

# FAA Microscale Combustion Calorimeter Inter-Laboratory Study Update

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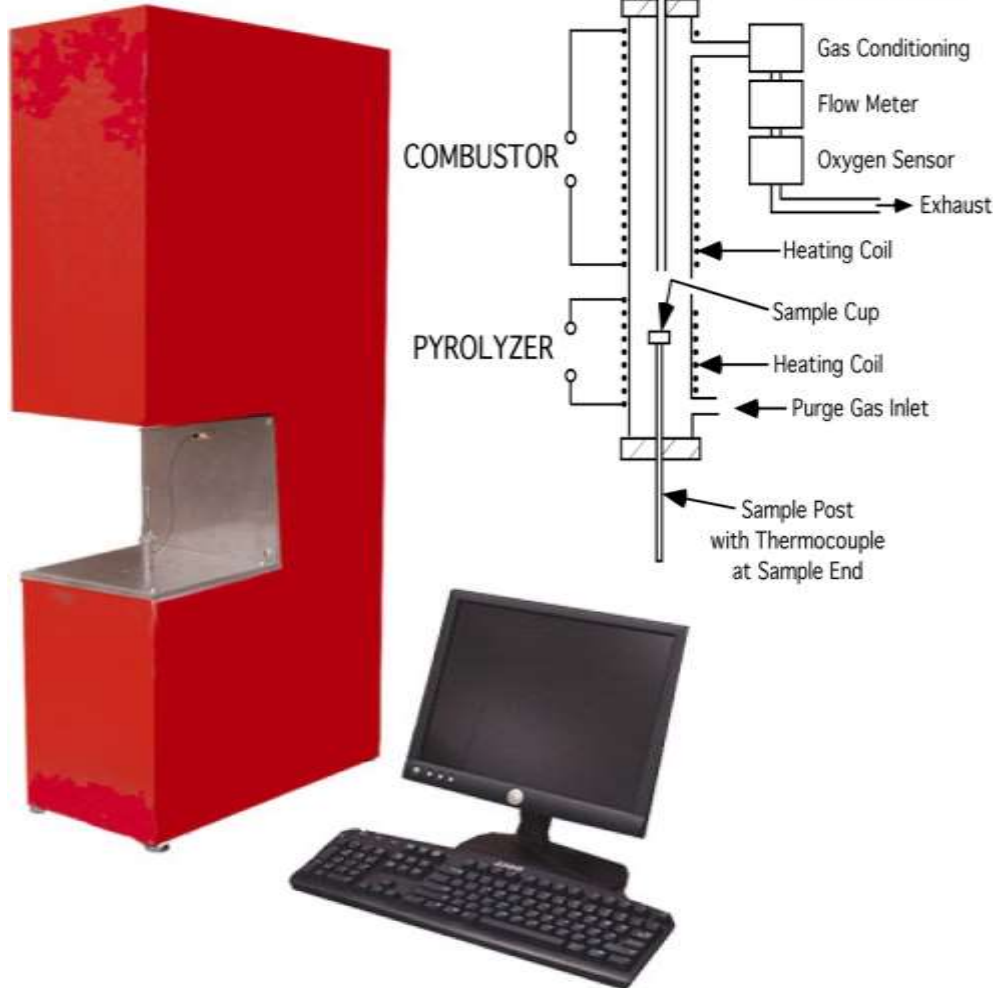


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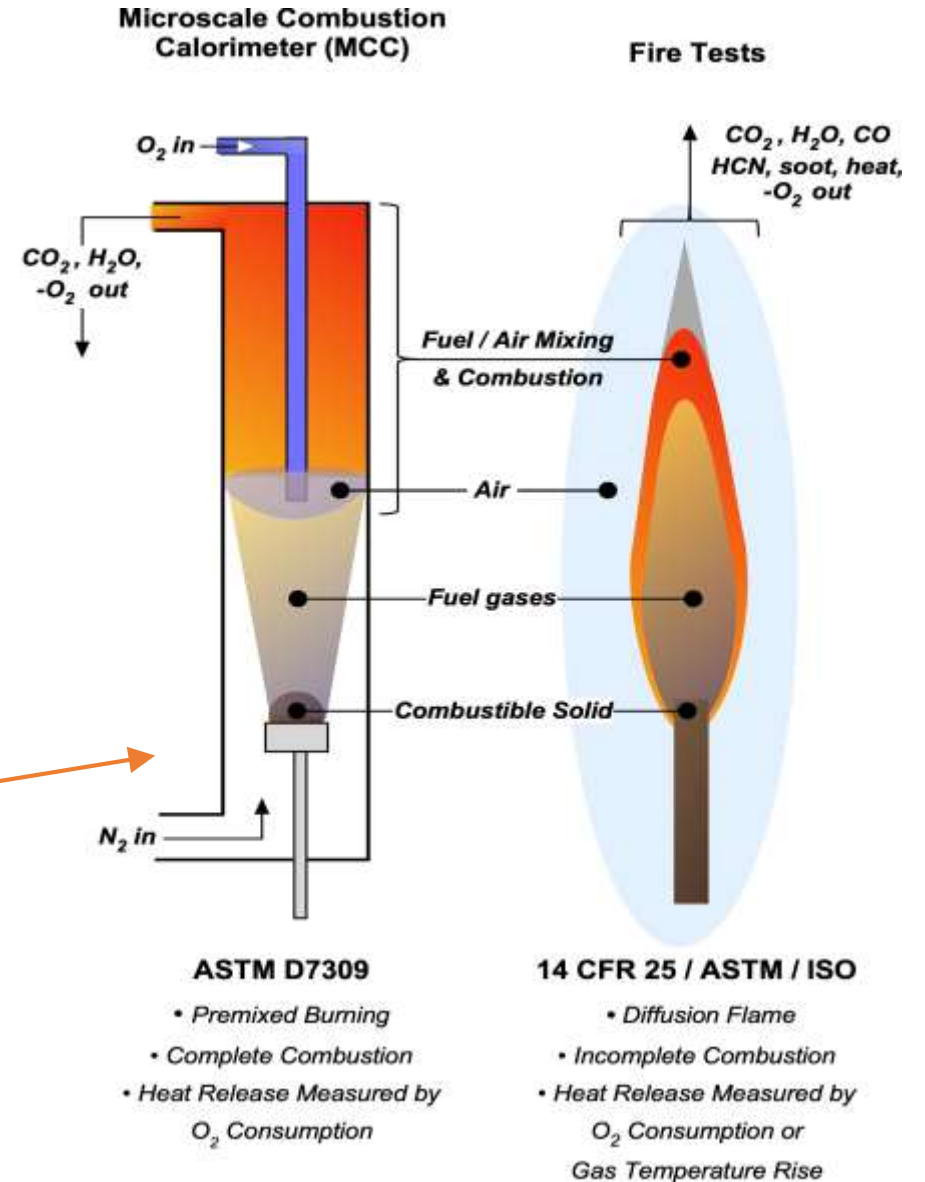
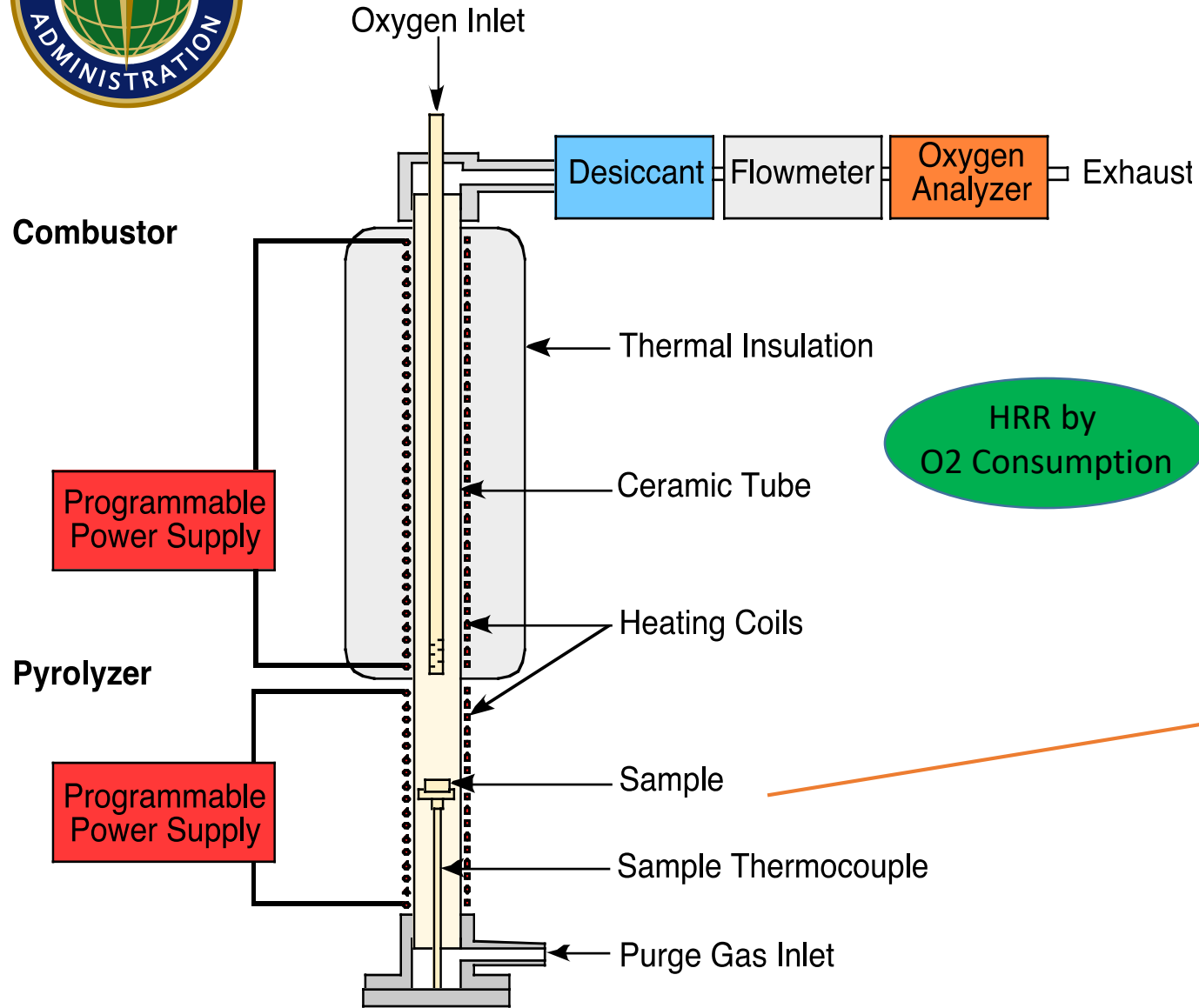
# FAA Microscale Combustion Calorimeter (ASTM D7309-21)



- ❑ The *MCC* test measures Materials Properties related to Flammability on a milligram size scale.
- ❑ One of the outputs: Fire Growth Capacity. The ***FGC*** is a measure of ignitability and burning rate of a material, i.e., the **total fire hazard**
- ❑ *MCC* is proposed method for **alternate means of compliance** when a small change is made to a construction

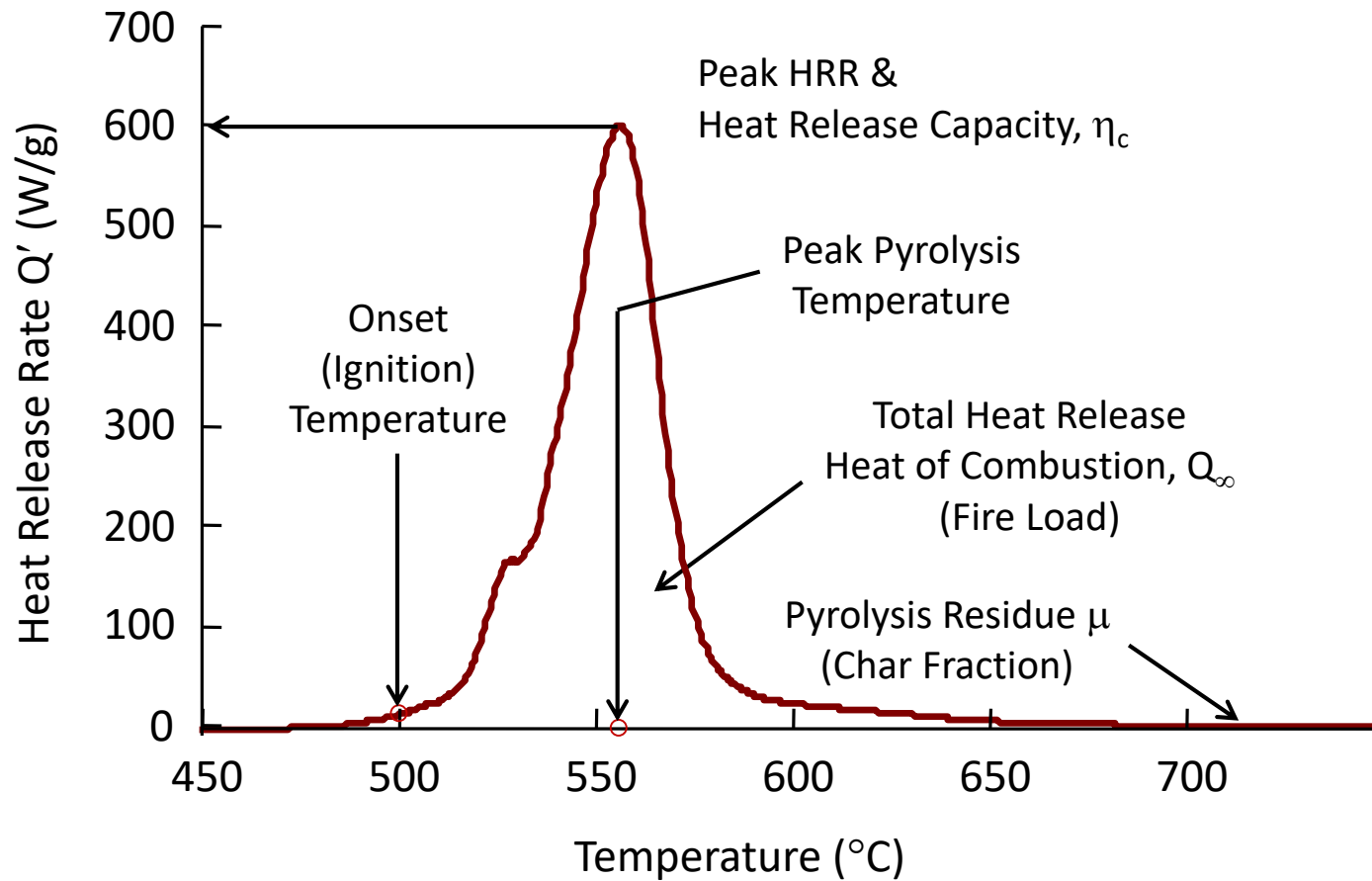


# FAA Microscale Combustion Calorimeter



# Standard Test ASTM D 7309 (Method A)

Anaerobic pyrolysis at 1 K/s + complete combustion of gases at 900°C, 20% O<sub>2</sub>

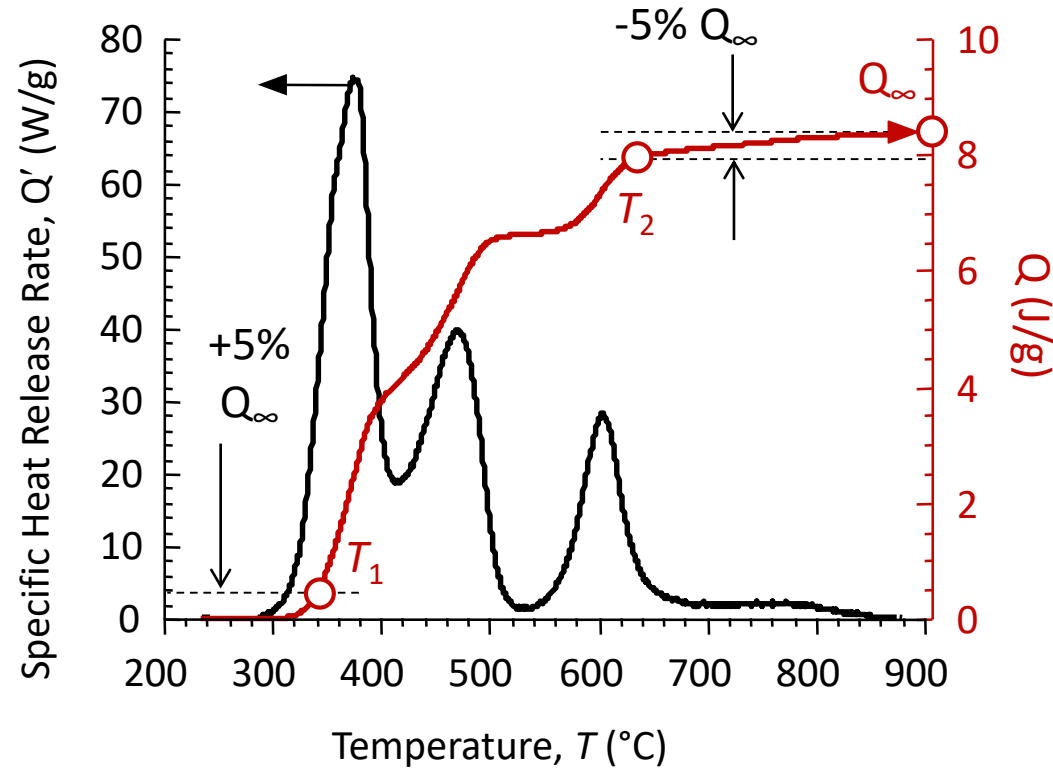


Microscale Combustion Calorimeter





# Fire Growth Capacity - FGC



## MCC procedure for FGC

1. Measure specific heat release rate  $Q'$  versus temperature  $T$  as per ASTM D7309 (5 replicates)
2. Integrate  $Q'/\beta$  versus  $T$  to obtain  $Q$  versus  $T$ , i.e.,  $Q(T)$
3. Obtain total heat release  $Q(T_\infty) = Q_\infty = h_c(J/g)$
4. Obtain  $T_1$  at 5% deflection from  $Q(T)$  baseline, i.e., at  $0.05 Q_\infty$
5. Obtain  $T_2$  at  $Q_\infty$  i.e.,  $0.95 Q_\infty$ .
6. Calculate Fire Growth capacity (FGC)

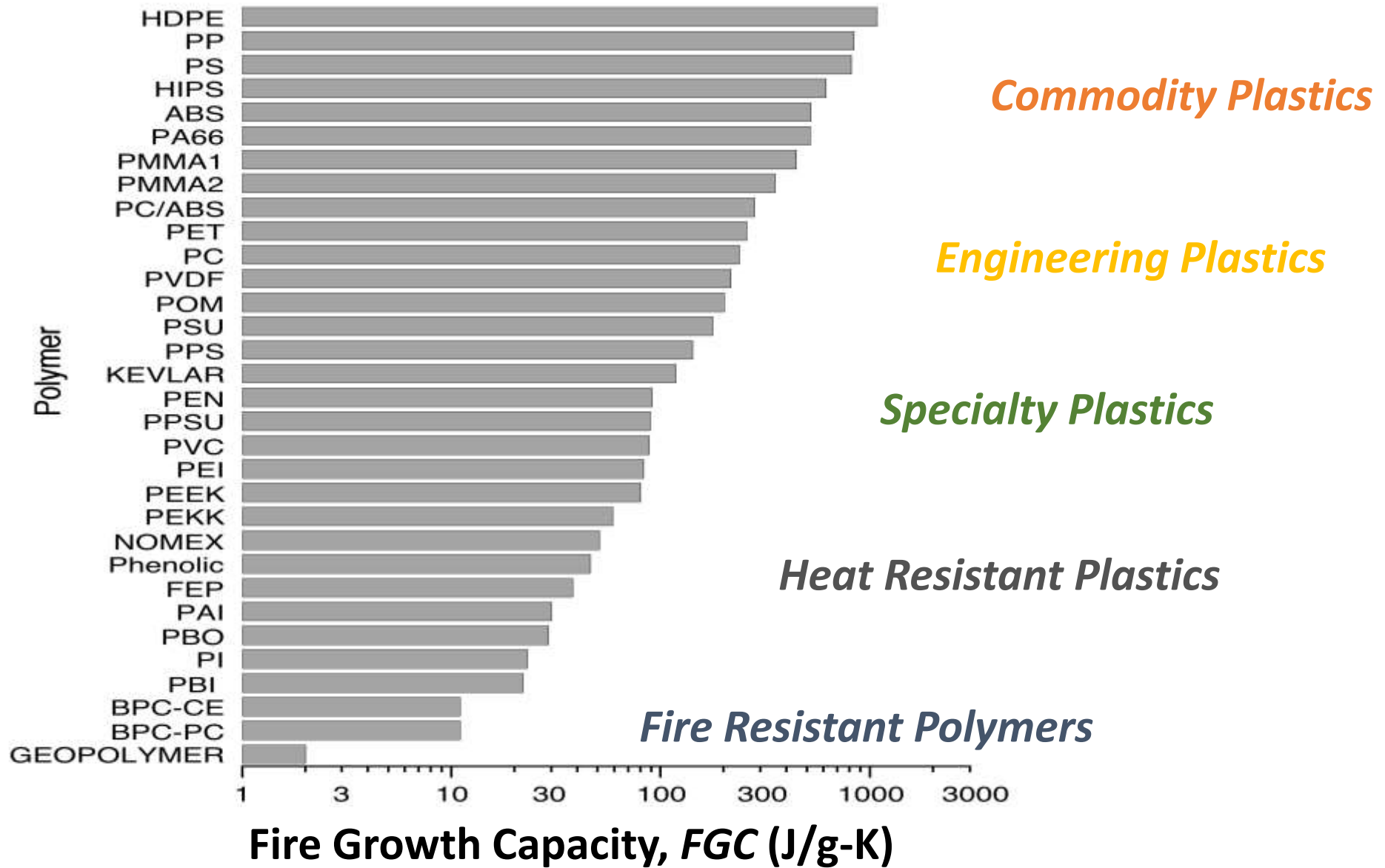
$$T_0 = 25^\circ\text{C} (298\text{K})$$

$T_1 = \text{Ignition temperature}$

$T_2 = \text{Burning temperature}$

$$FGC = \left( \frac{Q_\infty}{T_2 - T_1} \right) \left( \frac{T_2 - T_0}{T_1 - T_0} \right)$$

# Ranking of Materials



# ASTM E691

## Conducting an Inter-Laboratory Study to Determine the Precision of a Method

- **Repeatability**
  - An action, event, or other thing that is done again
- **Reproducibility**
  - Create something very similar to (something else) in a different medium or context





# Measured Values



**Sample Weight (mg)**

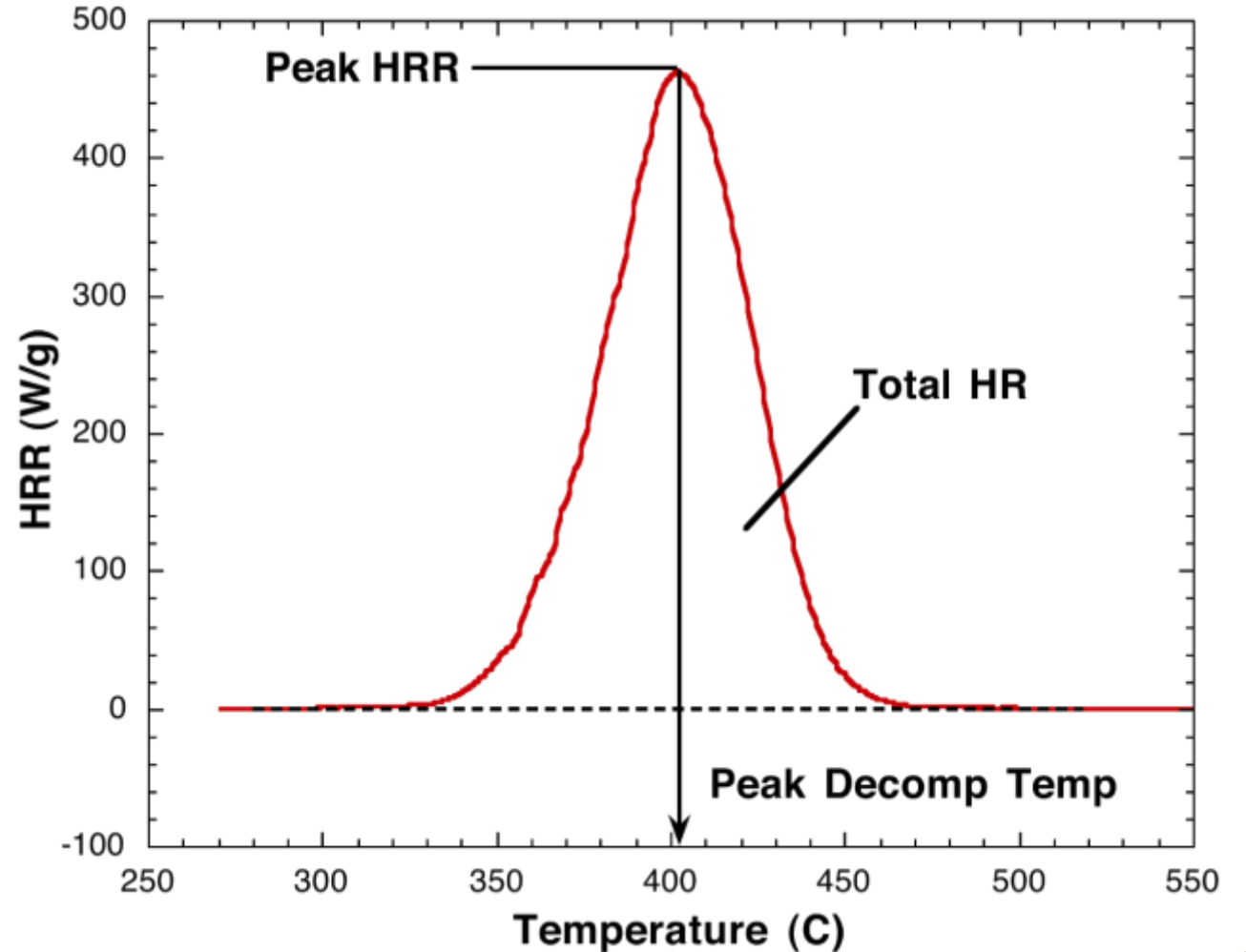
**Char Yield (%)**

**Peak Heat Release Rate (W/g)**

**Total Heat Released (kJ/g)**

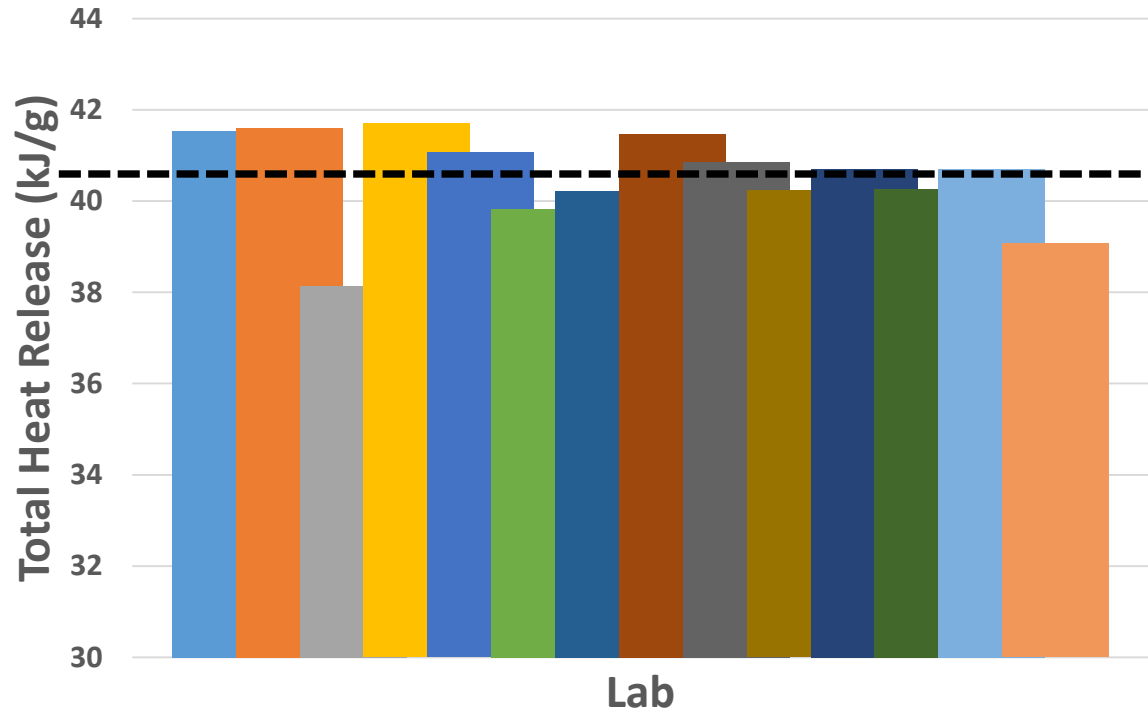
**Peak Heat Release Temperature (C)**

**Fire Growth Capacity (J/g-K)**





# Preliminary Lab Comparison – Polystyrene



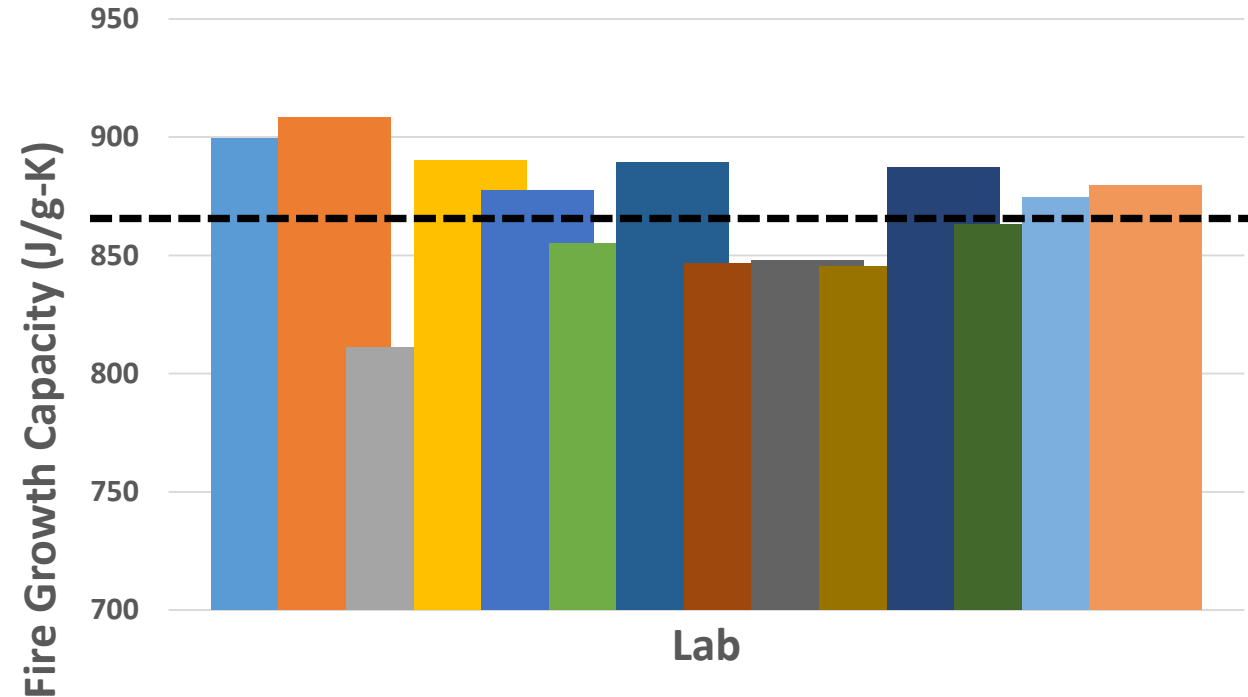
## Total Heat Release

40.6 +/- 1.1 J/g-K  
2.8%

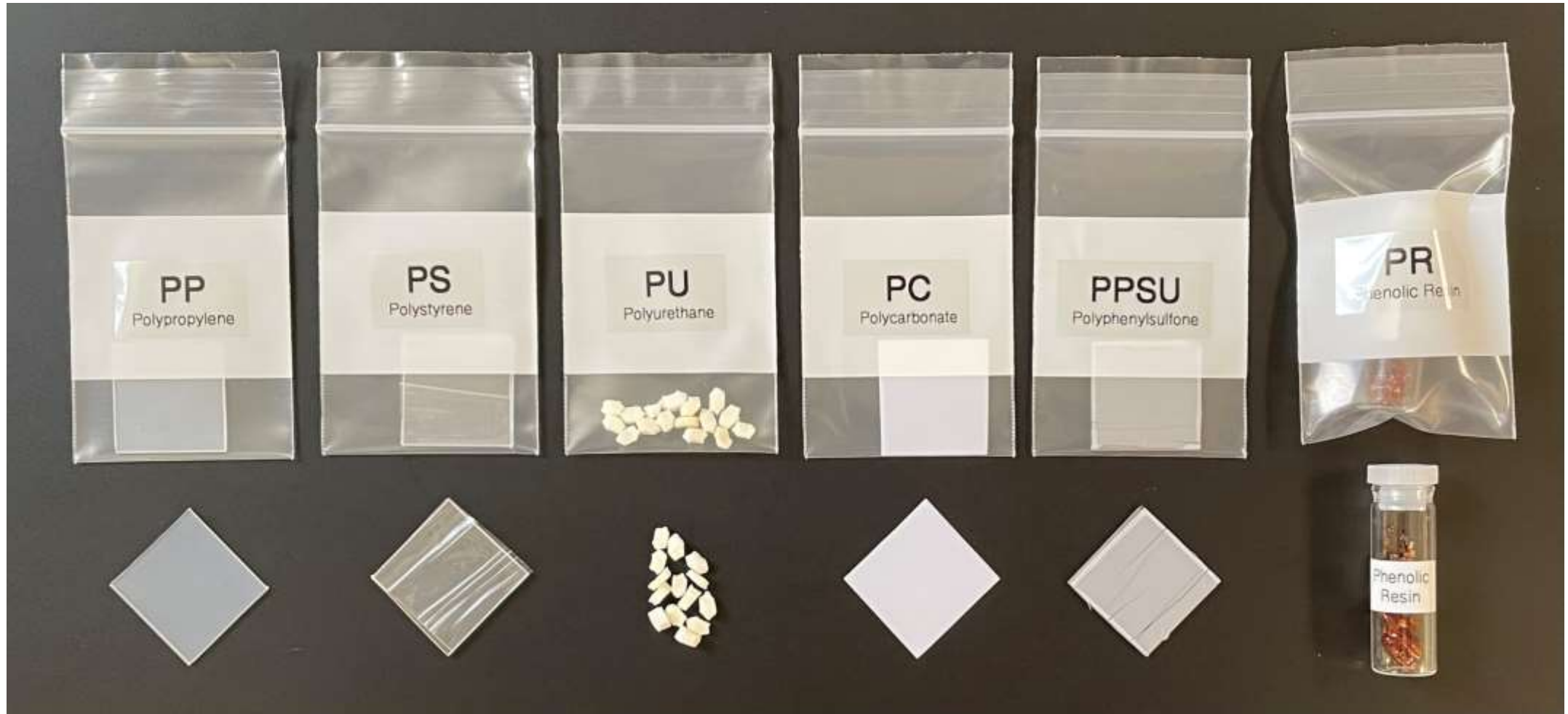


## Fire Growth Capacity

868 +/- 35 J/g-K  
4.1%



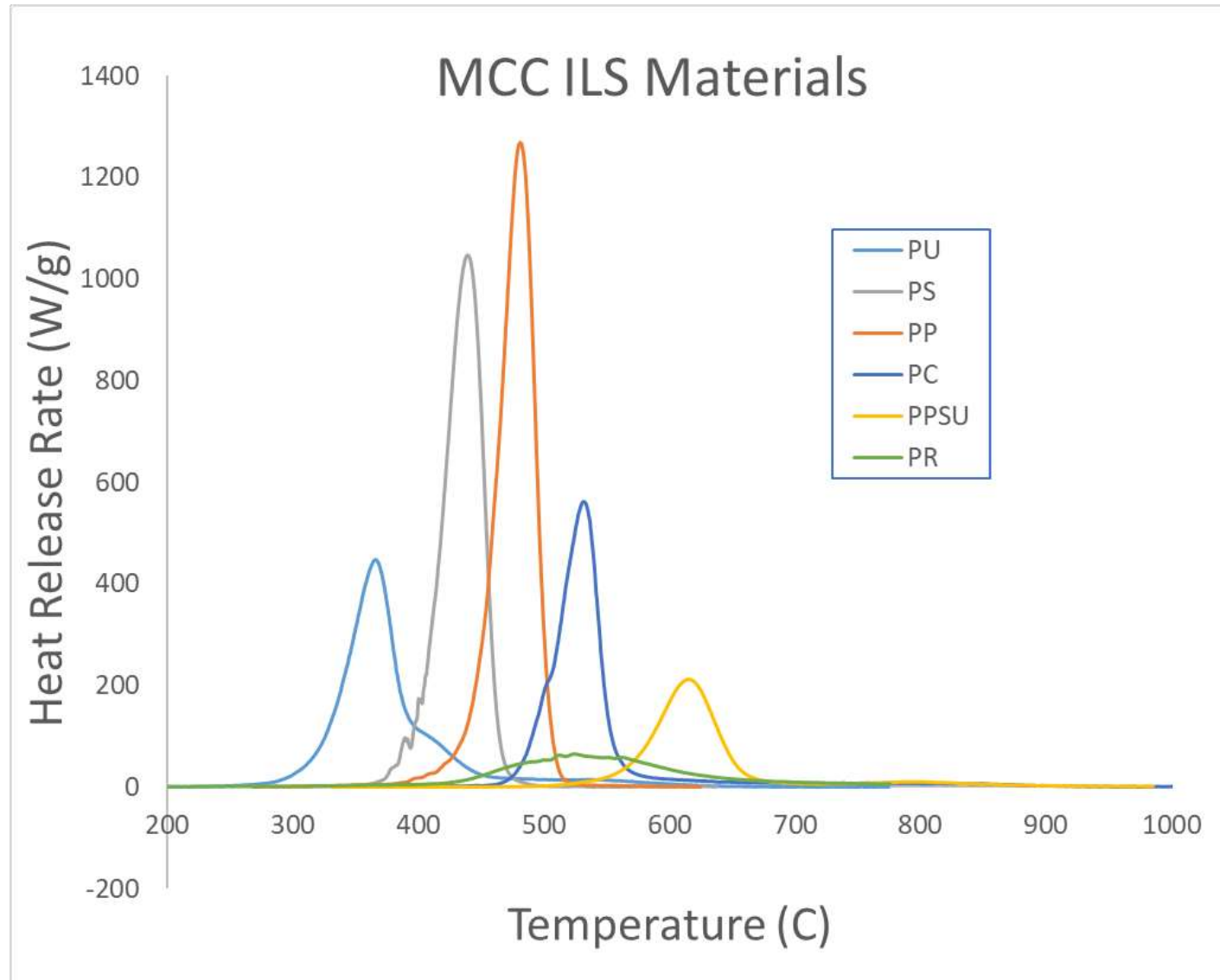
# Samples for ASTM FAA MCC ILS



# Aircraft Material - Boeing Phenolic Resin



# MCC Heat Release Rate – ILS Materials



# Inter-Laboratory Study

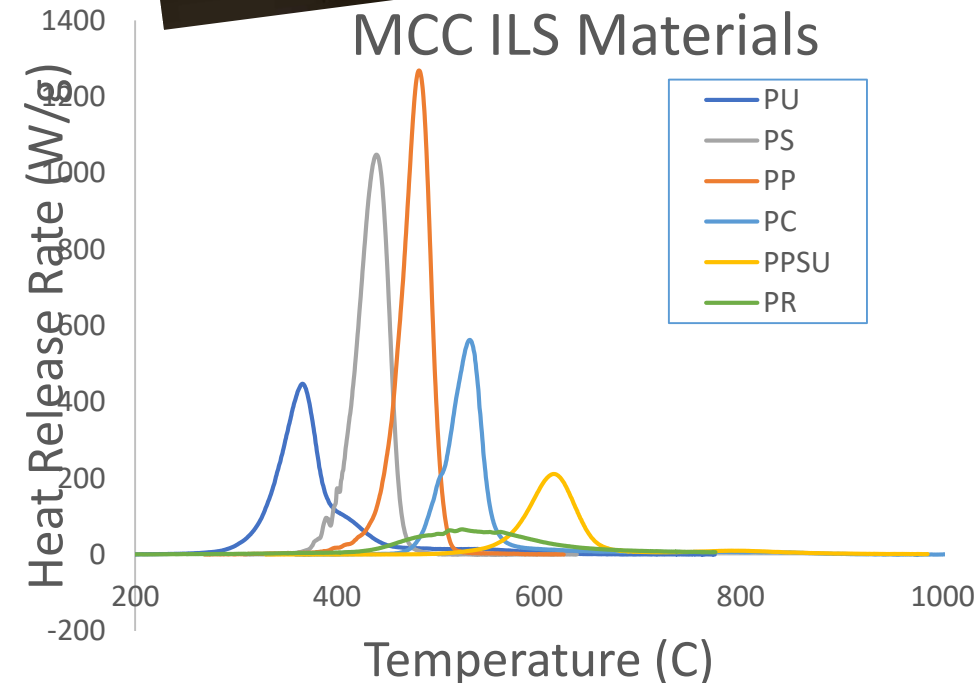
- **4 manufacturers/licensees of the MCCs out there**
  - Deatak
  - Fire Testing Technologies
  - Concept Equipment Ltd
  - Me
- **Samples shipped to labs**
  - 18 labs claimed they are able to participate
  - Several labs have more than one unit
- **Data to be sent directly to ASTM**
  - Statistical analysis
  - Update ASTM D7309 precision & bias statement





# Summary & Future Work

- **Inter-laboratory study – Round 1 (FAA)**
  - Preliminary results received
  - Offered suggestions to operators to reduce errors
- **Inter-laboratory study – Round 2 (ASTM)**
  - Sample set selected (6 Samples)
  - Participants verified (20+ MCCs)
  - FGC software created
  - Input values into ASTM ILS website
  - Finalize with ASTM
  - Send samples to labs
  - Submit results directly to ASTM
  - ASTM handling the statistics
  - Update precision & bias statement
  - FAA report



# FAA Reports - <https://www.fire.tc.faa.gov/>

DOT/FAA/TC-20/30

Federal Aviation Administration  
William J. Hughes Technical Center  
Aviation Research Division  
Atlantic City International Airport  
New Jersey 08403

## Microscale Fire Test for Component Substitutions in Aircraft Cabin Materials

# Similarity

September 2009

Final Report

This document is available to the U.S. public  
through the National Technical Information  
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DOT/FAA/TC-12/53, R1

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## Principles and Practice of Microscale Combustion Calorimetry

# MCC

April 2013

Final Report

Revised: December 2014

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DOT/FAA/TC-28/08

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## A Physical Basis for Comparing Flammability of Aircraft Cabin Materials Using a Microscale Combustion Calorimeter

# FGC

August 2009

Final report



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## Microscale Combustion Calorimeter: Interlaboratory Study of Precision and Bias

# ILS

Richard N. Walters  
Richard E. Lyon

December 2012

Final Report

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