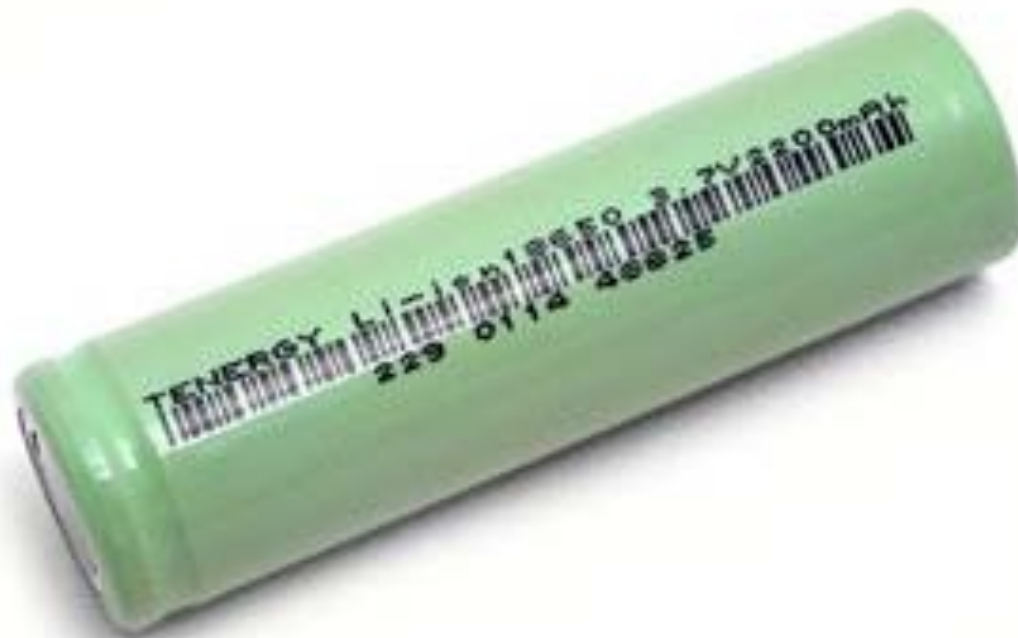


# ENERGY RELEASE OF LITHIUM ION BATTERIES AT DIFFERENT STATES OF CHARGE



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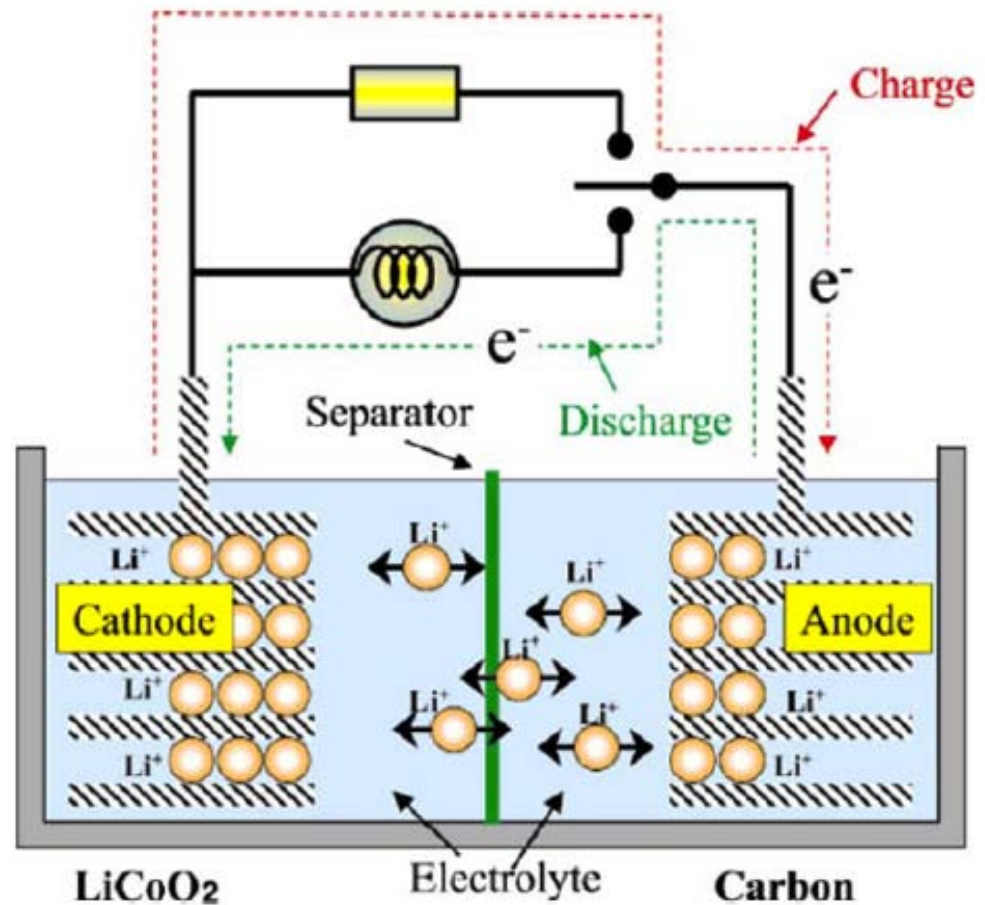
# Objective

- **Characterize energy released from batteries at different states of charge (SOC)**



# Typical Li Ion Battery

- 18650: 18 mm diameter, 65 mm length
- ~ 44 g
- Anode
  - Graphite, Copper
- Cathode
  - Li oxide, aluminum
- Separator
  - PE or PP membrane
  - Regulates ion flow
- Electrolyte
  - Organic combustible



# Causes of Thermal Runaway

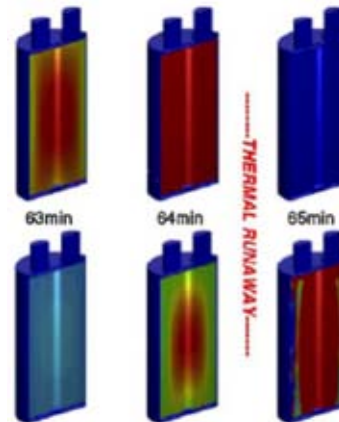
- Thermal
  - Critical temperature triggers
    - Due to heat source
    - Separator melts, inoperative
- Mechanical
  - Physical damage
    - Li dendrite grows to cause short
- Electrical
  - Overcharge
  - Rapid discharge



All lead to temperature increase and acceleration of chemical decomposition

# Background

- Over 30 years of R&D on Li-ion batteries
- Increasing applications
- Modeling of Runaway
  - Sophisticated
  - Up to 6 decomposition reactions
  - CFD thermo-chem-electrical analyses
- Experimental
  - Component studies
  - DSC, ARC devices



Phone

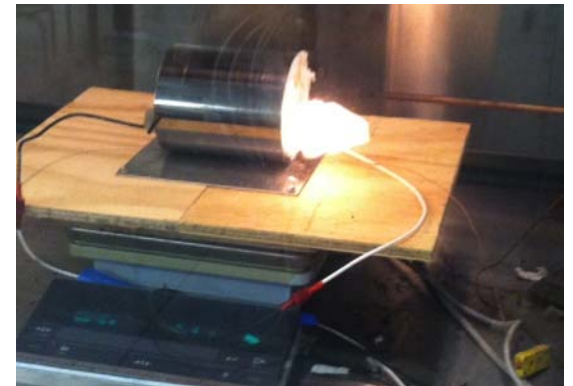
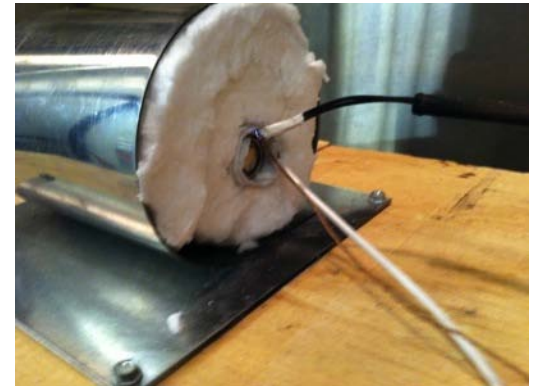


Car

# Thermal Runaway Energy Measurements

## 18650 Li-Ion Battery

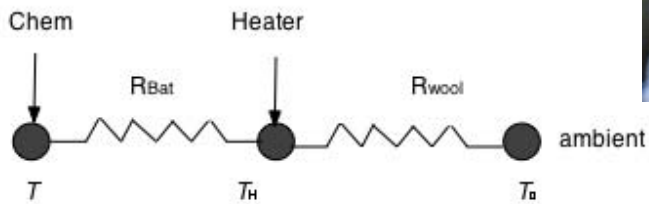
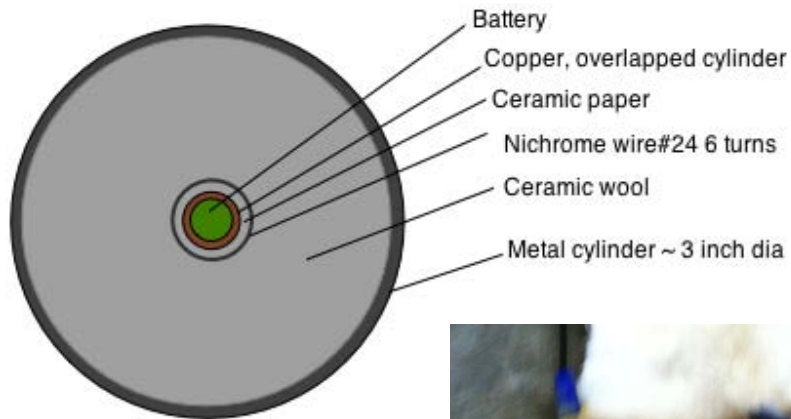
- Build calorimeter with known heat input
- 18650 characteristics
  - 2.6 Ah @ 3.7 V or 34.6 kJ of electric power available
  - Separator softens at 130°C and melts > 150°C
  - PRV @ ~ 200 psi: white vapor emitted
  - ~ 250 °C onset of runaway, jump to ~ 800°C in seconds
  - Gases emitted: CO<sub>2</sub>, CO, H<sub>2</sub>, CH<sub>4</sub>, other HC
  - Solids emitted: Cu, Graphite, Molten Al
- Chemical – Electrical mechanisms not discerned





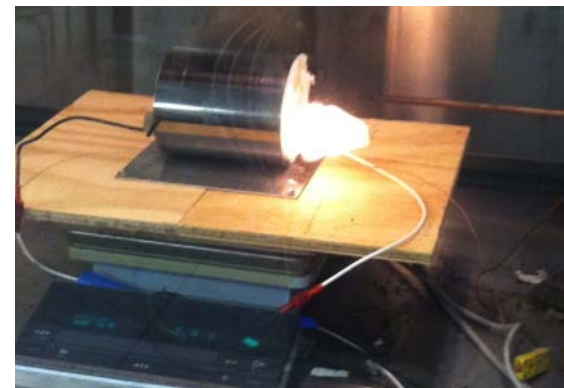
# Apparatus Design

## Temperature & Mass Measured

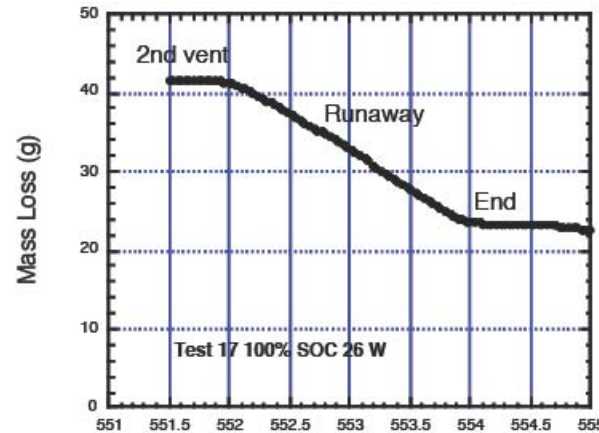
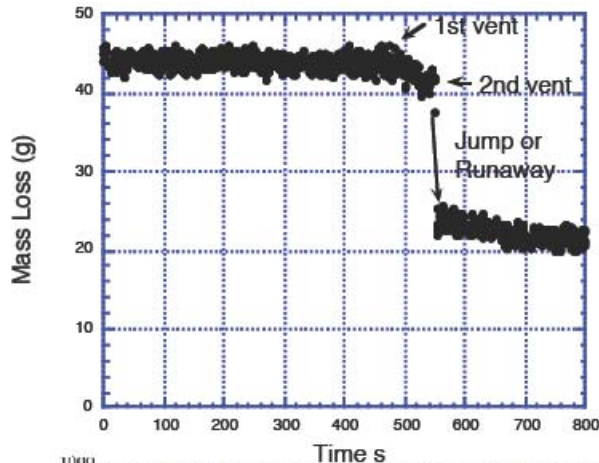


THERMAL MODEL

Heater:  $I * V = \text{Watts}$

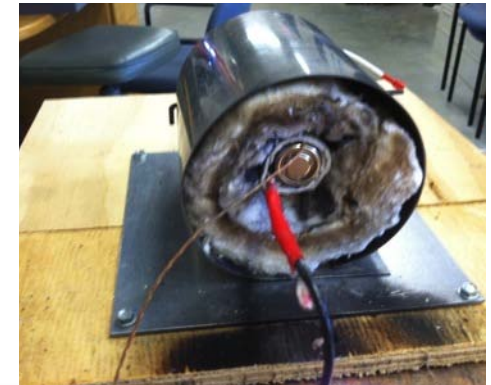
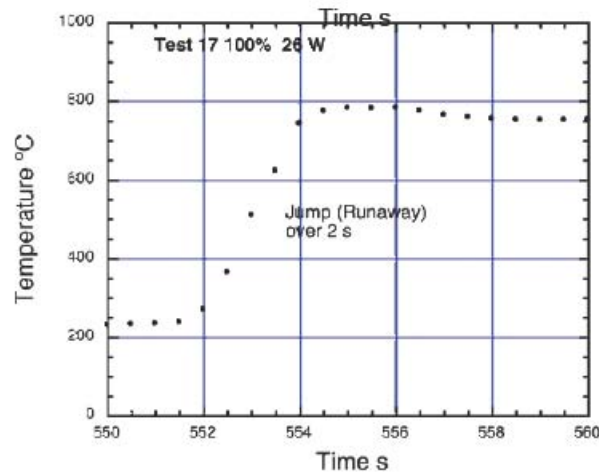
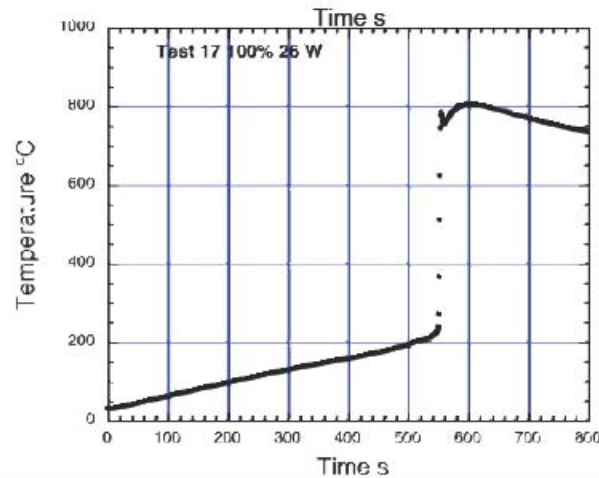


# Typical Battery Thermal Runaway 100% SOC



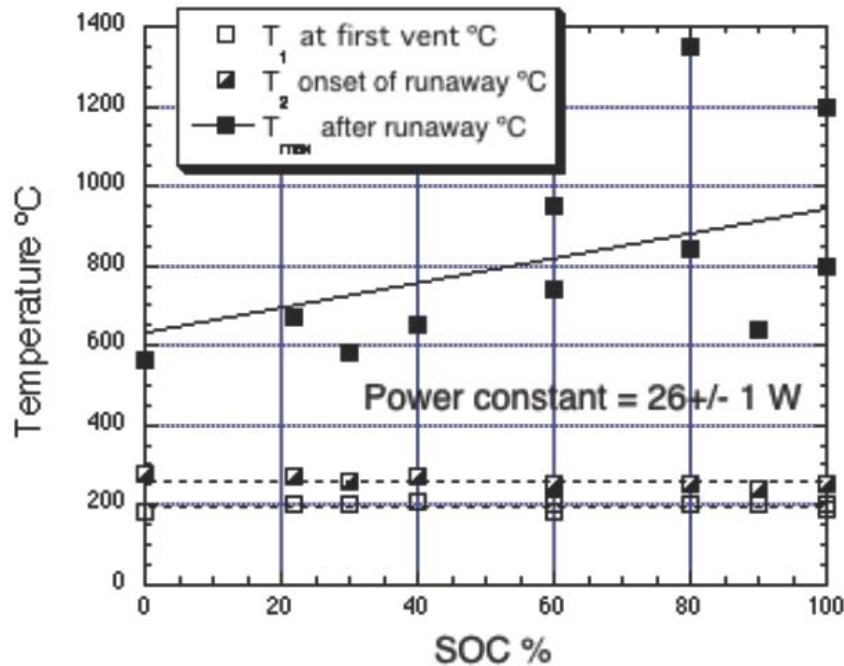
1<sup>st</sup> vent @ 470 s  
 Lose ~ 2 - 3 g  
 ~ 200 °C

2<sup>nd</sup> vent @ 552 s  
 Runaway ~ 2 s  
 Lose ~ 17 g  
 T: ~ 250 °C to ~ 800 °C





# Temperature and Mass Loss Measurements

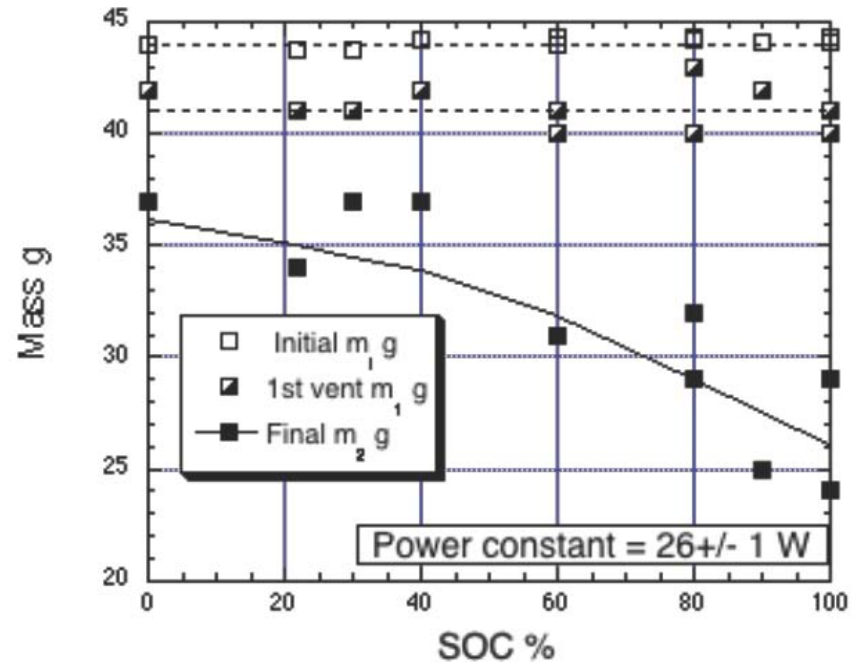


-Temperature of venting and thermal runaway onset independent of SOC

-Maximum temperature of runaway dependent on SOC

-Mass lost during 1<sup>st</sup> vent independent of SOC

-Total mass lost during runaway dependent on SOC



# **Chemical Energy Measurements**

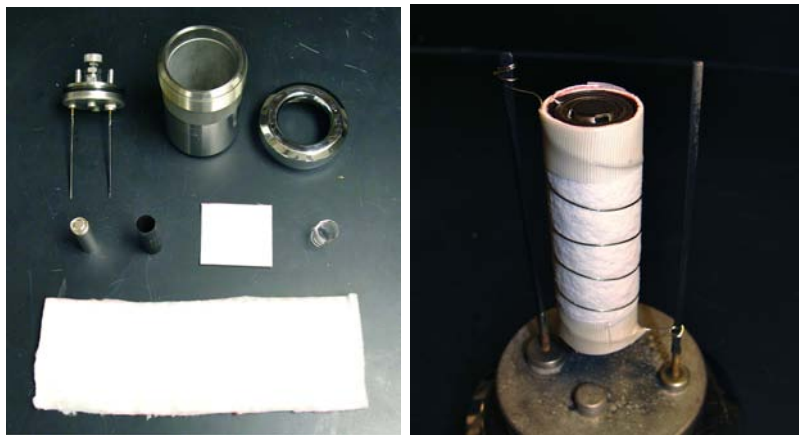
- **ASTM bomb calorimeter method was modified to measure chemical energy of battery exotherm during thermal runaway**
- **Battery heated in bomb pressurized with N<sub>2</sub> until thermal runaway**
- **Energy release calculated from temperature rise of bomb**
- **Baseline test was run after battery test without disturbing contents to keep mass the same**
- **Chemical energy from reaction at different SOC measured**

# Bomb Calorimeter

**Parr Instruments Model 1341 Plain Jacket Oxygen Bomb Calorimeter**

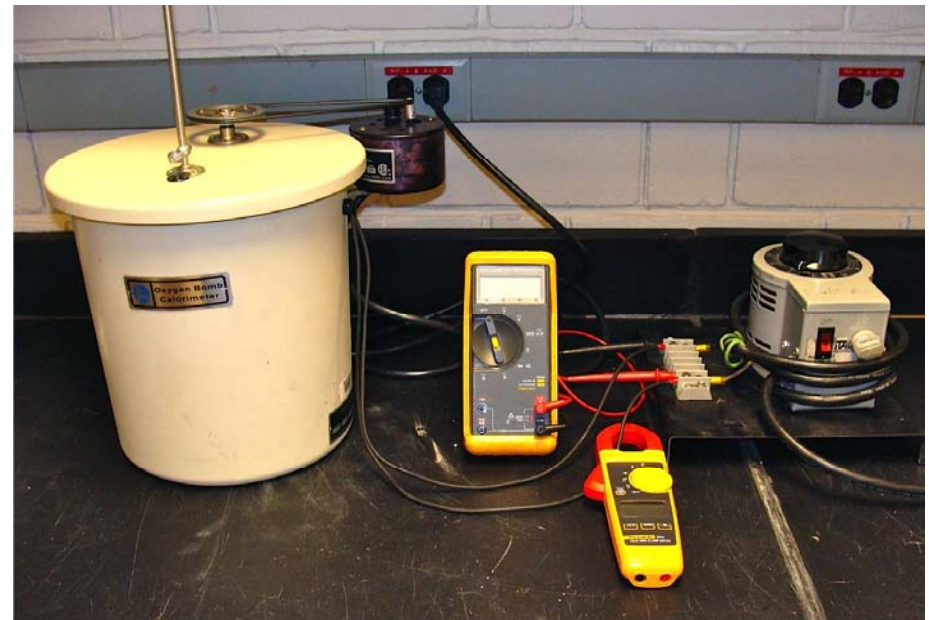
**Voltage and current applied to force thermal runaway was the same for all tests**

**Temperature data logged for all tests**

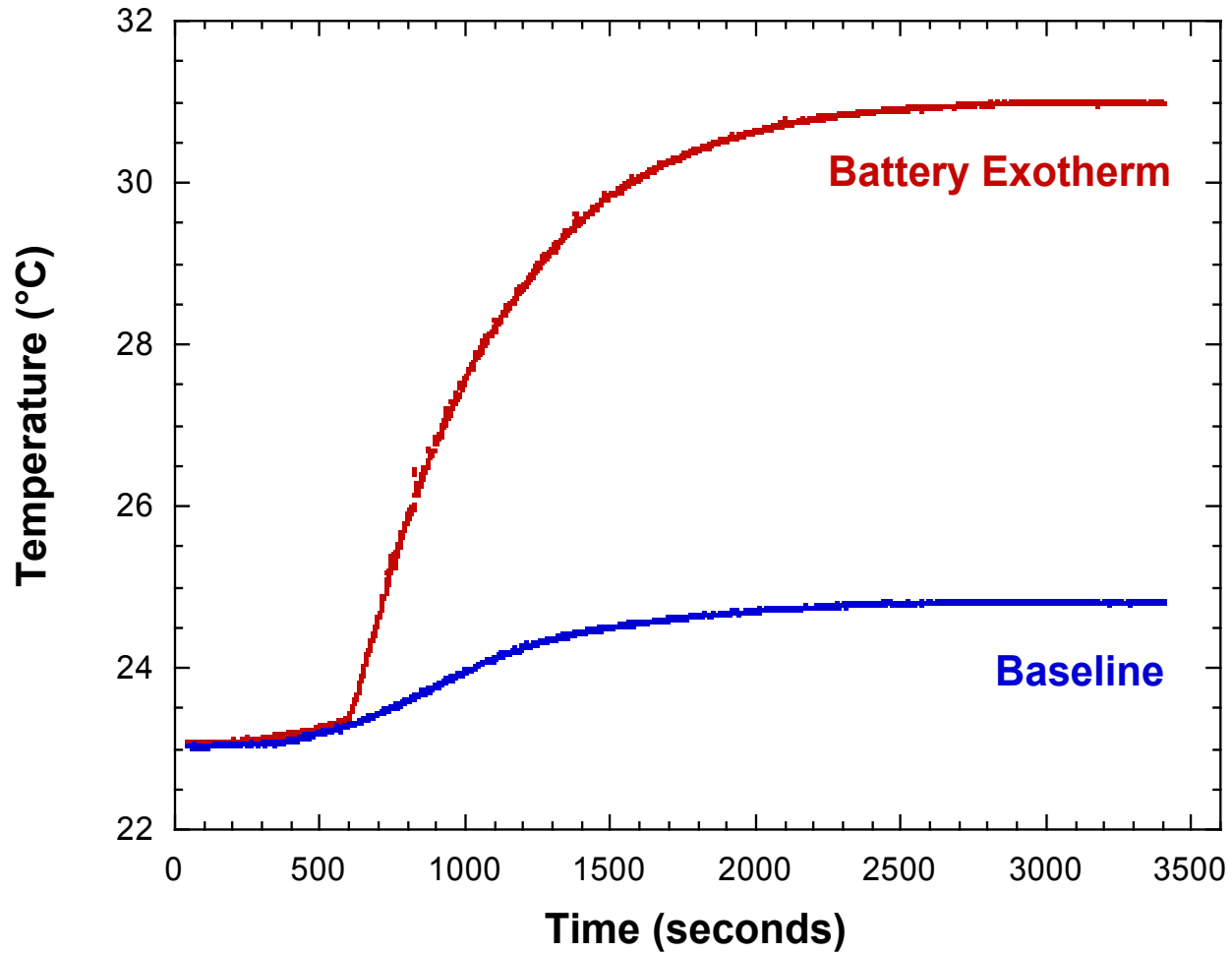


**Bomb and other components for 18650 battery tests**

## Experimental Setup

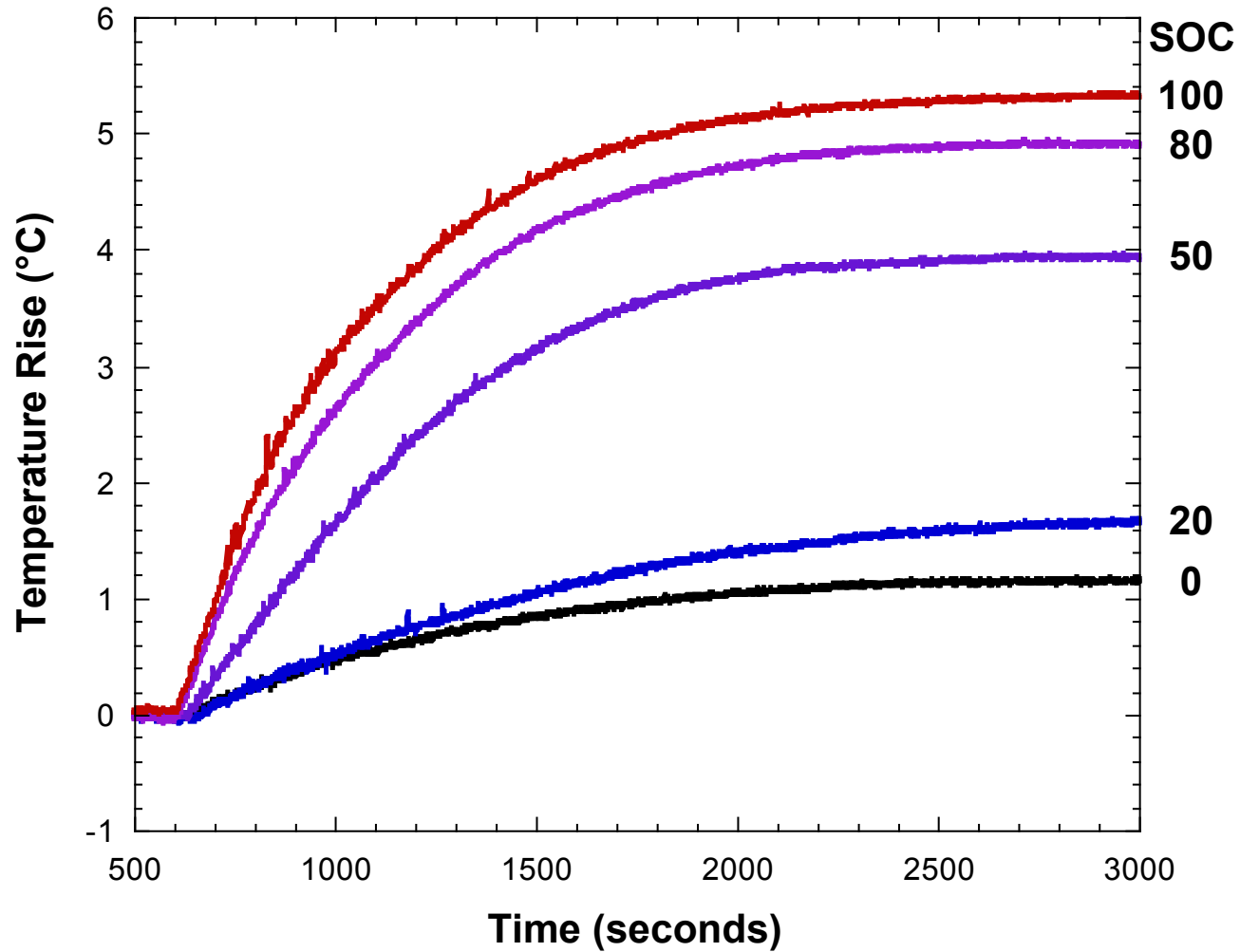


# Test Data from 100% SOC



**Baseline from heater subtracted from battery exotherm to get temperature rise from battery only**

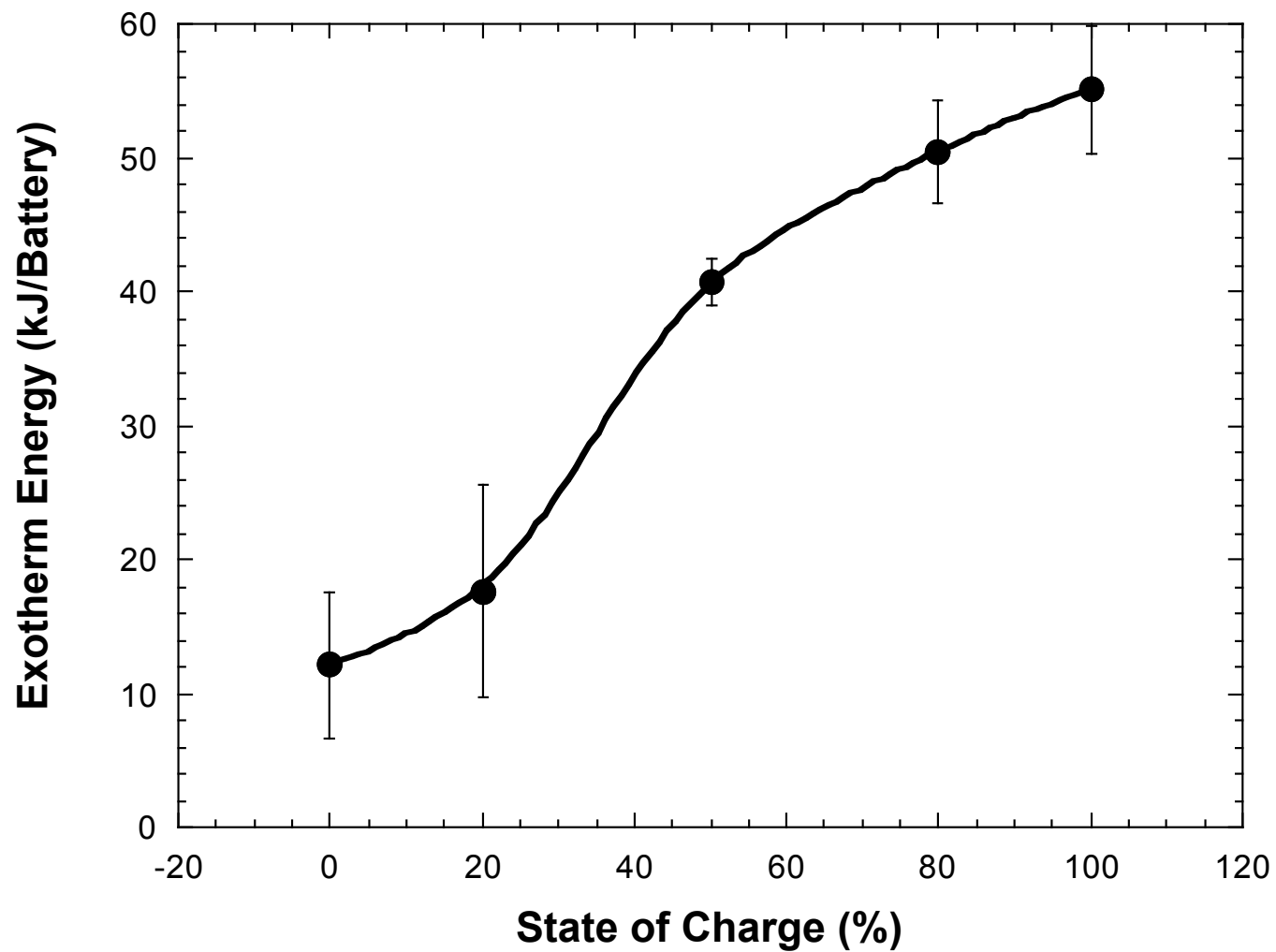
# Temperature Rise In Bomb



Baseline corrected temperature rise data



# Battery Chemical Exotherm Energy



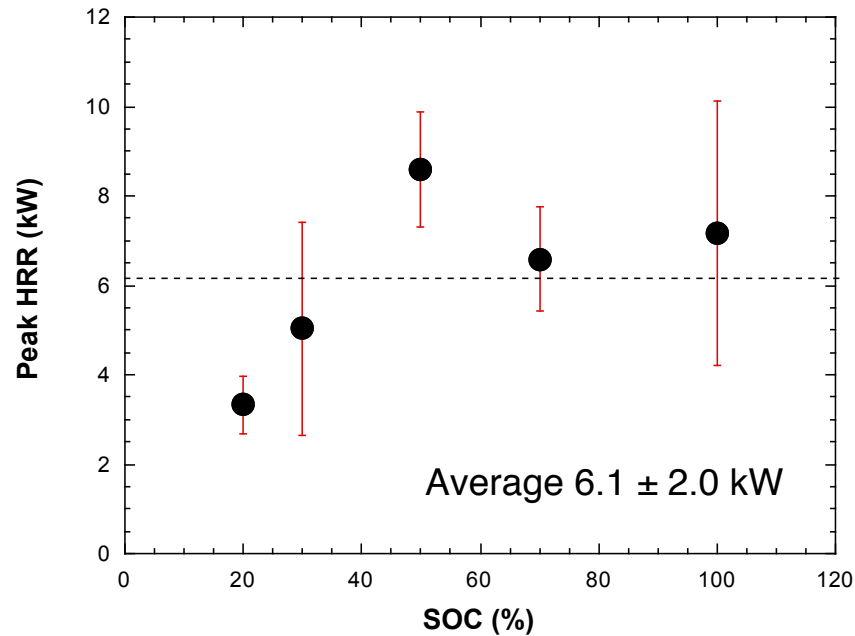
# Cone Calorimeter

- **ASTM E1354**
  - **Peak HRR**
  - **Total HR**
  - **Time to Ignition**
  - **Time to Peak**

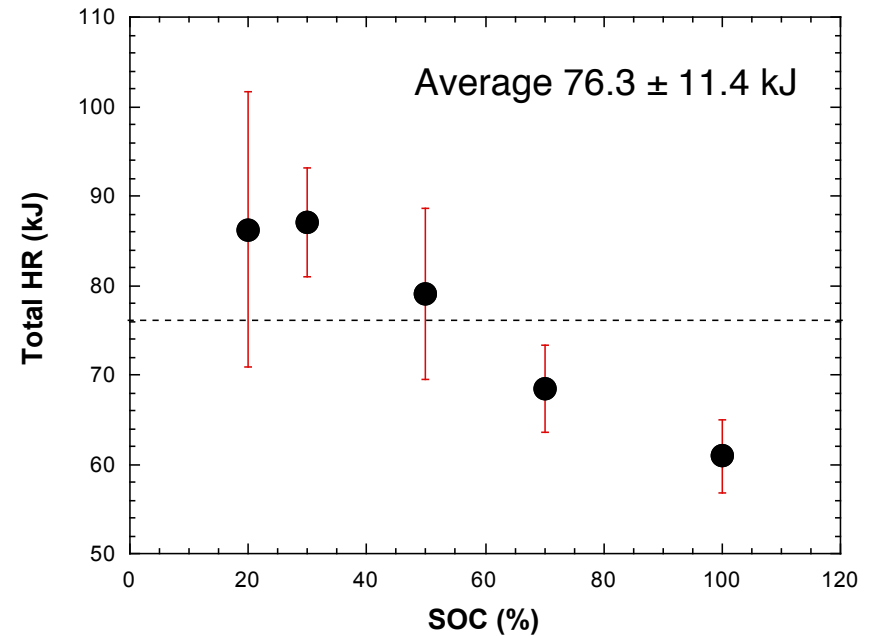


# Cone Calorimeter 50 kW/m<sup>2</sup> - Varying SOC

## Peak HRR

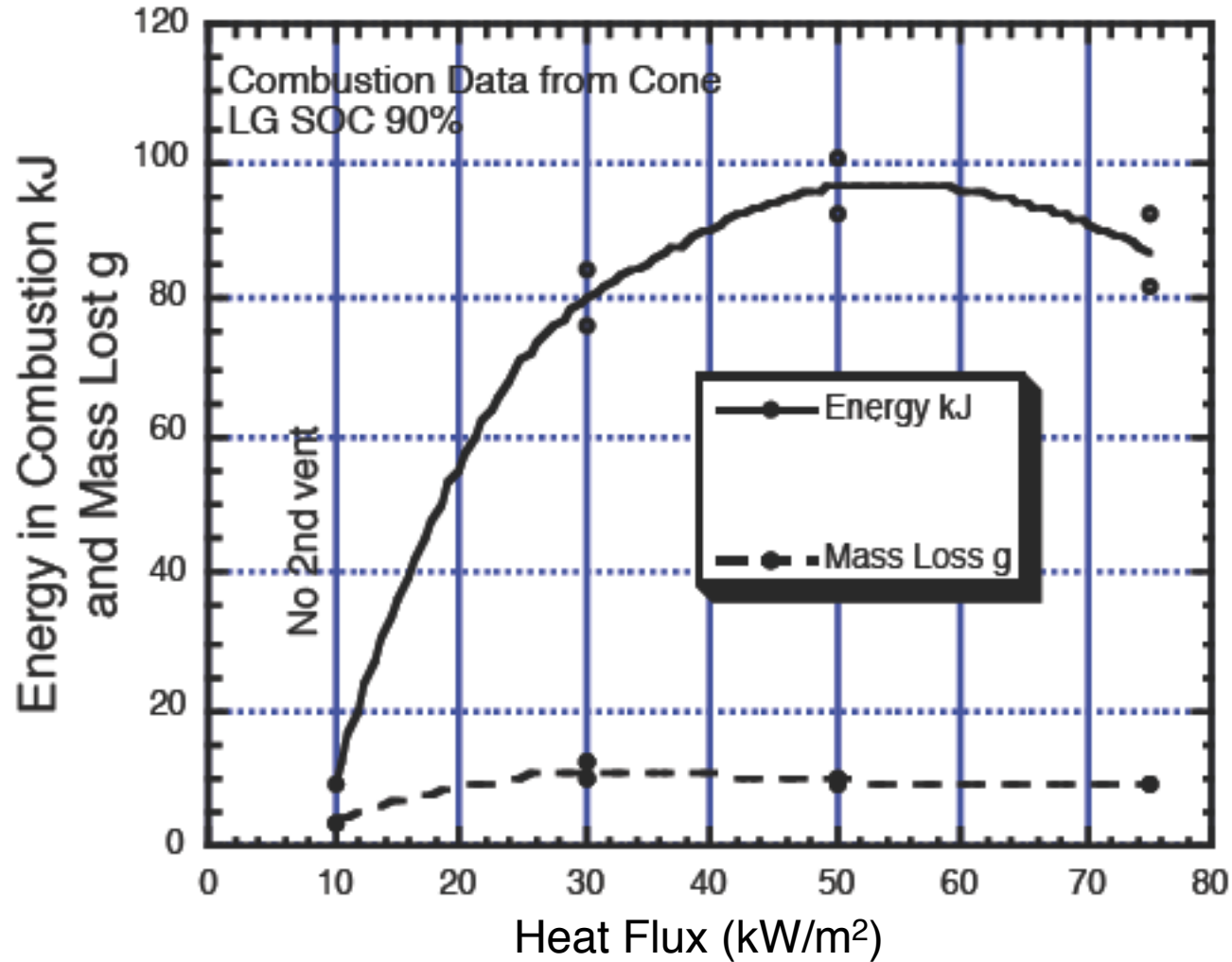


## Total HR



- **Peak and total heat release independent of SOC**
- **Combustion products evolved do not change**
- **Chemical heat release not detected in cone calorimeter**

# Cone Calorimeter 90% SOC - Varying Heat Flux



**Total heat release and mass lost independent of heat flux**

# Findings

- For 18650 Li ion battery with 34.6 kJ electric power and 44 g
- Runaway results most dependent on SOC
  - Tests at 0% and at 0.5 V display limited or no runaway
  - Mass loss increases from about 8 to 19 g with increasing SOC
  - Exothermic runaway energy increases from 10 to 60 kJ with SOC (Bomb)
  - Combustion energy doesn't vary much with SOC and heat flux (Cone data)
  - Temperatures of battery and exit debris increase from about 600 to 1000°C with SOC
  - Duration of runaway ~ 2 s
  - Duration of flaming ~ 10 s
  - Duration of hot battery ~ 100+ s



# **Acknowledgement**

- **Thank you to Steve Summer, FAA, for supplying and charging the batteries**