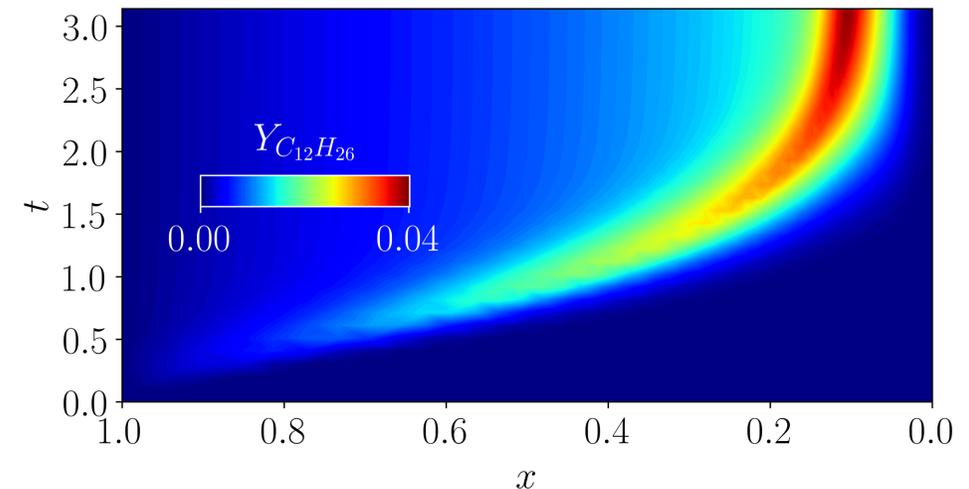
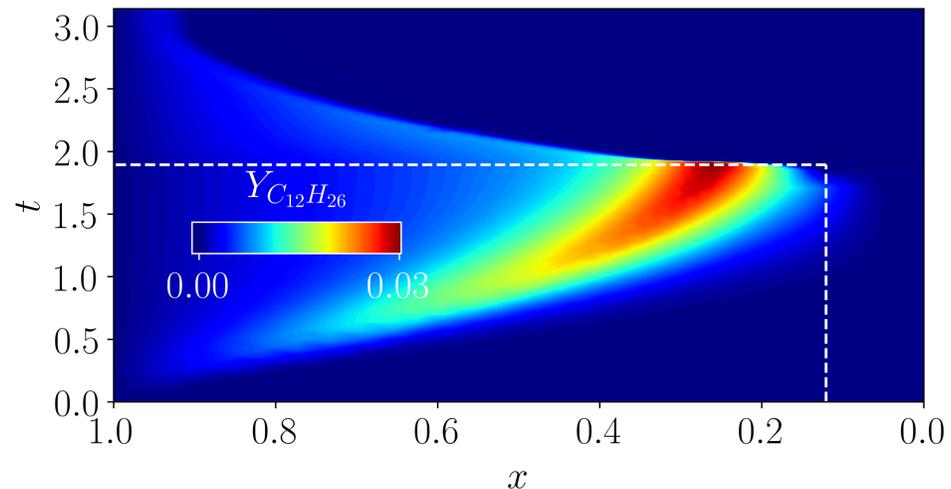
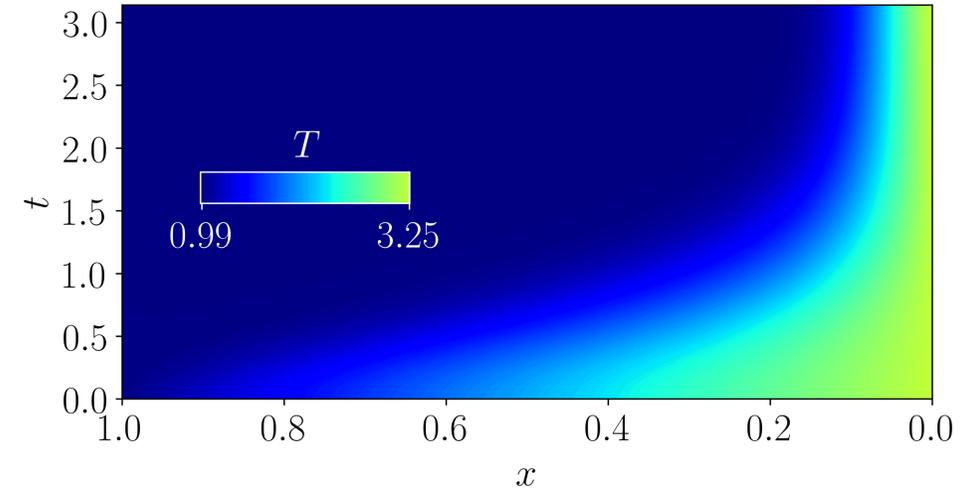
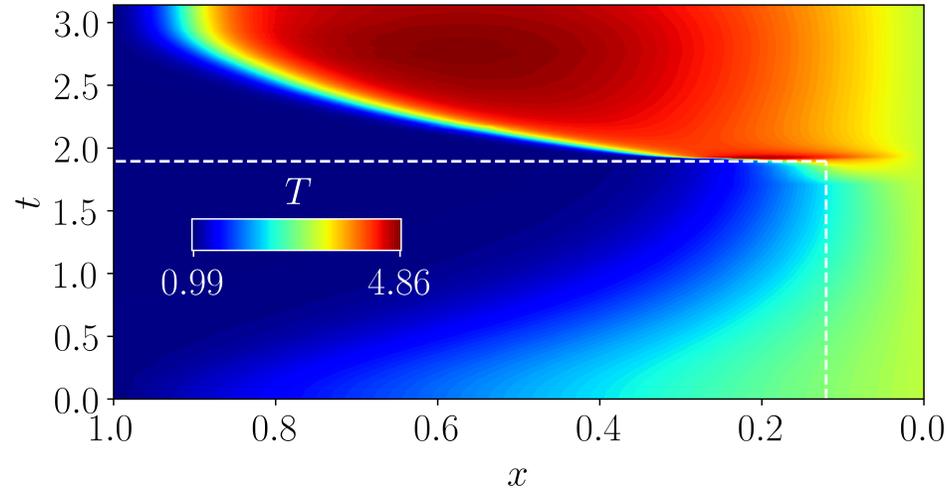
$$T_e = 3.25,$$

$$\phi_0 = 1.0,$$

$$St = 0.1$$

$$t = \frac{t^*}{t_0^*}, \quad t_0^* \equiv \frac{1}{a^*}$$

Thermochemistry



Igniting ($Da=120$)

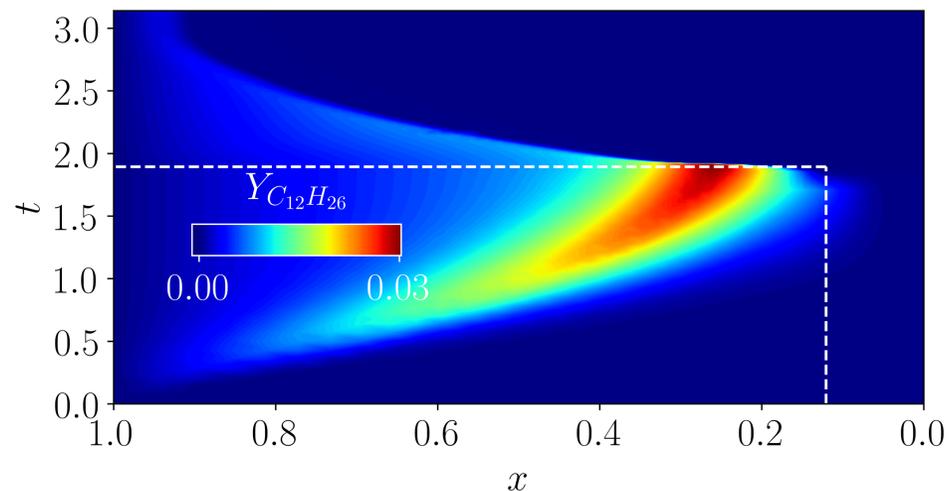
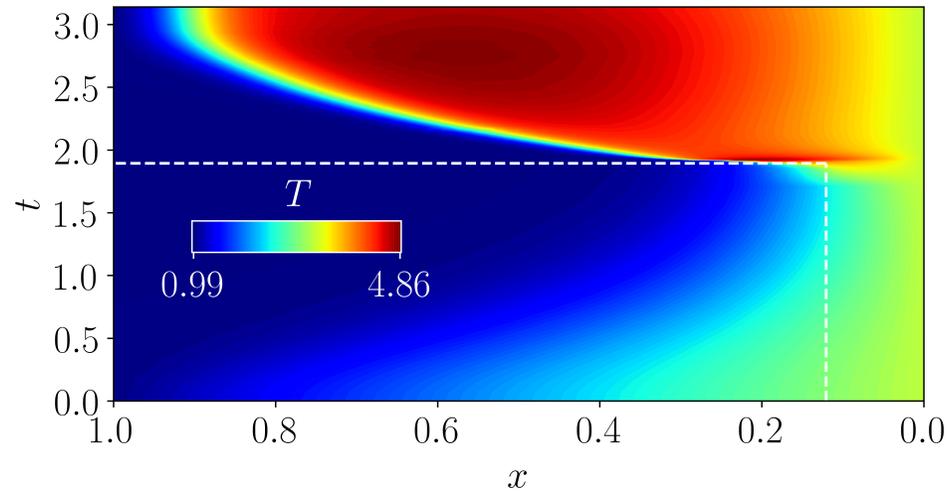
Non-igniting ($Da = 30$)

Results: Ignition phenomenology

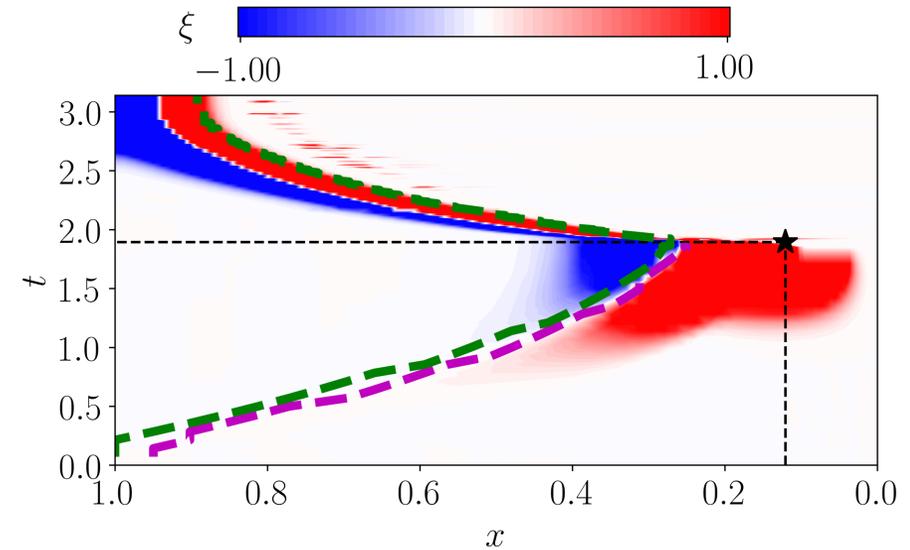
$$T_e = 3.25, \quad \phi_0 = 1.0, \quad St = 0.1$$

$$t = \frac{t^*}{t_0^*}, \quad t_0^* \equiv \frac{1}{a^*}$$

Thermochemistry



Igniting (Da=120)



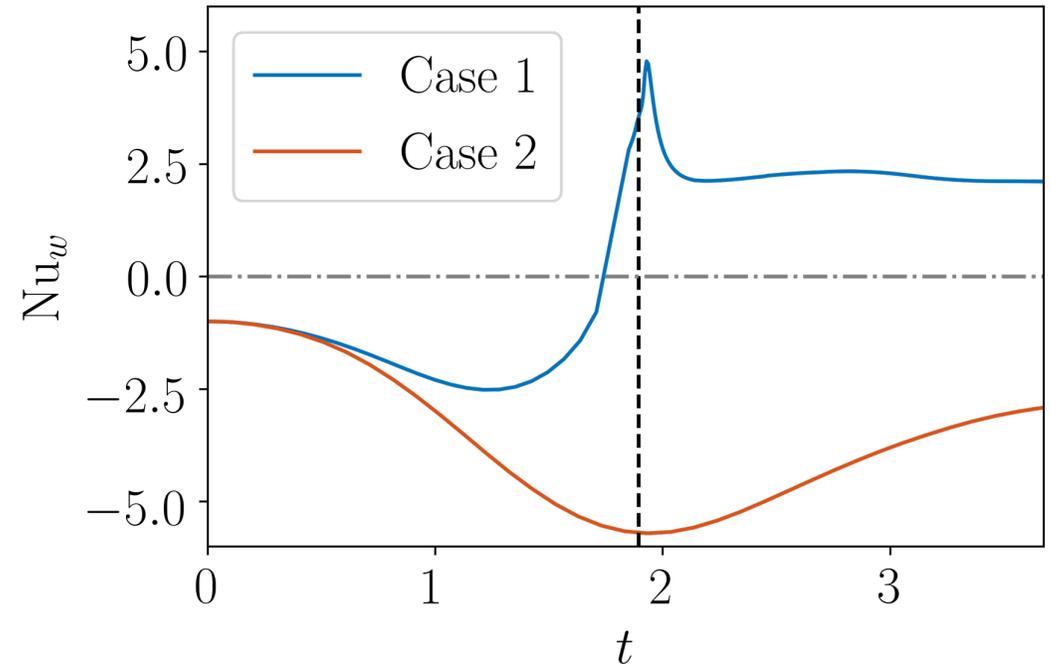
- ξ : Takeno flame index
- $\xi > 0$: premixed flame
 - $\xi < 0$: non-premixed flame

Igniting (Da=120)

Results: Ignition phenomenology

Wall heat transfer

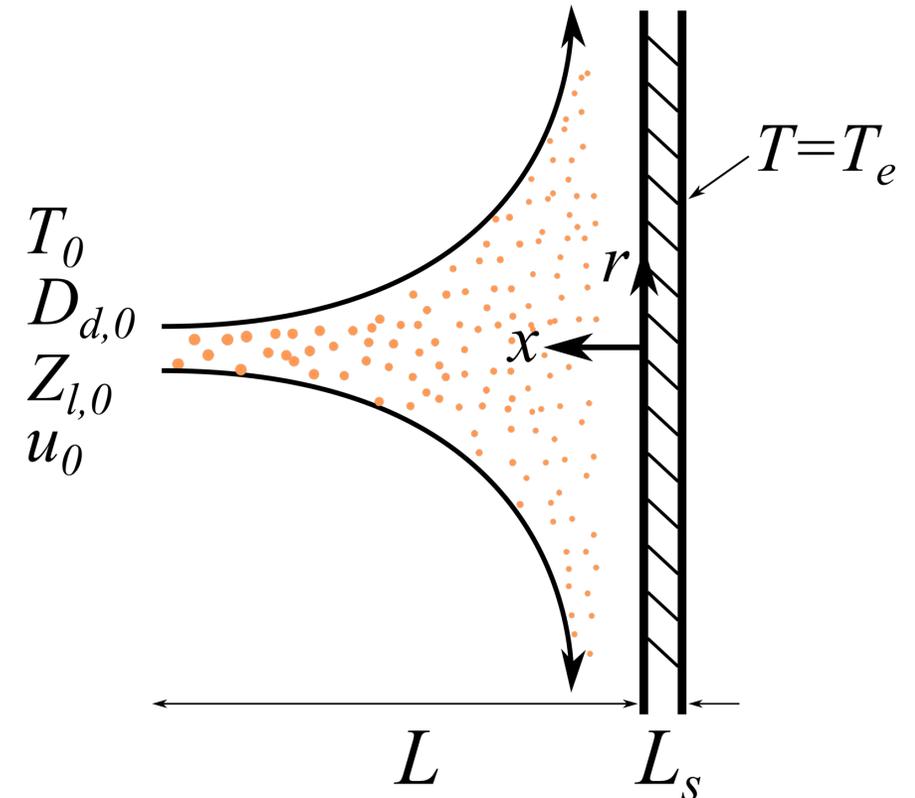
- Case 1: igniting
 - $t_{ign} \approx 1.9$
 - $\max(\text{Nu}_w) \approx 4.9$
- Case 2: non-igniting
 - Steady-state at $t \approx 5$ of $\text{Nu}_w \approx -3$
- Conjugate effects small on ignition time scale, wall is nearly isothermal
 - $1/\text{Fo} \sim 10$



Parametric study

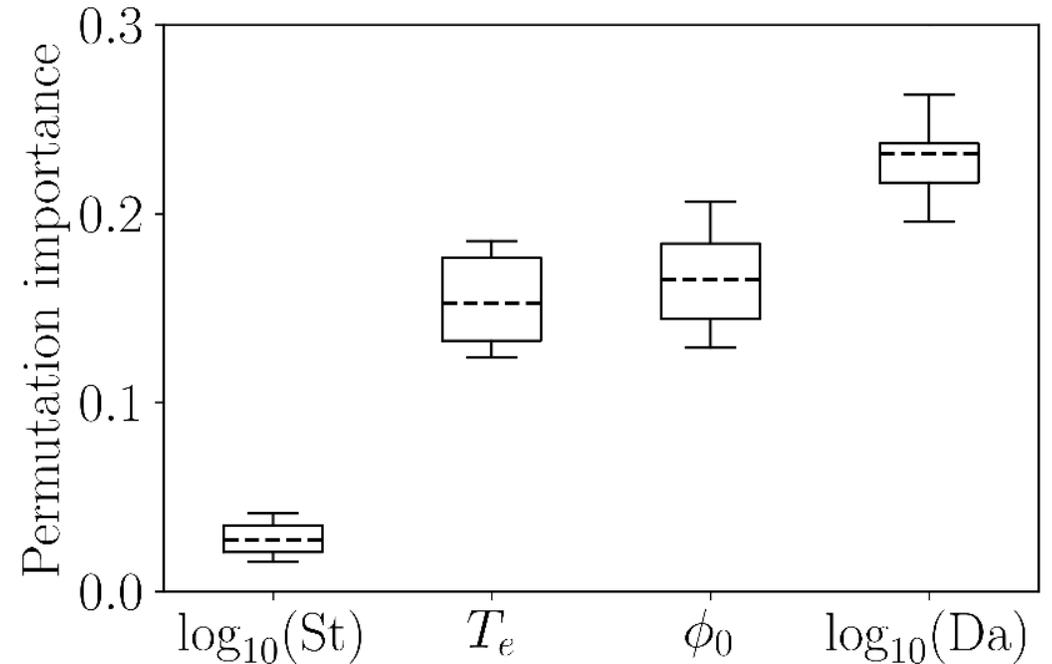
- For each simulation, consider a set of parameters
 - T_e, ϕ_0, St, Da
- Each simulation gives a set of quantitative results
 - Igniting (yes/no), $t_{ign}, \phi_{ign}, x_{ign}$
- Consider ~1000 simulations
 - Identify parametric sensitivities and system behavior

Parameter	Range
T_e	[2.75, 3.25]
ϕ_0	[0.5, 11.5]
St	[0.001, 0.3]
Da	$[10^1, 10^3]$



Parametric study

- To quantify importance of various parametric dependencies of ignition quantities, apply data analysis technique from machine learning
- Non-dimensional parameters \rightarrow features
 - T_e, Da, St, ϕ_0
- Solution information \rightarrow outputs
 - Ignited (yes/no), $t_{ign}, x_{ign}, \phi_{ign}$
- Trained data-driven model: random forest (RF)
 - Ignited (yes/no) \rightarrow RF classifier
 - $t_{ign}, x_{ign}, \phi_{ign} \rightarrow$ RF regressor
- Considered “permutation importances” for each output



Permutation importances for ignition classifier

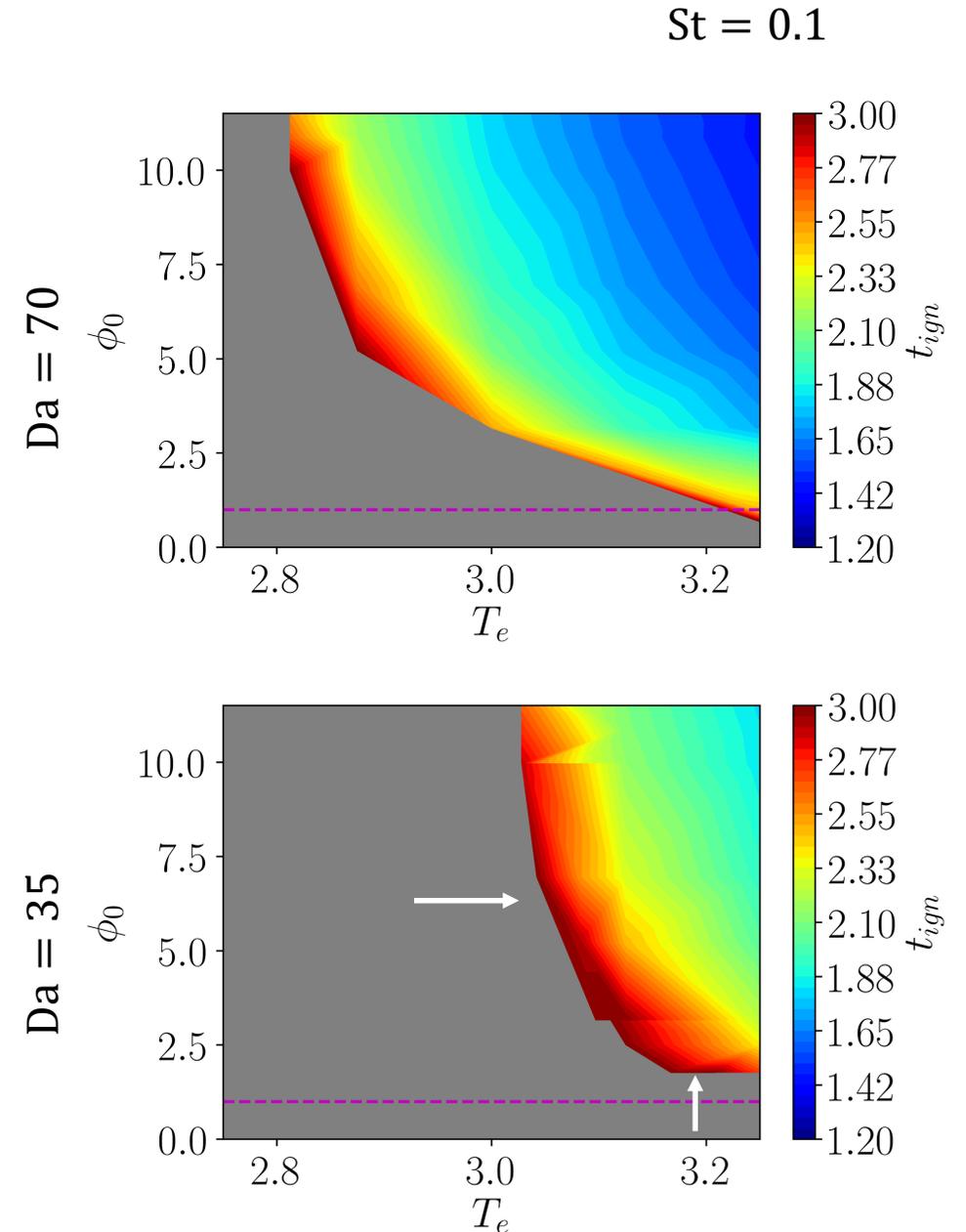
Results: Parametric study

Ignition limits:

- Reduced Da makes lean and near-stoich ϕ_0 non-igniting
- Large contraction in T_e

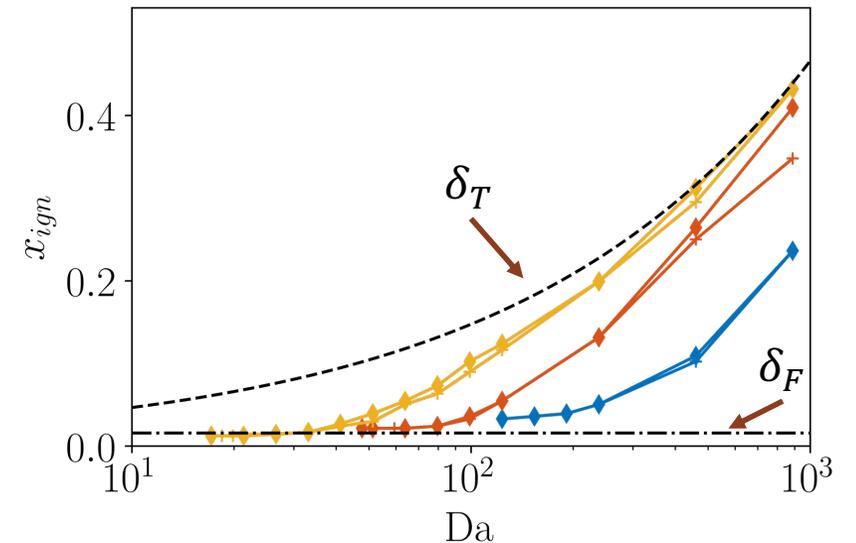
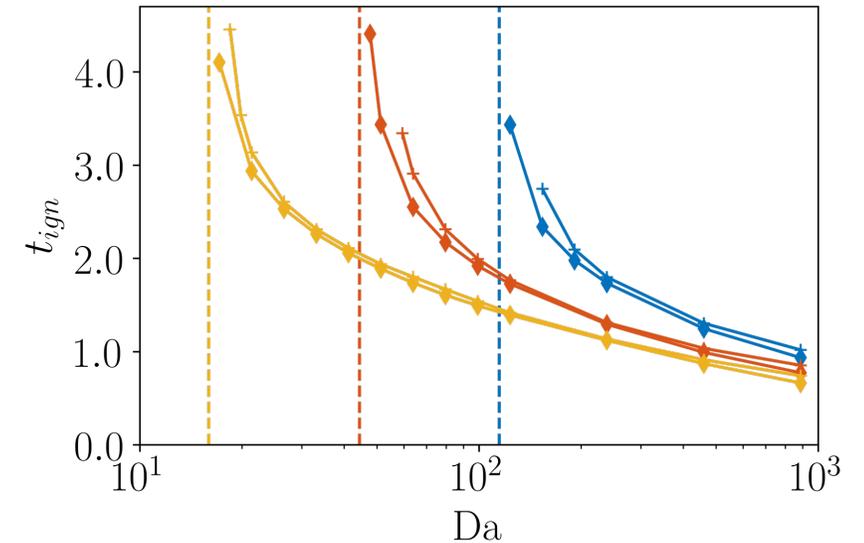
Ignition delay time:

- Reaches maximum at ignition limit
- Minimum at high T_e, ϕ_0
- Modest variation relative to purely chemical ignition delay



Results: Parametric study

- Reduced Da results in increased t_{ign} as ignition limit is approached
- Increased St causes small contraction of ignition limit
- At high Da , ignition location x_{ign} varies proportionally to thermal boundary-layer thickness δ_T
- At low Da and up to the ignition limit, ignition kernels do not form closer to the wall than the laminar premixed flame thickness δ_F
 - $Pe_{\delta_F} = x/\delta_F \sim 1$, as in flame quenching literature



$\phi_0 = 3.0$ $T_e = 2.75,$
 $St = 0.01$ (\blacklozenge) $T_e = 3.00,$
 $St = 0.1$ ($+$) $T_e = 3.25$

Summary

Hot surface ignition of wall-impinging fuel sprays

- Important phenomenon in analysis of industrial and aero-engine safety
- Can occur due to fuel leakage near surfaces at elevated temperatures

Detailed modeling of spray hot-surface ignition

- Demonstrated and analyzed ignition kernel formation, propagation
- Shorter ignition delay from higher wall temperatures can result in reduced transient wall heat flux

Lower-order modeling of spray hot-surface ignition

- Allowed direct comparison of igniting vs. non-igniting phenomenology
- Damköhler number, wall temperature, total equivalence ratio and Stokes number determine ignition limits

Further research needs

Lack of validation data

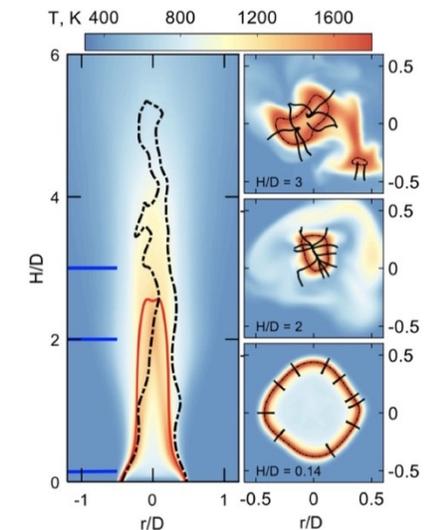
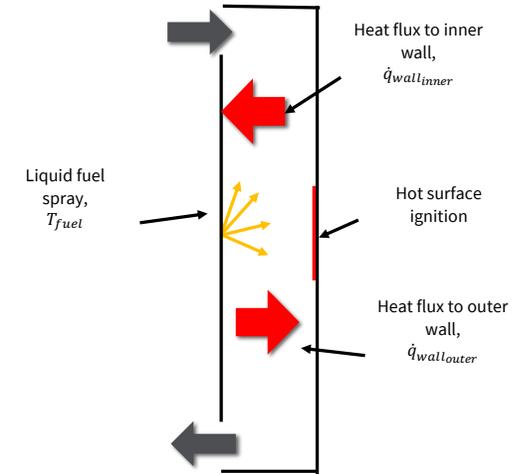
- Critical need to complement computational investigation with experimental measure to validate simulations
- Canonical experimental to enable parametric studies and support certification
- Knowledge transfer to FAA for certification

Computational modeling

- Extend modeling effort to account for equally important physical processes of
 - › Heat-transfer and structural degradation
 - › Leakage and pool fire formations

Innovative data analytics

- Integrate data-analytic models to support
 - Development of low-order models
 - Discovery of physical relations and evaluate stability limits
 - Inform certification processes



Heat Transfer Analysis for Hot Plate Design

Consider the hot surface design heater needs as an energy balance problem:

Some efficiency of heat transferred
from heaters to material

Heat Needed
From Heaters

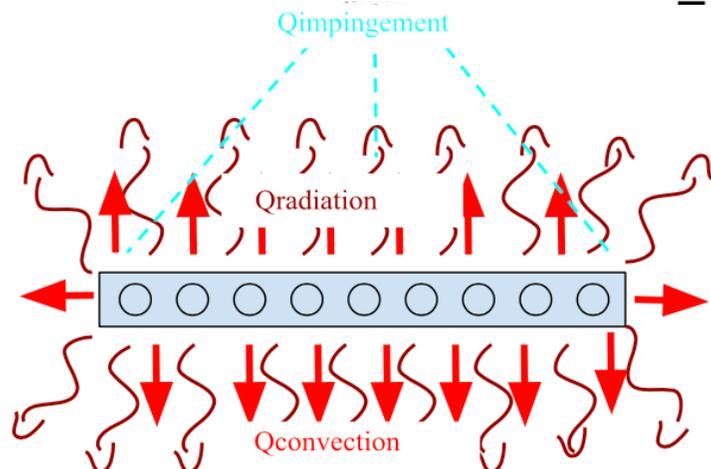


Image By Jesus Cervantes

$f(m, C_p, \Delta T)$

Heat Needed to Raise
Mass to $T_{Hot\ Surface}$

Heating Time Scale
(Specified)

Convection
Losses to Fluid

Conductive
Losses to
Insulation

Heat Lost to
Environment

Convection
Losses to Air

Radiative
Losses

Heat Transfer Analysis | Methodology

$$Nu = 0.59Ra^{0.25} \text{ for } 10^4 < Ra < 10^9$$

for a vertical, isothermal plate
(M. Bahrami, SFU)

Total Heating Needs [J]

$$Q = mC_p\Delta T$$

*Integral of C_p over T if $C_p(T)$

Q = Total Heat [J], m = Hot Surface Mass [kg]

C_p = Specific Heat Capacity [J/kg-K]

ΔT = Change in Temperature [K]

Natural Convection Loss [kW]

$$Ra = GrPr = g\beta(T_w - T_\infty) \frac{H^3}{\nu} \alpha$$

$\beta = \frac{1}{T_{film}}$ for ideal gases where $T_{film} = \frac{T_\infty + T_w}{2}$ [K], T_w = Wall Temp

[K], T_∞ = Ambient Temp [K], H = Hot Surface Height [m], ν = Kinematic Viscosity [m²/s], α = Thermal Diffusivity [m²/s]

Used to find Nu, which is used to find heat transfer coefficient, h

Fluid Impingement Loss [kW]

$$Q = Q_{vap}$$

Conservative Estimate: Assume all fluid contacting plate evaporates, so the heat loss is associated with the heat necessary for the phase change. **Lack of good models to estimate these impacts.**

Radiation Loss [kW]

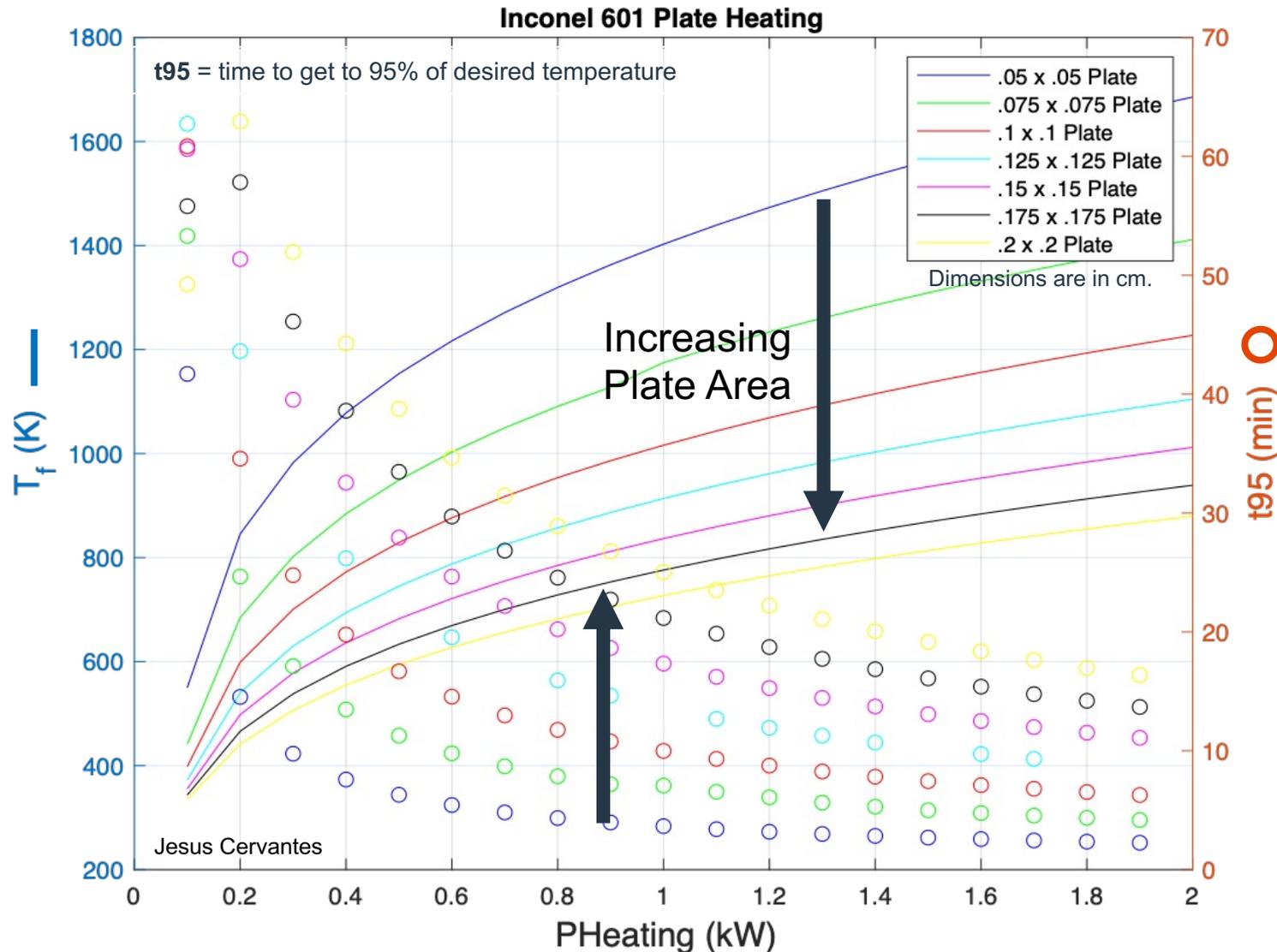
$$Q = \varepsilon\sigma A(T_{surface}^4 - T_{surroundings}^4)$$

*Emissivity varies with temperature

Q = Heat Flux [W], ε = Emissivity,

σ = Stefan-Boltzmann constant, T_{surf} = Surface Temp [K], $T_{surroundings}$ = Environment Temp [K]

Heat Transfer Analysis | Plate Dimensions Impact

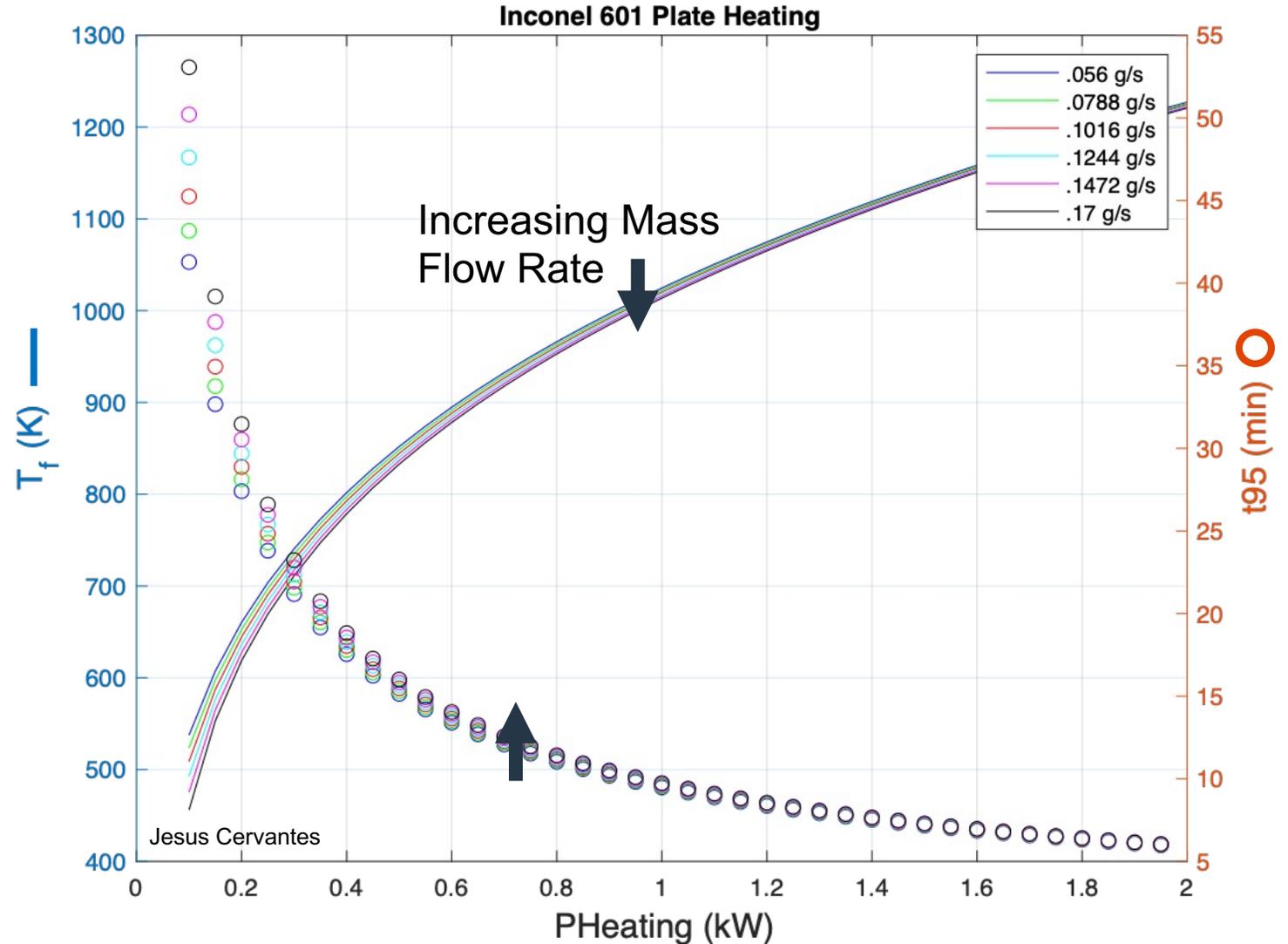


- A**
- Increased plate area **substantially decreases temp** and **increases t95**
 - 30 to 65 min t95 range
 - 900 to 1700 K range
 - **Δ 35 minutes @ 2 kW** between min and max area
 - **Δ 800 K @ 2 kW** between min and max area

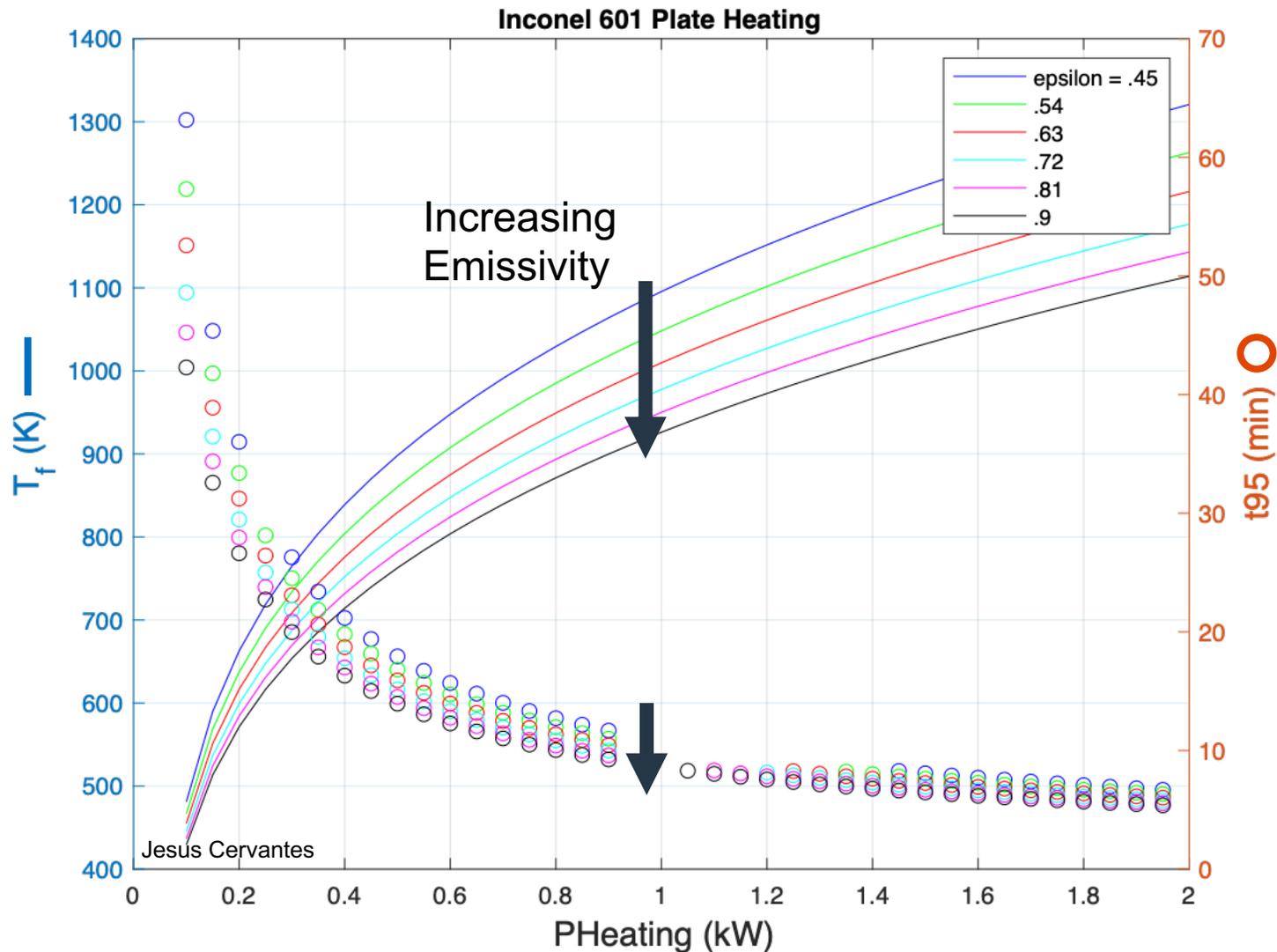
Heat Transfer Analysis | Mass Flow Impact

m

- Increased mass flow rate minimally **decreases temp** and **increases t95**
 - 50.5 to 51 min t95 range
 - Asymptotes to ~420 K
- $\Delta 30\text{ s}$ @ 2 kW between min and max mass flow
- $\Delta < 10\text{ K}$ @ 2 kW between min and max mass flow

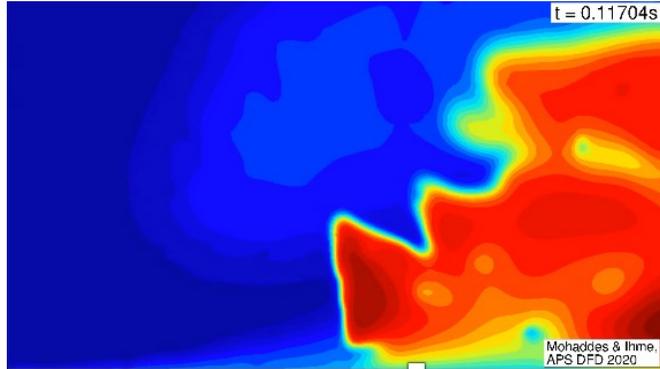


Heat Transfer Analysis | Emissivity Impact


 ϵ

- Increased emissivity notably **decreases temp and t95**
 - 50 – 65 min t95 range
 - Asymptotes to ~480 K
- **Δ 15 min @ 2 kW** between min and max area
- **Δ 5 K @ 2 kW** between min and max area

Conclusions



Hot surface ignition of wall-impinging fuel sprays
important phenomenon in analysis of industrial and
aero-engine safety

Can occur due to fuel leakage near surfaces at elevated
temperatures

Through detailed simulation and low-order modeling, we
identified key parameters that control hot surface
ignition: L, T_e, ϕ_0, St, Da

Identified low-surface temperature conditions ($< 1000\text{K}$)
to be more critical

- Enhanced wall heat flux
- Longer ignition delay
 - more vaporized fuel available
 - more air entrainment

Experimental testing imperative for certification